PSG COLLEGE OF ARTS & SCIENCE (AUTONOMOUS)

BSc DEGREE EXAMINATION MAY 2017

(Third Semester)

Branch-CHEMISTRY

GENERAL CHEMISTRY-IH

Time : Three Hours

Maximum : 75 Marks

SECTION-A (20 Marks)

Answer ALL questions ALL questions carry EQUAL marks (10x2 = 20)

1 Give any two uses of Diborare.

 $2\,$ -. Write the structure of SO $_5$ $^{2\prime}$ and $S_2\,O_g\,^{2\prime}$ ions.

3 Complete the reaction
$$\swarrow$$
 CH₃COcl ?

- 4 Draw the Cis- and trans isomers of crotonaldehyde.
- 5 Write any two preparations of Dihydric alcohols.
- 6 Give the IUPAC name of the following halide compound. X

- 7 State second law of thermodynamics.
- 8 Define Entropy and efficiency.
- 9 Write the Gibbs-Deuhem equation.
- Write the expression for Kp and Kc. 10

SECTION - B (25 Marks)

Answer ALL Questions

ALL Questions Carry EQUAL Marks (5x5 = 25)

11 a Explain the structure of silicones arid mention any two uses.

b Describe the structure and preparation of IF₅.

12 a Discuss the preparation and properties of Acroleis.

OR

b How does chinnamic acid reacts with the following reagents.

i) Na / Hg, H₂0 ii) Chromic acid ' iii) NaOH, CaO.

13 a Explain the preparation and any three properties of CCI4.

OR

OR

b Describe the preparation and properties of Dihydric pherol.

a Explain the Entropy change of an ideal gas with charges in P, V and T. 14

b Derive the Gibbs - Helmholtz equation.

15 a Write a note on Nernst heat theorem. OR b Derive the Claoevron-clausius eauation.

1Ac.HCJC>2

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<u>SECTION - C (30 Marksl</u> 'Answer any THREE Questions ALL Questions Carry EQUAL Marks (3x10 = 30)

16 Discuss the preparation and properties of i) Barazole and ii) Dithionic acid.

17 a Explain the properties of Berzaldehyde in detail. *

b Discuss the preparation and any three properties of Alorylic acid,

18 a Explain the preparation, any two properties and uses of CHC1₃.

b How will you distinguish primary, secondary and tertiary alcohols?

19 Explain the Carnot cycle in detail.

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20 a Briefly discuss the Le-Chatelier's Principle.

b Derive the Van't Hoff equation.

Z-Z-Z END