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**Lecture - 48**  
**Solution Resistance Effects**

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So in the previous class we have seen how to handle instabilities specifically when potential when the dc potential changes with time okay. Today I want to show you how we can handle solution resistance effect especially potentiostatic measurements okay. Earlier we have seen in galvanostatic measurements, solution resistance effects are easy to handle okay. Now we want to look how in case of potentiostatic measurements we can handle the situation okay.

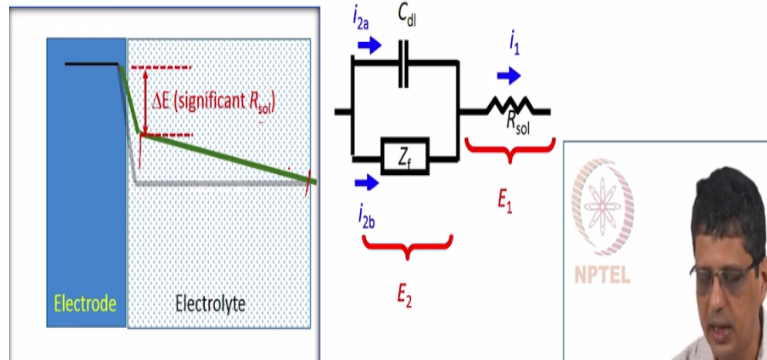
So I want to show 2 examples. One is for a simple electron transfer reaction and second is for a two step reaction with intermediate adsorbed species okay and we will also go through an example code for this.

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## NLEIS- $R_{sol}$ Effects

S. Santhanam, V. Ramani, S. Ramanathan, J. Solid State Electrochem. 16(3) (2012) 1019-1032

If  $R_{sol}$  is significant, then part of the applied potential drop occurs across interface. Remaining occurs in solution



So some of this work we have published in 2012 this is a schematic, illustrating how the potential drops occurs this electrode, electrolyte interface if we have negligible solution resistance we can say that all the potential drop occurs across the interface and then we can model this with a circuit given here that is double layer capacitance being parallel with the faradaic impedance.

In case the solution resistance is significant then part of the applied potential occurs across the solution and part of the applied potential occurs across the interface. So you can say that the same potential drop as you measure or as you apply occurs but then only part of it occurs at the interface this  $\Delta E$  in case of significant  $R_{sol}$  the remaining potential drop occurs across the solution.

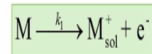
So we have to use a slightly different circuit which is to add this  $R_{sol}$  to this schematic or to this equivalent circuit representation. So when we have this what are the governing equations.

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## NLEIS- $R_{sol}$ Effects. Electron Transfer Reaction

Development of equations

Example with simple electron transfer reaction



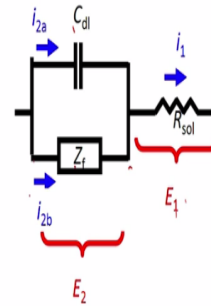
$$i_{2a} = C_{dl} \frac{dE_2}{dt}$$

$$i_{2b} = Fk_{10} e^{\frac{\hbar}{2}(E_{2dc} + E_{2ac})} = Fk_{10} e^{\frac{\hbar}{2}E_2}$$

$$i_1 = \frac{E_1}{R_{sol}} = \frac{E - E_2}{R_{sol}}$$

$$i_1 = i_{2a} + i_{2b}$$

$$C_{dl} \frac{dE_2}{dt} + Fk_{10} e^{\frac{\hbar}{2}E_2} = \frac{E - E_2}{R_{sol}} \quad \checkmark$$



In case of a simple electron transfer reaction metal dissolving into a solution with a release of an electron. The current through this double layer capacitance is represented as  $i_{2a}$  current through this faradaic impedance is represented by  $i_{2b}$ . And of course the current through the solution resistance is given as  $i_1$  okay across the double layer capacitance the current is given by capacitance multiplied by  $dE/dt$  which is rate of change of potential with respect to time and for the faradaic impedance we know how to relate this current to the potential.

In general, we would write it as  $E$  in this case we would write it as  $E_{dc} + E_{ac}$  but that is only  $E$  occurring across this interface. So we call this part as  $E_2$  we call this potential we call this potential drop as  $E_1$  and the sum of  $E_1$  and  $E_2$  is what we apply. In this case this is more general representation and this is we represent this as  $dc + ac$ , but we do not mean that all the time that  $ac$  is a sinusoidal.  $Dc$  is the average value  $ac$  is any value that is fluctuating.

