

MSc DEGREE EXAMINATION DECEMBER 2018  
(First Semester)

Branch - CHEMISTRY

**ORGANIC CHEMISTRY -1**

Time : Three Hours

Maximum : 75 Marks

**SECTION-A (10 Marks)**

Answer ALL questions

ALL questions carry EQUAL marks (10 x 1 = 10)

Identify the more stable carbocation by using hyperconjugation concept.

- (i)  $\text{CH}_3\text{CH}_2\text{CH}_2^+$  (ii)  $\text{CH}_3\text{CH}=\text{CH}-\text{CH}_2^+$  (iii)  $\text{CH}_3\text{C}^+\text{H}(\text{CH}_3)_2$  (iv)  $\text{CH}_3\text{C}^+\text{H}(\text{CH}_3)$

Which hypothesis describes the geometrical structure of the transition state in an organic chemical reaction?

- (i) Microscopic reversibility (ii) Taft equation  
(iii) Hammett equation (iv) Hammond postulate

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Find the R<sub>s</sub> Notation of the following organic compound,

- (i) R (ii) S  
(iii) Optically inactive (iv) Z

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What are stereospecific reactions?

- (i) A reaction where one stereoisomer of a product is preferentially formed  
(ii) A reaction that produces two enantiomers  
(iii) A reaction in which starting material determines the stereochemistry of the product  
(iv) A reaction in which stereogenic centre is introduced

Which of the following is a best leaving group?

- (i) F<sup>-</sup> (ii) Cl<sup>-</sup>  
(iii) Br<sup>-</sup> (iv) I<sup>-</sup>

Identify the product formed in the following reaction :

- (i) 3 - aminopyridine (ii) 2 - aminopyridine  
(iii) 3, 5 - diaminopyridine (iv) 2, 6 - diaminopyridine



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What is the product formed when retinol undergoes ozonolysis?

- (i) acetic acid (ii) formic acid  
(iii) geronic acid (iv) succinic acid

8 Which of the following saccharide has cross-linked structure?

- (i) Starch (ii) Cellulose (iii) Maltose (iv) Sucrose

9 What is / are the product(s) formed when cyanin chloride undergoes hydrolysis in the presence of KOH?

- (i) Phloro glucinol (ii) Protocatechuic acid  
(iii) both (i) and (ii) (iv) Phthalic acid

10 Identify the product formed in the following reaction :

- (i) flavones (ii) pyrazoles  
(iii) luteoline (iv) anthocyanins

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**SECTION - B (25 Marks)**

Answer ALL questions

ALL questions carry EQUAL Marks (5 x 5 = 25)

11 a Explain the factors that affect the strength of acids and bases.

OR

12 a Illustrate the optical activity in biphenyl compounds.

OR

b Discuss the various chemical methods that are used to determine the configuration of a geometrical isomers.

13 a Illustrate the effect of substrate in aliphatic nucleophilic substitution reactions.

OR

b Sketch the mechanism of the following :

(i) Acid hydrolysis of ester (ii) Benzyne mechanism

(2 V<sub>2</sub> + 2 'A)

14 a Justify the structure of reserpine.

OR

b Explain the structural elucidation of sucrose.

15 a Discuss the structure and synthesis of flavones.

OR

b Discuss the synthesis and reactivity of oxazole.

### **SECTION -C (40 Marks J**

Answer ALL questions

ALL questions carry EQUAL Marks (5 x 8 = 40)

16 a Construct the Hammett equation and explain the application of *a* and *p*.

OR

b Justify the uses of primary and secondary kinetic isotopic effects in determining the mechanism of organic reaction.

17 a Analyse the conformation of disubstituted cyclohexane and predict the stable isomers.

OR

b i) Compare the stereospecific and stereoselective synthesis.

ii) Construct any two chiral compounds using asymmetric synthesis.

18 a Justify the increase in the rate of organic reactions by using neighbouring group participation concept.

OR

b i) Design the mechanism of Zeigler alkylation reaction.

ii) Differentiate SN1 and SN2 aliphatic nucleophilic substitution reactions. (4+4)

19 a Elucidate the structure of Zingiberene.

OR

b Elucidate the structure of retinol.

20 a Justify the structure of luteoline.

OR

b i) Construct any two suitable synthesis for isoflavones. (4)

ii) Elucidate the structure of cyanin chloride. (4)

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**END**