PSG COLLEGE OF ARTS & SCIENCE

(AUTONOMOUS)

MSc DEGREE EXAMINATION MAY 2018

(First Semester)

Branch - CHEMISTRY

ORGANIC CHEMISTRY-I

Time: Three Hours Maximum: 75 Marks Answer ALL questions ALL questions carry EQUAL marks $(5 \times 15 = 75)$ Describe the difference between primary and secondary isotope effect. (6) Illustrate Baker – Nathan effect with suitable example. b **(4)** Write a note on Hammond postulate. (5)Explain the following with suitable reasons: (i) p – methylbenzoic acid is weaker than p-nitrobenzoic acid (ii) N- elkylated amines are stronger bases than aniline (2+2)Explain the following: (i) Hammett equation (ii) Taft equation. (4+4)Write a note on hydrogen bonding. (3) 2 Write a note on optical isomerism of spiranes. (3) Give the absolute configuration of the hook of the ho Give the absolute configuration of each of the following compounds: Briefly explain the following terms: (i) Sawhorse projections (ii) Newman projections (iii) Erythro and threo nomenclature (iv) Diastereoisomerism Discuss the stereochemistry of sulphur compounds. (4) Explain the possible conformations for dimethylcyclohexanes. (5) Discuss the stereospecific and stereoselective reactions with examples. (6) f Explain the mechanism of SNⁱ reaction. (3) 3 Propose the mechanism of the given reaction and predict the products. (4) Hic-c-CH3 \xrightarrow{AcOH} A+8

Hic-c-CH3 \xrightarrow{AcOH} A+8 \xrightarrow{AcOH} A+B c) Explain: (i) Chichibabin reaction (ii) Zeigler alkylation. (4+4)Discuss the ambident substrates and ambident nucleophile. **(4)** d Describe the stereochemistry of SN¹ and SN² reaction. (8) (3) Explain the reactivity of benzyne intermediate.

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4	a	What is the structural difference between starch and cellulose?	(2)
	b	Outline the structure of cadenine.	(5)
	С	Elucidate the structure and synthesis of tylophorine. OR	(8)
	d	Explain why sources is not a reducing sugar.	(2)
	e	Discuss the structure, configuration and conformation of maltose.	(8)
	f.	Discuss the chemistry of thiamine.	(5)
5	a	How will you distinguish between flavones and isoflavones?	(2)
	b	Discuss the synthesis, reactivity and applications of pyrazoles.	(8)
	c	Explain the chemistry of kaemferol. OR	(5)
	d	Elucidate the structure and synthesis of isoflavones.	(8)
	e	Describe the isolation and structure of cyanin chloride.	(7)

Z-Z-Z

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