

**PSG COLLEGE OF ARTS & SCIENCE**  
(AUTONOMOUS)  
**BSc DEGREE EXAMINATION DECEMBER 2018**  
(Second Semester)

Branch - **CHEMISTRY**

**GENERAL CHEMISTRY - II**

Time : Three Hours

Maximum : 75 Marks

**SECTION-A (20 Marks)**

Answer **ALL** questions

**ALL** questions carry **EQUAL** marks (10 x 2 = 20)

- 1 Give the anomalous behaviour of Be.
- 2 Why are noble gases called inert?
- 3 Define intensive property with an example.
- 4 What are the path functions? Give an example.
- 5 Define Unit cell.
- 6 Sketch CsCl crystal lattice.
- 7 What is anti-Markownikoff's rule?
- 8 Why are alkynes acidic? Give an example.
- 9 Draw the resonance structures of benzene.
- 10 Write the Friedel-Crafts reaction.

**SECTION - B (25 Marks)**

Answer **ALL** Questions

**ALL** Questions Carry **EQUAL** Marks (5 x 5 = 25)

- 11 a Explain the position of noble gases in periodic table.  
OR  
b Give the diagonal relation between Li and Mg.
- 12 a State and explain Hess's law of heat summation.  
OR  
b Derive the relation between  $C_p$  and  $C_v$ .
- 13 a Distinguish between amorphous and crystalline solids.  
OR  
b Derive the Bragg's equation.
- 14 a What is the mechanism of ozonolysis reaction? Give an example and its uses.  
OR  
b State and explain Diels-Alder reactions.
- 15 a What are activating and deactivating substituents? Give examples.  
OR  
b Define and explain Huckel's rule.

**SECTION - C (30 Marks)**

Answer any **THREE** Questions

**ALL** Questions Carry **EQUAL** Marks (3 x 10 = 30)

- 16 Describe the ores and extraction of Beryllium.
- 17 Explain the following
  - a) Kirchoff's equation (5)
  - b) Enthalpy of neutralization (5)
- 18 Sketch and explain the various symmetry elements in crystals.
- 19 State and explain
  - a) Saytzeff and Hofmann rule (5)
  - b) Nucleophilic addition reactions (5)