

PSG COLLEGE OF ARTS & SCIENCE
" (AUTONOMOUS)

BSc DEGREE EXAMINATION MAY 2017
(First Semester)

Branch - **CHEMISTRY**

GENERAL CHEMISTRY -1

Time : Three Hours

Maximum : 75 Marks

SECTION-A (20 Marks)

Answer **ALL** questions

ALL questions carry **EQUAL** marks (10 x 2 = 20)

- 1 State Heisenberg's uncertainty principle.
- 2 Write short notes on principal quantum number.
- 3 Define the term ionization energy.
- 4 State electron affinity,
- 5 Explain the hydrogen bonding involved in o-nitro phenol and m-nitro phenol.
- 6 How covalent bond is formed?
- 7 Define bond length.
- 8 Define bond energy.
- 9 Explain the relative stability of primary, secondary and tertiary carbonium ions.
10. Write a note on homolytic cleavage of covalent bond. '

SECTION - B (25 Marks)

Answer **ALL** Questions

ALL Questions Carry **EQUAL** Marks (5 x 5 = 25)

- 11 a State the postulates of Bohr about an atom.
OR
b Write the rules for the filling of orbitals in the ground state of an atom.
- 12 a Define electro negativity and discuss Mulliken electro negativity scale.
OR
b Explain atomic and ionic radii.
- 13 a Discuss Fajan's rule.
OR
b Discuss Born-Haber cycle calculations.
- 14 a Compare MO and VB theories.
OR
b Discuss the MO diagram of N₂.
- 15 a Discuss the hybridization and geometry of methane.
OR
b Discuss the hybridization and geometry of ethylene.

SECTION - C (30 Marks)

Answer any **THREE** Questions

ALL Questions Carry **EQUAL** Marks (3 x 10 = 30)

- 16 a Explain De Broglie's theory and derive De Broglie equation. (5)
- b Explain azimuthal and magnetic quantum number. (5)
- 17 Discuss the modern periodic table.
- 18 Discuss the characteristics of electrovalent compounds.
- 19 a Write the postulates of VB theory,
b Discuss MO diagram of CO.
- 20 Explain addition, substitution and elimination reactions.