18CHP15

PSG COLLEGE OF ARTS & SCIENCE (AUTONOMOUS)

MSc DEGREE EXAMINATION DECEMBER 2022

(Third Semester)

Branch - CHEMISTRY

PHYSICAL CHEMISTRY-III

Time: Three Hours	1.5		35 1 5035 4
Time. Timee Hours			Maximum: 50 Marks
* .		 2.1	

SECTION-A (5 Marks)

Answer **ALL** questions

ALL questions carry EQUAL marks $(5 \times 1 = 5)$

- 1 Which of the following is incorrect about transition state theory?
 - (i) It is based on statistical mechanics
 - (ii) The formation of activated complex is rapid
 - (iii) The decomposition of activating complex is slow
 - (iv) The formation of activated complex is the rate determining step
- 2 Flash photolysis is a
 - (i) Pump-probe technique
- (ii) Jump-probe technique
- (iii) Temperature-probe technique
- (iv) Pressure-technique
- The surface coverage (θ) , related to pressure and equilibrium constant by Langmuir adsorption isotherm is
 - (i) $\theta = bP/(1+bP)$
- (ii) $\theta = bP / (1+bP^2)$
- (iii) $\theta = (1+bP)/bP$
- (iv) $\theta = bP$
- 4 Maxwell-Boltzmann statistics cannot be applied to
 - (i) Atoms

(ii) Molecules

(iii) Photons

(iv) Lattice

5 Phase-Space is a

b

- (i) 3-dmensional space
- (ii) 1-dimensional space
- (iii) 2-dimensional space
- (iv) 6-dimensional space

SECTION - B (15 Marks)

Answer ALL Questions

ALL Questions Carry EQUAL Marks

 $(5 \times 3 = 15)$

- 6 a Show that Lindemann's hypothesis helps to understand the change in the order of reaction due to changes in pressure with some gaseous reactions.

 OR
 - Distinguish primary and secondary kinetic isotopic effects with examples.
- 7 a How does Michaelis-Menten equation, explains the observed kinetics of enzyme catalyzed reactions?

OR

- b Compare and contrast stopped flow and continuous flow techniques for the study of kinetics of fast reactions.
- 8 a. Deduce a BET adsorption isotherm equation. Mention its applications.

OR

b. What are hydrated electrons? Explain their characteristics and importance.

Cont...

Calculate the translational partition function of NO molecule at 300 K in a volume 1000 m³. Assuming that NO behaves ideally.

- b. Derive Seckur-Tetrode equation.
- Compare the salient features of Bose-Einstein and Fermi-Dirac statistics. 10 a.

OR

What is meant by negative absolute temperature? How it is achieved? b.

SECTION -C (30 Marks)

Answer ALL questions ALL questions carry EQUAL Marks

 $(5 \times 6 = 30)$

11 a. Derive an expression for the rate constant of a bimolecular reaction based on absolute reaction rate theory. In what way ARRT is superior to other theories.

- b Deduce an expression for the kinetics of H_2 Br_2 thermal reaction.
- 12 a. Derive the rate law of Arrhenius and Van't Hoff intermediate complexes.
 - b. Apply the double sphere model and deduce a relationship between the solvent dielectric constant and rate constant.
- 13 a. Sketch a schematic diagram of Fricke dosimeter. Explain its mode of functioning and applications.

OR

- b Describe the Langmuir-Hinshelwood theory for a bimolecular surface reactions. Mention its merits and limitations.
- 14 a Prove that complete partition function for a system is the product of translational, rotational vibrational and electronic partition functions.

OR

b Distinguish the following pairs:

- (i) Classical statistical mechanics and Quantum statistical mechanics **(2)**
- (ii) Distinguishable particles and indistinguishable particles

(2) (iii) Canonical ensemble and grand canonical ensemble **(2)**

Derive Einstein's equation of heat capacity of solids in terms of partition 15 a function.

b. Propose a distribution law applicable to bosons. Derive it.

Z-Z-Z

END