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PSG COLLEGE OF ARTS & SCIENCE

(AUTONOMOUS)

BSc DEGREE EXAMINATION MAY 2024

(Third Semester)

Branch - CHEMISTRY

GENERAL CHEMISTRY - III

	Time	e: Three Hours		Maximum: 50 Marks
SECTION-A (5 Marks) Answer ALL questions ALL questions carry EQUAL marks (5 x 1 = 5)				
1	(i)		between 2 electrons, 1 atoms 2 electrons, 2 atom	
2	(i)		owing. Malonic acid Cinnamic acid	
3	What is the reagent which will react with cumene to give phenol? (i) Nitrogen (ii) Hydrogen (iii) Oxygen (iv Ozone			
4	(i)		rsible process $\Delta S_{Total} < 0$ $\Delta S_{Total} = 0$	
5	be (i)	Equilibrium mixture (ii)	rates are equal in th Dynamic equilibrium Static equilibrium	
SECTION - B (15 Marks) Answer ALL Questions ALL Questions Carry EQUAL Marks (5 x 3 = 15)				
6	a Discuss about the electron acceptor behavior of boron hydrides. OR			
	b	Explain the preparation, properties	and structure of Per	disulphuric acids.
7	a			
	b	OR List out the preparation of Acetop	henone.	
8	a	Give the details preparation, proper		oroform.
		OR		
	b	How is 1,2 dichloro ethane prepare		
9	a	Illustrate the limitation of first law OR	of thermodynamics.	
	b	Derive Gibbs Helmholtz equation.		
10	a	Derive Gibbs Duhem equation. OR	-	
	b	State the third law of thermodynam	nics.	

SECTION -C (30 Marks)

Answer ALL questions
ALL questions carry EQUAL Marks

 $(5 \times 6 = 30)$

11 a Discuss indetail about preparation, properties and structure of borazole.

OR

- b Write the preparation and structure of AB₅ and AB₇ of Interhalogen compounds. (3+3)
- 12 a Describe the preparation and properties of Cinnamic acid.

OR

- b Explain the preparation, properties of oxalic acid and Succinic acid.
- 13 a Differentiate between primary, secondary and tertiary alcohols by lucas and victor meyer methods. (3+3)

OR

- b List out the preparation and properties of resorcinol.
- 14 a Elucidate the Carnot's theorem.

OR

- b Evaluate an expression for the entropy of a mixture of idealgas.
- 15 a Derive the Clapeyron clausius equation with its application.

OR

b Applying the Le-chatelier's principle and physical equlibria.

END