

Statistical Thermodynamics for Engineers
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Lecture 32
Hermite Polynomials as Vibrational Energy Mode Solution

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related to oscillator frequency.

Harmonic oscillator matches with Morse potential when $r \approx r_e$

$r < r_e \dots$ force is repulsive
 $r > r_e \dots$ force is attractive

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$Z_{vib} = Z_{rot} \cdot Z_{vib}$

$V(r)$ needs an expression most popular choice is Morse potential

$$V(r) = D_0 \left[1 - e^{-\beta(r-r_e)} \right]^2$$

Two fitting parameters

r_e : internuclear distance at thermodynamic equilibrium

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$r < r_e \dots$ force is repulsive
 $r > r_e \dots$ force is attractive
 as $r \rightarrow \infty$: D_e : dissociation limit
 $U(r) < D_e$ we have stable bond betw 2 nuclei

Welcome to lecture number 22 of this Statistical Thermodynamics for Engineers. So, previously in the previous section we found out that when you analyze the vibrational mode of energy we can use something called harmonic oscillator which is close to the actual morse potential. In the morse potential if you recall is given by these two fitting parameters out of that this parameter D_e is particularly important because it designates the dissociation potential.

So, the harmonic oscillator works very well when the distance r is within a particular band of the equilibrium inter nuclear radii in the nuclear distance and when r goes to infinity you have the dissociation. So, normally when your potential is less than D_e we have a stable bond between the two nuclei concerned.

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$$\frac{d^2 S}{dx^2} + \frac{2\mu}{\hbar^2} \left[E_{vib} - \frac{1}{2} k_0 x^2 \right] S = 0$$

$$S(x) = r R(r); \quad x = r - r_e \dots \text{assuming harmonic oscillator.}$$

$$\text{put } H(y) = S(x) \exp\left(\frac{y^2}{2}\right)$$

$$y = \left[\frac{2\pi\mu r}{\hbar} \right]^{\frac{1}{2}} x$$

$$\lambda = \frac{2E_{vib}}{h\nu}$$

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$$\lambda = \frac{2E_{vib}}{h\nu}$$

putting all these

$$\frac{d^2 H}{dy^2} - 2y \frac{dH}{dy} + (\lambda - 1)H = 0$$

This soln is given by Hermite polynomials of degree 'σ'.

So, let us now analyze the vibrational mode. So, this is given as $\frac{d^2 S}{dx^2} + \frac{2\mu}{\hbar^2} [E_{vib} - \frac{1}{2} k_0 x^2] S = 0$ where S is capital R . And x is $r - r_e$ this is all assuming harmonic oscillator. So, we assume harmonic oscillator and therefore this is the expression that we will write. So, this is basically your Sr and this is basically your x . So, now put $H(y) = S(x) \exp\left(\frac{y^2}{2}\right)$. So, y is equal to $\left[\frac{2\pi\mu r}{\hbar} \right]^{\frac{1}{2}} x$ and λ is equal to $\frac{2E_{vib}}{h\nu}$.

So, if we put all these things now in these expression. So, you have now we proposed three things. We have proposed that H is equal to $S(x) \exp\left(\frac{y^2}{2}\right)$ we have put y is equal to $\left[\frac{2\pi\mu r}{\hbar} \right]^{\frac{1}{2}} x$ and then we have put λ is equal to this. So, this

is like a wavelength so lambda is equal to this. So, as you can see over here now putting all these things together you have $d^2 H$ by $d y^2$ minus $2 y d H$ by $d y$ plus λH is equal to 0. So, this is the expression which is the reduced form of expression after we applied all the variable changes. So, this solution is given by hermite polynomials of degree let us assume v .

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$H_0(y) = (-1)^0 \exp(y^2) \frac{d^0}{dy^0} \exp(-y^2)$

For which continuity requires

$v = \frac{1}{2}(\lambda - 1) \quad v = 0, 1, 2, \dots$

Complete soln

$S_0(\alpha^{1/2} x) = \frac{1}{(2^0 0!)^{1/2}} \left(\frac{\alpha}{\pi}\right)^{1/4} \exp\left(-\frac{\alpha x^2}{2}\right) H_0(\alpha^{1/2} x)$

$d = \frac{2\pi h\nu}{h}$

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So, the solution to this particular equation is again ODE second order. So, $H_v y$ minus $1 v$ exponential y square $d v$ divided by $y v$ exponential minus y square, for continuity for which continuity requires v equal to half lambda minus 1, v equal to 0, 1, 2 dot, dot. So, this is called the hermite polynomials of degree v .

