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PSG COLLEGE OF ARTS & SCIENCE (AUTONOMOUS)

BSc DEGREE EXAMINATION JUNE 2014

(Third Semester)

Branch - PHYSICS

CHEMISTRY-I

Time: Three Hours

Maximum: 75 Marks

SECTION-A (20 Marks)

Answer ALL questions

ALL questions carry EQUAL marks

 $(10 \times 2 = 20)$

- 1 Define Coordination compounds.
- What is Chelates?
- 3 State isoprene rule.
- 4 Write any one preparation of Teflon.
- 5 Define: Symmetry.
- 6 What is axis of symmetry?
- 7 Define rate constant.
- 8 Give any 2 examples for 2nd order reactions.
- 9 What is conventional source of energy?
- 10 Differentiate nuclear fission and fusion.

SECTION - B (25 Marks)

Answer ALL Questions

ALL Questions Carry EQUAL Marks $(5 \times 5 = 25)$

11 a Give an account of Werner's coordination theory with suitable example.

OR

- b Write the preparation and properties of ICI, IF₅?
- 12 a Discuss the isolation of nicotine and give its uses.

OR

- b Write a note on acid and basic dyes with example.
- 13 a Describe the symmetry elements.

OR

- b Define the miller indices and Weiss indices and mention its uses.
- 14 a What are complex reactions? Explain with suitable examples.

OR

- b Give an account of consecutive, parallel and opposite reactions? With examples.
- 15 a Write a note on the various components of environment.

OR

b Define photovoltaic effect, and about Si solar cell.

SECTION - C (30 Marks)

Answer any THREE Questions

ALL Questions Carry EQUAL Marks (3 x 10 = 30)

- 16 a) Give the preparation and properties of perdi-sulpuric acid. Mention its uses.
 - b) Write the analytical application of coordination compounds.
- 17 a) Explain the isolation and uses of menthol.
 - b) Describe the preparation and uses of polyesters.
- 18 a) Illustrate the structure of NaCl and diamond.
 - b) Differentiate isomorphism and polymorphism.
- 19 Derive the 1st order kinetics for
 - i) Decomposition of H2O2 ii)
- ii) Hydrolysis of Ethyl acetate
- 20 Describe briefly about the radioactive pollution effects and control measures.

Z-Z-Z

END