

PSG COLLEGE OF ARTS & SCIENCE
(AUTONOMOUS)

BSc DEGREE EXAMINATION MAY 2025
(Fifth Semester)

Branch – CHEMISTRY

INORGANIC CHEMISTRY-I

Time: Three Hours

Maximum: 50 Marks

SECTION-A (5 Marks)

Answer ALL questions

ALL questions carry EQUAL marks

(5 x 1 = 5)

- 1 An element in the d-block that shows maximum oxidation states is:
(i) Scandium (ii) Vanadium
(iii) Chromium (iv) Manganese
- 2 The coordination number of the central metal atom in $[\text{Fe}(\text{CN})_6]^{3-}$ is
(i) 4 (ii) 8
(iii) 6 (iv) 10
- 3 Which of the following ions will have the maximum magnetic moment?
(i) $[\text{Mn}(\text{H}_2\text{O})_6]^{4+}$ (ii) $[\text{Fe}(\text{H}_2\text{O})_6]^{2+}$
(iii) $[\text{Co}(\text{H}_2\text{O})_6]^{3+}$ (iv) $[\text{Cu}(\text{H}_2\text{O})_6]^{2+}$
- 4 What kind of ligand is the porphyrin ring in hemoglobin?
(i) Bidentate (ii) Monodentate
(iii) Tetradentate (iv) Hexadentate
- 5 In which compound does the metal-ligand π -backbonding play a significant role?
(i) $[\text{Fe}(\text{H}_2\text{O})_6]^{2+}$ (ii) $[\text{Ni}(\text{CO})_4]$
(iii) $[\text{AgCl}]^{2-}$ (iv) $[\text{Cu}(\text{NH}_3)_4]^{2+}$

SECTION - B (15 Marks)

Answer ALL Questions

ALL Questions Carry EQUAL Marks

(5 x 3 = 15)

- 6 a Explain the trend in ionic radii of the first transition series elements.
OR
b State Chelate and its give any one application.
- 7 a Explain Sidgwick's Atomic number Rule and its significance in understanding molecular structures.
OR
b Sketch and explain the d-orbital splitting pattern in the Octahedral complexes.
- 8 a What are the limitations in CFT.
OR
b Describe the trans effect with trans effect series.

Cont...

- 9 a Analyse the structure of hemoglobin and its function in oxygen transport.
OR
b Narrate the structure and function of Vitamin B₁₂.
- 10 a How will you preparation and structure of sodium nitroprusside.
OR
b Describe the preparation and structure of nitroso ferrous sulphate.

SECTION -C (30 Marks)

Answer ALL questions

ALL questions carry EQUAL Marks

(5 x 6 = 30)

- 11 a Explain the trend in metallic character and oxidation state of the first transition series elements.
OR
b Discuss the factors that influence the stability of metal complexes.
- 12 a Outline the Werner's theory of coordination complexes.
OR
b Justify the various applications of the CFSE.
- 13 a Distinguish between the CFT and VBT.
OR
b Explain the trans effect using pi-bonding theory.
- 14 a Highlight the role of essential and trace elements in biological system.
OR
b Analyze the sodium-potassium pump in biological system.
- 15 a Discuss the preparation, properties and bonding in ferrocene.
OR
b Discuss the preparation and bonding nature in [Cr(CO)₆] and [Fe₃(CO)₁₂].

Z-Z-Z

END