PSG COLLEGE OF ARTS & SCIENCE (AUTONOMOUS)

MSc DEGREE EXAMINATION MAY 2019

(Second Semester)

Branch-CHEMISTRY

INORGANIC CHEMISTRY - II

	INURGAN	IC CHEMISTRY - II	
Time	e: Three Hours	M	laximum: 75 Marks
	Answ	TON-A (10 Marks) ver ALL questions ons carry EQUAL marks	$(10 \times 1 = 10)$
1	Identify the following Organometallic reaction. *Cl Cl1*' • I I M *Cl Cl1*' • I I M *Cl Cl1*' • I I I I I I I I I I I I I I I I I I		
	a''Cl	of'' a	
	(i) Oxidative addition (iii) Elimination Reaction	(ii) Nucleophilic Reactio (iv) Reduction	n
9	What is the other name of dative bond?		
	(i) Ionic bond	(ii) Covalent bond	
	(iii) Hydrogen bond	(iv) Co-ordinate covaler	nt bond
3	Identify the process in which ethylene is converted into acetaldehyde in presence of 'Pd' catalyst.		
	(i) wacker process	(ii) oxo process	
	(iii) Reppe process	(iv) Reduction process	
4	Which catalyst is widely used in the Olefin hydrogenation process?		
	(i) Wilkinson's Catalyst	•	
	(iii) $[Rh(CO)_2I_2]'$	(iv) Co ₂ (CO)g	
5	Identify the correct formula for Bent-Sandwich structure.		
	(i) $[(tl^5-C_sH_5)_2M]$ (ii) $[(t_1^5-C_5H_5)_2ML_{\chi}]$		
	(iii) [(T, ⁵ - c ₅ H ₅) ML _s]	(iv) $\mathbf{K} \cdot 1^5 - \mathbf{c}_6 \mathbf{H}_5 \mathbf{)}_2 \mathbf{M}$	
6	What is the oxidation state of iron in ferrocene?		
	(i) 0	(ii) +1	
	(iii) $+2$	(iv) +3	
7	Identify the type of C ₂ B ₄ H ₈ carboranes.		
	(i) Closo	(ii) nido	
	(iii) arachno	(iv) hypo	
8	Find out the number of doubly bridging CO ligand in Fe ₂ (CO) <>.		
	(i) 1 (ii) 2		
	(iii) 3	(iv) 4	
9	What is the difference in ionic radius between high spin Fe(II) and low spin Fe(II)?		
	(i) 75pm	(ii) 92pm	
	(iii) 85pm	(iv) 17pm	
10	Which shows more intense absorption among the following?		

(ii) Azurin

(iv) $[Cu(NH_3)_6]^{2+}$

Plastocyanin

(i)

(iii) Cyanin

SECTION - B (25 Marks)

Answer ALL questions

ALL questions carry EQUAL Marks ($5 \times 5 = 25$)

a Illustrate the mechanism of insertion of CO to M-C bonds.

OR

b (i) State 18 electron rule.

- **(2)**
- (ii) Analyse the stability of the following metal complexes using 18 electron rule: $V(CO)_6$, $Fe(CO)_5$, $Mn_2(CO)_{10}$ (3)
- 12 a Sketch the mechanism of hydrosilylation reaction.

ΩR

- b Illustrate the mechanism of hydro carbonylation of olefins.
- ,13 a Explain briefly the structure of arene half sandwich complexes.

ΛR

- b Discuss briefly the structure of cyclopentadienyl complexes.
- 14 a Explain briefly the photo substitution reaction with suitable example.

OR

- b Discuss briefly the preparation and structure of 1,2-dicarbaclosododecaborane.
- 15 a Discuss the role of Zn(II) in carboxypeptidase.

OR

b Determine the host-guest complexes of Clathrand.

SECTION -C (40 Marks!

Answer ALL questions

ALL questions carry EQUAL Marks ($5 \times 8 = 40$)

16 a Classify the metal-carbon bond types and explain them.

OR

- b Construct a synthesis of carbene complexes and explain their properties.
- 17 a Construct the catalytic loop for olefin hydrogenation.

OR

- b Enumerate the Zeigler-Natta polymerisation of olefins.
- 18a Design the synthesis of allyl complexes and discuss their structure.

OR

- b Design the synthesis of arene complexes and explain their structure.
- 19 a Criticize the application of IR spectroscopy in the study of metal carbonyls.

OF

- b Compare photoreduction and photooxidation reactions with suitable examples.
- 20 a Evaluate the role of Vaska's iridium complexes.

OR

b Justify the Perutz cooperative mechanism of hemoglobin.

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