So, the complete solution is $S \sqrt{\lambda} \frac{1}{2^v v!} \alpha^{v/2} \exp(-\alpha x^2/2) H_v(\alpha^{1/2} x)$ where α is given as $2\pi \nu \gamma / \hbar$. So, this is the total expression that you get and this is the expression for your α where α is 2π nu frequency divided by \hbar . So, we get the point now.

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$$\psi(x) = \frac{1}{\sqrt{\lambda}} \exp\left(-\frac{\alpha x^2}{2}\right) H_v\left(\alpha^{1/2} x\right)$$

$$\alpha = \frac{2\pi \nu \gamma}{\hbar}$$
 Now $\lambda = \frac{2 E_{vib}}{h \nu}$

$$v = \frac{1}{2}(\lambda - 1) \rightarrow (\lambda - 1) = 2v$$

$$\text{or } \lambda = 2v + 1$$

$$\therefore (2v + 1) = \frac{2 E_{vib}}{h \nu}$$

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$$\therefore (2v + 1) = \frac{2 E_{vib}}{h \nu}$$

$$\text{or } E_{vib} = \left(v + \frac{1}{2}\right) h \nu$$

$$v = 0, 1, 2, \dots$$
 For $v = 0$, for the lowest vibrational energy level, zero point energy

$$E_{vib} = \frac{1}{2} h \nu$$

So, now your λ is 2 vibrational level divided by $h \nu$. So, v is equal to half λ minus 1 which leads to the fact that $\lambda - 1$ is equal to $2v$ or $\lambda = 2v + 1$. So, therefore $2v + 1$ is equal to 2 vibrational divided by $h \nu$ or E_{vib} is equal to $v + \frac{1}{2} h \nu$. v is equal to $0, 1, 2, \dots$. So, for v is equal to 0 that is for the lowest vibrational energy level zero-point energy is $E_{vib} = \frac{1}{2} h \nu$.

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For harmonic oscillator
$$E_{vib} = \frac{p^2}{2\mu} + \frac{kx^2}{2}$$

If $E_{vib} = 0$ means both p and x have to be equal to zero in violate Heisenberg uncertainty principle.

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$v = 0, 1, 2, \dots$
For $v=0$, for the lowest vibrational energy level, zero point energy
$$E_{vib} = \frac{1}{2} h\nu$$

For harmonic oscillator
$$E_{vib} = \frac{p^2}{2\mu} + \frac{kx^2}{2}$$

If $E_{vib} = 0$ means both p and x have to be equal to zero in violate

So, for the harmonic oscillator E_{vib} is equal to p^2 by 2μ plus Kx^2 by 2 . Now, if E_{vib} was equal to 0 say for example means both p and x both have to be equal to 0 , if that was true then this will invalidate Heisenberg's uncertainty principle so that is why the vibrational energy cannot be equal to 0 . So, the zero-point energy if you look at it for the lowest vibrational level is always $\frac{1}{2} h\nu$.

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If wave number

$$G(\nu) = \frac{E_{\text{vib}}}{hc} = (\nu + \frac{1}{2}) \omega_e$$
$$\omega_e = \frac{\nu}{c} = \frac{1}{2\pi c} \sqrt{\frac{k_0}{\mu}}$$

$\nu = 0, 1, 2, \dots$

vibrational quantum # uniquely determines the Hermite



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determines the Hermite polynomial

$\nu_{\text{vib}} = 1$

vibrational degeneracy = 1

$$S_0 = \left(\frac{\alpha}{\pi}\right)^{1/4} \exp\left(-\frac{\alpha x^2}{2}\right)$$
$$S_1 = \left(\frac{\alpha}{4\pi}\right)^{1/4} (2\alpha^{1/2} x) \exp\left(-\frac{\alpha x^2}{2}\right)$$
$$S_2 = \left(\frac{\alpha}{4\pi}\right)^{1/4} (2\alpha x^2 - 1) \exp\left(-\frac{\alpha x^2}{2}\right)$$


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$$S_1 = \left(\frac{\alpha}{4\pi}\right)^{1/4} (2\alpha^{1/2} x) \exp\left(-\frac{\alpha x^2}{2}\right)$$

$$S_2 = \left(\frac{\alpha}{4\pi}\right)^{1/4} (2\alpha x^2 - 1) \exp\left(-\frac{\alpha x^2}{2}\right)$$

$$S_3 = \left(\frac{\alpha}{4\pi}\right)^{1/4} \left(2\alpha^{3/2} x^3 - 3\alpha^{1/2} x\right) \exp\left(-\frac{\alpha x^2}{2}\right)$$

So, if wave number is $G v$ let us put it as $G v$ vibrational divided by $h c$ that is equal to v plus half into $\omega_e v$ equal to once again $0, 1, 2, \dots$. ω_e is equal to γ by c which is nothing, but the frequency k naught is like the spring constant equal to that. So, this is in the wave number space. So, vibrational quantum number uniquely determines the hermite polynomial or in other words the g vibration is equal to 1.

