

(AUTONOMOUS)
BSc DEGREE EXAMINATION MAY 2018
(First Semester)

Branch – **CHEMISTRY**

ANALYTICAL CHEMISTRY

Time : Three Hours

Maximum : 75 Marks

SECTION-A (20 Marks)

Answer **ALL** questions

ALL questions carry **EQUAL** marks (10 x 2 = 20)

- 1 Write the threshold limit value of acetone and carbondioxide.
- 2 Give the handling procedure for carcinogenic chemicals.
- 3 Write the spot test for Ni²⁺ ion.
- 4 Give an example for complexation reaction.
- 5 Define: Molarity.
- 6 What is an indicator? Give an example.
- 7 What are sequestering agents? Give an example.
- 8 What are chelating precipitants? Give two examples.
- 9 What do you mean by vaccum distillation?
- 10 How is boiling point used as a test for purity?

SECTION - B (25 Marks)

Answer **ALL** Questions

ALL Questions Carry **EQUAL** Marks (5 x 5 = 25)

- 11 a Write short notes on the temperature and volumetric glassware.
OR
- b Describe the waste disposal in the laboratory.
- 12 a Write a note on washing of precipitates.
OR
- b Give the procedures for the removal of any two interfacing radicals from a inorganic mixture.
- 13 a What are the characteristics of Primary standard solution?
OR
- b Discuss the principle and theory involved in redox titration.
- 14 a Explain post-precipitation and co-precipitation with suitable examples.
OR
- b What are the advantages and disadvantages of using organic precipitants?
- 15 a Explain the azeotropic distillation method with suitable example.
OR
- b What do you mean by desiccant? Discuss the types of desiccants.

SECTION - C (30 Marks)

Answer any **THREE** Questions

ALL Questions Carry **EQUAL** Marks (3 x 10 = 30)

- 16 a) Describe the safe limits of vapour concentration in the laboratory.
- b) Explain the advantage of using standard joint apparatus.
- 17 Write a note on the various types of reactions involved in qualitative analysis.
- 18 Explain the theory of acid-base indicators.
- 19 Discuss the principle, theory and estimation of nicked by gravimetric method.
- 20 Describe the theory of distillation.

Z-Z-Z

END