

**PSG COLLEGE OF ARTS & SCIENCE
(AUTONOMOUS)**

**BSc DEGREE EXAMINATION MAY 2022
(Sixth Semester)**

Branch – CHEMISTRY

INSTRUMENTAL METHODS OF CHEMICAL ANALYSIS

Time: Three Hours

Maximum: 75 Marks

SECTION-A (10 Marks)

Answer ALL questions

ALL questions carry EQUAL marks

$(10 \times 1 = 10)$

1. The significant figures of 0.00567 and 2.34 are
 - a) 3,1
 - b) 5,1
 - c) 3,3
 - d) 6,3
2. The number of weight losses observed in the TG thermogram of $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ when heated to 900°C is
 - a) two
 - b) one
 - c) three
 - d) four
3. The number of modes of vibration for water molecule is
 - a) six
 - b) five
 - c) four
 - d) three
4. The region of frequency (in IR) for C=C stretching is
 - a) $3700\text{-}2500 \text{ cm}^{-1}$
 - b) $2500\text{-}2000 \text{ cm}^{-1}$
 - c) $2000\text{-}1600 \text{ cm}^{-1}$
 - d) $1600\text{-}1450 \text{ cm}^{-1}$
5. On the addition of a few drops of dilute aqueous NaOH solution, the UV absorption spectrum of pure $\text{C}_6\text{H}_5\text{OH}$ in $\text{C}_2\text{H}_5\text{OH}$ will show which one of the following characteristics?
 - a) λ value remains unchanged
 - b) λ value suffers a hypsochromic shift
 - c) the intensity and λ value remain unaffected
 - d) λ value suffers bathochromic shift
6. For which of the following groups $\pi \rightarrow \pi^*$ transition does not occur in UV-Visible spectroscopy?
 - a) Acetaldehyde
 - b) Nitromethane
 - c) Azomethane
 - d) Acetone
7. How many signals would you expect for toluene in NMR?
 - a) 9
 - b) 1
 - c) 2
 - d) 3
8. Pick out the incorrect statement about DPPH
 - a) Its g value is 2.0039
 - b) used as standard substance in ESR study
 - c) Gives three peaks with relative intensities 1:3:1 in the ESR spectrum
 - d) It generates free radical
9. Diffusion current is
 - a) directly proportional to concentration
 - b) indirectly proportional to concentration
 - c) equal to concentration
 - d) a limiting current
10. In polarography method of analysis, the current is measured as
 - a) migration current
 - b) eddy current
 - c) limiting current
 - d) diffusion current

Cont...

SECTION - B (25 Marks)

Answer ALL questions

ALL questions carry EQUAL Marks (5 x 5 = 25)

- 11 a How is accuracy determined by qualitative analysis?
 OR
 b Explain the thermometric titration of HCl and NaOH.
- 12 a How is force constant determined?
 OR
 b State and explain the mutual exclusion rule.
- 13 a Bring out the applications of UV spectroscopy in quantitative and qualitative analysis.
 OR
 b State and explain the laws of colorimetric analysis.
- 14 a Outline the various factors influencing the chemical shift in NMR spectroscopy.
 OR
 b Sketch and describe the block diagram of instrumentation of ESR spectroscopy.
- 15 a Explain the following, i) Residual current ii) Diffusion current.
 OR
 b Describe briefly about half wave potential.

SECTION - C (40 Marks)

Answer ALL questions

ALL questions carry EQUAL Marks (5 x 8 = 40)

- 16 a Discuss the methods of least squares and correlation coefficient with examples.
 OR
 b Discuss the principle, instrumentation and application of TGA.
- 17 a Distinguish between the IR and Raman spectroscopy.
 OR
 b Discuss the techniques and instrumentation of Raman spectroscopy.
- 18 a Outline the principle and instrumentation of UV spectroscopy.
 OR
 b Discuss the methods of colour measurements by colorimetric titration and Duboscq colorimeter.
- 19 a Discuss about the structural determination and kinetic studies using NMR spectroscopy.
 OR
 b Highlight of the following in ESR spectroscopy,
 i) g factor ii) hyperfine splitting
- 20 a Discuss the principle and instrumentation of polarography.
 OR
 b Outline the following.
 i) Organic polarography ii) Pulse polarography.

Z-Z-Z

END