PSG COLLEGE OF ARTS & SCIENCE (AUTONOMOUS)

MSc DEGREE EXAMINATION DECEMBER 2018

(Third Semester)

Branch - CHEMISTRY

MOLECULAR SPECTROSCOPY & APPLICATIONS					
Ti	me	: Three Hours Maximum : 75 Marks			
Answer ALL questions ALL questions carry EQUAL marks $(5 \times 15 = 75)$					
1	a	Explain the term 'Transition Probability'.	(3)		
	b	Discuss the different types of electronic transition.	(5)		
	С	Explain the solvent effects in $\pi \to \pi^*$ and $n \to \pi^*$ transitions. OR	(7)		
	d	Calculate the λ max for the following molecules:	(6)		
		i) iii) $0 = C - CH_3$			
	e	Write a short notes on absorption and intensity shifts.	(5)		
	f	Discuss any four applications of UV - visible spectra to organic compounds.	(4)		
2	a	Explain the various factors influencing vibrational frequency.	(5)		
	b	Define Fermi Resonance.	(3)		
	С	Discuss the number of fundamental vibrations of H_2O and CO_2 Molecules. OR	(7)		
	d	Explain stoke and antistoke lines.	(3)		
	e	Bring out the differences between IR and Raman spectroscopy.	(5)		
3	a	Define chemical shift.	(3)		
	b	What are the factors influencing chemical shift?	(5)		
	С	Give a detailed account of spin-spin interaction with an example. OR	(7)		
	d	Explain Nuclear Overhauser effect.	(3)		
	e	Write short notes on chemical shift reagents.	(5)		
	f	Discuss the geminal and vicinal coupling with an example.	(7)		
4	a	Explain the salient features of 'off-resonance decoupling'.	(5)		
	b	How will you determine the structure of complexes of WF_6 and structure of BOF_4 ?	(5)		

c Write short notes on '2D NMR'

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	e	Explain the basic principle of ¹³ C NMR spectra.	(5)
	f	Discuss the applications of NMR spectra in the structural determination of AsF ₃ and SO ₃ .	n (5)
5	a	Discuss the basic principle of ESR spectra.	(8)
	b	Explain zero field splitting and Kramer's degeneracy with an example. OR	(7)
	c	What is isomer shift?	(3)
	d	Explain the quadrupole splitting with an example.	(5)
	е	Discuss the applications of ESR spectra in structural determination of complexes [Fe(CN) ₅ NO] ³⁻ and CuSiF ₆ . 6H ₂ O.	f (7)
		Z-Z-Z EN	D