That essentially means that the degeneracy is 1 vibrational degeneracy is equal to 1. There is only one type of way that you can have a vibrational world. So, if you look at now few of these vibrational solutions of v of a half into x if you look at some of the solutions like for example S naught. S naught will be α by π one fourth exponential minus αx^2 by 2 S_1 which is α by 4π one fourth that is equal to 2α by half into x exponential minus αx^2 by 2.

S_2 is equal to α by 4π into one fourth then $2\alpha x^2 - 1$, exponential minus αx^2 by 2 and S_3 is equal to α by 4π one fourth into $2\alpha^{3/2} x^3 - 3\alpha^{1/2} x$ exponential minus αx^2 by 2. So, these are the few solutions of this vibrational, first few solutions for the harmonic oscillator model. So, now let us look at the kind of a roadmap.

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So rotational and vibrational energies of a diatomic molecule is modeled as

rigid rotor

fixed inter nuclear distance (r_e)

harmonic oscillator

mandates small vibrations around r_e (not large)

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rigid rotor

fixed inter nuclear distance (r_e)

harmonic oscillator

mandates small vibrations around r_e (not large)

for small r_e ensures that Morse potential is well approximated by harmonic oscillator

So, rotational and vibrational energies of a diatomic molecule is modeled as... We can have this (15:07). So, this is the rigid rotor which translates to fixed inter nuclear distance is r_e and on the other side you have the harmonic oscillator which basically it mandates what is this mandate? Mandate small vibration around r_e , remember small vibrations not large. So, this fits nicely into that.

Around r_e if this is r_e this is the well... so it is just vibrates around this particle level within a particular band. So, what it implies is that small r_e for small r_e ensures that most potentials is well approximated by a harmonic oscillator and that is in general that is it ties very well with the rigid rotor model because r is fixed invariant and therefore there is only a small amount of oscillations about r that is what ensures that it is oscillation is kind of well behaved.

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the steady state S. wave equation for an one-dimensional harmonic oscillator in coordinate x is given as

$$\frac{d^2\psi}{dx^2} + \frac{2\mu}{\hbar^2} \left(E_{\text{vib}} - \frac{1}{2}kx^2 \right) \psi = 0$$

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given as

$$\frac{d^2\psi}{dx^2} + \frac{2\mu}{\hbar^2} \left(E_{\text{vib}} - \frac{1}{2}kx^2 \right) \psi = 0$$

ψ_0 \rightarrow one dimensional harmonic oscillator equivalent to SO

ψ_0, ψ_1 are orthonormal



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ψ₀ → one dimensional harmonic oscillator equivalent to S₀

ψ₀, ψ₁ are orthogonal

proper normalization requires

$$\int_{-\infty}^{+\infty} \psi^* \psi dx = 1$$

For v=0

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ψ₀ = 1

vibrational degeneracy = 1

$$S_0 = \left(\frac{\alpha}{\pi}\right)^{1/4} \exp\left(-\frac{\alpha x^2}{2}\right)$$

$$S_1 = \left(\frac{\alpha}{4\pi}\right)^{1/4} (2\alpha^{1/2} x) \exp\left(-\frac{\alpha x^2}{2}\right)$$

$$S_2 = \left(\frac{\alpha}{4\pi}\right)^{1/4} (2\alpha x^2 - 1) \exp\left(-\frac{\alpha x^2}{2}\right)$$

$$S_3 = \left(\frac{\alpha}{4\pi}\right)^{1/4} (2\alpha^{3/2} x^3 - 3\alpha^{1/2} x)$$

So, that brings us... we will do one more small example say for a one dimensional the steady state so it will be a wave equation or for an one dimensional harmonic oscillator in Cartesian is given as one degree of freedom one dimensional psi divided by x square plus 2 mu h bar E vibrational minus half k naught x square psi is equal to 0.

