#### PSG COLLEGE OF ARTS & SCIENCE (AUTONOMOUS)

#### **BSc DEGREE EXAMINATION MAY 2024**

#### Branch - MICROBIOLOGY

#### **FUNDAMENTALS OF BIOCHEMISTRY**

Maximum: 75 Marks Time: Three Hours

#### SECTION-A (10 Marks)

Answer ALL questions

**ALL** questions carry **EQUAL** marks  $(10 \times 1 = 10)$ 

ALL questions carry EQUAL marks $(10 \times 1 = 10)$								
Module No.	Question No.	Question	K Level	СО				
1	1	How many H bonds can a water molecule form at a time? a)1 b) 2 c) 3 d) 4	K1	CO1				
	2	Buffers are mixtures of a) Strong acid and strong base b) Strong acid and weak base c) Weak acid and its conjugate base d) Weak base and its conjugate acid	K2	CO1				
2	3	Which one of the following is a non-reducing sugar?  a) Glucose b) Fructose c) Sucrose d) Maltose	K1	CO2				
	4	Where is chondroitin sulfate found in the body?  a) In nervous system b) In bloodstream c) In the liver d) In cartilage and connective tissues	K2	CO2				
3	5	Phospholipids are also called as a) Simple lipids b) Derived lipids c) Storage lipids d) Membrane lipids	K1	соз				
	6	The vitamin synthesized by only microorganisms is a) Vitamin B 12 b) Vitamin B1 c) Vitamin B2 d) Vitamin A	K2	CO3				
4	7	Among the standard 20 amino acids which one is least occur in proteins?  a) Glycine b) Alanine c) Tryptophan d) Methionine	K1	CO4				
	8	The back bone of nucleic acid structure is constructed by  a) Peptide bonds b) Glycosidic bonds c) Phosphodiester bonds d) All of them	K2	CO4				
5	9	In the feedback regulation the end product binds at a) Active site b) Allosteric site c) E-S complex d) None of these	K1	CO5				
	10	The serum enzyme elevated in viral hepatitis is a) Alanine transaminase b) Aspartate transaminase c) Alcohol dehydrogenase d) Gamma glutamyl transpeptidase	K2	CO5				

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## 22MBU208N/ 22MBU208/ 18MBU07

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# SECTION - B (35 Marks) Answer ALL questions

ALL questions carry EOUAL Marks  $(5 \times 7 = 35)$ 

Module No.	Question No.	Question	K Level	СО
1	11.a.	Discuss on elements present in macro molecules.		
	Mary March	(OR)	K2	CO1
	11.b.	Enumerate the unique properties of water.		
	12.a.	What are anomers? Explain mutarotation with example.		
2		(OR) K2		CO2
	12.b.	What are glycosaminoglycans? Show the N- glycosidic and O- glycosidic bond formation.	11.2	002
3	13.a.	Classify fatty acids. Explain their structure with suitable examples.	K2	
	(OR)			CO3
	13.b.	Discuss the role of vitamin D in bone health.		
4	14.a.	Compare fibrous proteins and globular proteins.		
	(OR)			CO4
	14.b.	Classify amino acids based on their R groups.		
5	15.a.	Explain feedback inhibition of enzyme activity with example.		CO5
		(OR)	K2	
	15.b.	Discuss about the clinical significance of ALP and ACP.		

### SECTION -C (30 Marks)

Answer ANY THREE questions

ALL questions carry EOUAL Marks  $(3 \times 10 = 30)$ 

Module No.	Question No.	Question	K Level	со
1	16	Describe the role of blood buffers in acid- base balance.	K2	CO1
2	17	Describe the classification of carbohydrates based on their reducing property.	K2	CO2
3	18	What are steroids? Explain the structure and functions of cholesterol.	K2	CO3
4	19	Sketch the structure and describe the functions of three types of RNA.	K2	CO4
5	20	Derive Michaelis-Menton equation for single substrate enzyme catalyzed reaction.	K2	CO5