

BSc DEGREE EXAMINATION MAY 2018

(Fifth Semester)

Branch – **CHEMISTRY**

PHYSICAL CHEMISTRY – I

Time : Three Hours

Maximum : 75 Marks

SECTION-A (20 Marks)

Answer **ALL** questions

ALL questions carry **EQUAL** marks

(10 x 2 = 20)

- 1 What is Conductance ratio?
- 2 State the Faraday's law of Electrolysis.
- 3 What is dissociation constant?
- 4 Mention the factors influencing degree of dissociation.
- 5 Write any two applications of Buffer solution.
- 6 What is p^H scale?
- 7 Define electrolytic cell.
- 8 What are acid-base titrations?
- 9 Give the basic components of electroplating.
- 10 Illustrate the term corrosion.

SECTION - B (25 Marks)

Answer **ALL** Questions

ALL Questions Carry **EQUAL** Marks (5 x 5 = 25)

- 11 a (i) Describe the mechanism of electrolysis.
(ii) Explain ionic mobility.
OR
b State and explain Kohlrausch law.
- 12 a Explain Arrhenius theory of electrolytic dissociation and its limitations.
OR
b Discuss the conductometric titration involving mixture of strong and weak acids with strong base.
- 13 a Write a short notes on: (i) Ionic product of water (ii) Buffer capacity
OR
b Calculate the degree of hydrolysis of 0.01M solution of sodium acetate at 25⁰C, $K_a=1.75 \times 10^{-5}$ and $K_w=1.008 \times 10^{-14}$
- 14 a Write a note on: Quinhydrone electrode
OR
b Distinguish between reversible and irreversible cells.
- 15 a Explain the principles of electroplating.
OR
b Describe the following types of plating (i) alloy (ii) electrolysis

SECTION - C (30 Marks)

Answer any **THREE** Questions

ALL Questions Carry **EQUAL** Marks (3 x 10 = 30)

- 16 What is Transport number? How it is determined by moving boundary method?
- 17 a) Discuss Debye-Huckel Onsager equation.
b) Explain Debye-Falkenhagen effect.
- 18 a) Explain briefly about common ion effect with example.
b) Write a note on: levelling effect.
- 19 Discuss the concentration cell with transference and without transference.
- 20 Discuss in detail the types of coating processes with suitable example.

Z-Z-Z

END