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PSG COLLEGE OF ARTS & SCIENCE

(AUTONOMOUS)

BSc DEGREE EXAMINATION MAY 2025

(Sixth Semester)

Branch - CHEMISTRY

INORGANIC CHEMISTRY-II

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Time: Three Hours		Three Hours Maximu	Maximum: 50 Marks	
		SECTION-A (5 Marks) Answer ALL questions ALL questions carry EQUAL marks	$(5 \times 1 = 5)$	
1.	(i) (iii	hat is the general electronic configuration of the lanthanides? $(n-2)f^{1-14}(n-1)d^{1-10} ns^2$ (ii) $(n-2)f^{1-14}(n-1)d^{1-2} ns^2$ (iv) $(n-2)f^{1-14}(n-1)d^0 ns^2$		
2.	(i) . (iii	hich of following elements are include in actinide? Th to Lr (ii) Ac to Lr i) Ac to No (iv) Th to No		
3.	(i) (ii	In Geiger Muller Counter, one "court" is directly due to (i) a primary electron (ii) a secondary electron S (iii) a beta particle (iv) many electrons and ions		
4.	(i) (ii	How is the separation of uranium isotopes carried out? (i) Centrifugation (ii) Electrodialysis (iii) Electrophoresis (iv) Thermal diffusion		
5.	Which of the following is non aqueous solvent? (i) Benzene (ii) Ether (iii) CCl4 (iv) All of these			
SECTION - B (15 Marks) Answer ALL Questions ALL Questions Carry EQUAL Marks (5 x 3 = 15)				
6.	a)	Explain briefly has f-block elements are classified as Actinides. (OR)	Lanthanides and	
	b)	Summarize the uses of various lanthanide compounds.		
7.	a)	Outline the preparation of Transuranic Elements. (OR)		
	b)	Compare Lanthanides and Actinides.	-	
8.	a)	Describe the applications of radioactive radiation. (OR)		
	b)	Explain the group displacement law in briefly.		
9.	a)	Summarize the structure of isobars and isotones. (OR)		
	b)	Describe the structural features of Isotopes of Hydrogen.		
10.	a)	How are the classification of Carbides? (OR)		
	b)	Describe the structure features of LiAlH ₄ .		

SECTION -C (30 Marks)

Answer ALL questions
ALL questions carry EQUAL Marks

 $(5 \times 6 = 30)$

11.a) Analyze the Extraction of Lanthanides from Monazite sand.

(OR)

- b) Elucidate the Lanthanide contraction in detail.
- 12.a) Discuss the extraction and properties of Uranium.

(OR)

- b) (i) Elucidate the importance of colours and complex formation in actinides.
 - (ii) Enumerate the general properties of actinides.
- 13.a) Outline the classification of radioactive series.

(OR)

- b) Explain the function of Electroscope method and Geiger Muller Counter Method.
- 14.a) Discuss in detail about the detection of Aston and Dempster mass Spectograph.

(OR)

- b) Discover the separation of isotopes from Electromagnetic and Thermal diffusion Methods.
- 15.a) Outline the preparation, properties and uses of NaBH4.

(OR)

b) Discuss the general methods of preparation, properties and uses of Covalent

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