So, hence the wave function is basically sigma v describing the one dimensional a harmonic oscillator. It describes the one-dimensional harmonic oscillator equivalent to S v. So, we want you to show that psi naught and psi 1 are orthogonal. So, first and foremost proper normalization requires minus infinity to plus infinity psi star psi d(x) which is equal to 1. So, if you just look at the numbers that we just pulled out these numbers you just look at those numbers over here what you have for v is equal to 0.

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$$\int_{-\infty}^{\infty} \psi_0^2 dx = \left(\frac{\alpha}{\pi}\right)^2 \int_{-\infty}^{\infty} \exp(-\alpha x^2) dx$$

$$= \left(\frac{\alpha}{\pi}\right)^2 \left[\frac{\pi}{\alpha}\right]^{1/2} = 1$$

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similarly for $v=1$

$$\int_{-\infty}^{\infty} \psi_1^2 dx = 4\alpha \left(\frac{\alpha}{\pi}\right)^2 \int_{-\infty}^{\infty} x^2 \exp(-\alpha x^2) dx$$

$$= 2\alpha \left(\frac{\alpha}{\pi}\right)^2 \left(\frac{\pi}{\alpha}\right)^2 = 1$$

Two wave functions are properly normalized.
Two wave functions are orthogonal

Integral will be psi square d x which is equal to alpha divided by pi into half. Writing the integral below plus infinity exponential minus alpha x square d x. So, this translates to if we do the integration of alpha by pi half pi by alpha half this is equal to 1. Similarly, for v is equal to 1 that is the second wave function that is minus infinity to plus infinity psi 1 square d x equal to 4 alpha by 4 pi raise to the power half. Again, we are writing the integral next line minus infinity to plus infinity x square exponential minus x square into d x.

This is equal to 2 alpha by pi half pi half 2 alpha 3 by 2 equal to 1. So, the two wave functions are properly normalized. So, now they are least two wave functions are therefore properly normalized that is good. So, now we recall that if two wave functions are orthogonal if they are orthogonal.

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in the ground

$$\int_{-\infty}^{+\infty} \psi_0 \psi_1 dx = 0$$

$$\int_{-\infty}^{+\infty} \psi_0 \psi_1 dx = \alpha \left(\frac{2}{\pi}\right)^{\frac{1}{2}}$$

$$\int_{-\infty}^{+\infty} x \exp(-\alpha x^2) dx = 0$$

integrand is

$$\int_{-\infty}^{+\infty} \psi_0 \psi_1 dx = \alpha \left(\frac{2}{\pi}\right)^{\frac{1}{2}}$$

$$\int_{-\infty}^{+\infty} x \exp(-\alpha x^2) dx = 0$$

integrand is clearly an odd function of 'x'!

So, if you take the product the two bringing this case that means from minus infinity to plus infinity it should be 0 psi 1 into d x should be equal to 0. So, what we can do is therefore multiply the wave functions. So, therefore we evaluate plus minus psi naught into psi 1 x which is equal to now alpha 2 by pi raise to the power of half you know I am writing the integral in the next line plus minus infinity exponential x into exponential minus half by x square into d x is therefore equal to 0 if we can validate this integrand is what is clearly a odd function x with the integrand is clearly an odd function of x. So, this kind of shows that this is how you analyze. So, we have seen that how the vibrational energy and the rotational energy and the translational energy can be done.

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Electronic energy mode
of atomic Hydrogen,
one proton one electron
like two body
analysis that we
did.

Separate external and
internal energy modes
by transforming to a (CM)
and relative coordinate
system
only viable internal mode
arises from the electronic
structure of hydrogen atom.

Now, we can extend this kind of an analysis for the electronic energy mode of atomic hydrogen extended. So, atomic hydrogen you know that you have now this rigid rotor, harmonic oscillator the degeneracy you know the energy level you know basically everything.

Now, atomic hydrogen is composed of what? One proton and you have one electron. So, it is almost like the two-body analysis that we did. Therefore, we can again separate the external and the internal energy by transforming to a center of mass and relative coordinate system except here the only viable electronic internal mode arises from the electronic structure of that atomic hydrogen.

So, here also we can do the same thing. We can separate the external and internal energy modes by transforming to a center of mass type of a system and relative coordinate system except you should just note here that one thing that happens that the only viable internal mode arises for the electronic structure of atomic hydrogen, electronic structure of hydrogen atom. So, in the next class what we are going to see we are going to introduce the same separation of variables and kind of try to justify that how this thing can be attained, how this thing can be analyzed further using our knowledge that we gained from the previous analysis. Thank you.