So if we measure  $E_2$  it might be sinusoidal it might be another periodic pattern. We will say that this is  $E$  this is  $t$ . The average value is what we would take as  $dc$  and the value that is fluctuating we would call that as  $ac$  and then of course we have to employ Fourier transform to calculate what is the response of fundamental, what is the response of other harmonics etcetera.

Now the  $i_1$  can be related to the  $E_1$ . We are seeing the simple expression  $E_1/R_{sol}$  solution is  $i_1$ .  $E_1$  I can write it as  $E$  that is the applied potential- the potential drop across this interface. So whatever current that comes through the double layer and that comes through the faradaic process that sum of this algebraic sum of this should be = current that is passing through the

solution.

So we can write  $i_1$  as  $i_{2a} + i_{2b}$ . We have an expression for  $i_1$  written in terms of  $E_2$ . We have an expression for  $i_{2a}$  written in terms of  $E_2$  expression for  $i_{2b}$  written in terms of  $E_2$ . So we can put them together we have a ordinary differential equation incorporating the solution resistance effect we know  $E$  we control  $E$  we do not know  $E_2$  we have to find the  $E_2$  value, we have to find the  $E_1$  value so that we can estimate the  $i_1$  value  $i_1$  is the current that measure okay.

So we basically apply a dc bias and estimate the potential drop across the interface. We can only estimate we cannot measure it in the experiments. If we can minimize a solution resistance, we would just minimize it or eliminate it and then measure only the potential drop across the interface or control the potential drop across the interface. Since we cannot do that for whatever reason we have to estimate it.

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**NLEIS-  $R_{sol}$  Effects. Electron Transfer Reaction**

Steady state: Under application of DC potential only

$$C_{dl} \frac{dE_2}{dt} + Fk_{10} e^{b_1 E_2} = \frac{E - E_{2c}}{R_{sol}}$$

$$C_{dl} \frac{dE_{2dc}}{dt} + Fk_{10} e^{b_1 E_{2dc}} = \frac{E - E_{2dc}}{R_{sol}}$$

0

Solve the nonlinear algebraic equation using numerical methods, to get  $E_{2dc}$

$k_{10} = 10^{-12} \text{ mol cm}^{-2} \text{ s}^{-1}$ ,  $b_1 = 20 \text{ V}^{-1}$ ,  $E_{2c} = 0.4 \text{ V vs. OCP}$ ,  $C_{dl} = 10 \text{ } \mu\text{F} / \text{cm}^2$

$R_{sol} = 1 \text{ } \Omega / \text{cm}^2$        $R_{mt} = 100 \text{ } \Omega / \text{cm}^2$

$E_{2dc} = 0.4 \text{ V vs. OCP}$        $E_{2dc} = 0.38 \text{ V vs. OCP}$

$M \xrightarrow{k_1} M_{sol}^+ + e^-$

So if I apply dc potential and do not apply any ac potential that means  $E_1$  is going to be constant. Constant meaning it is going to be invariant with respect to time  $E_2$  is going to be constant. We can say that this goes to 0  $dE_1/dt$  goes to 0 and the remaining terms can be solved numerically it is an algebraic equation non linear algebraic equation, but you can solve this and get the value for  $E_2$  dc.

So it is a non linear algebraic equation you can use the numerical method to get  $E_2$  dc so it is an example.  $K_{10}$  is  $10^{-12}$  moles per centimeter square per second  $b_1$  is 20 volts

inverse dc potential that we apply is 0.4 volts with respect to OCP and we say that double layer capacitance is 10 microfarad per centimeter square all right and if the solution resistance 1 ohm's centimeter square.

You can solve this I am not showing the code for this but the value of E2 dc is going to be pretty much 0.4 it would not be exactly 0.4 but it will be close to 0.4 that means potential drop most of it occurs across the interface. If I change the solution resistance to 100 ohm's centimeter square about 20 millivolts across the solution resistance 380 millivolts still occurs across the interface.

So make the solution resistance even larger then you would see that more potential drop occurs the solution. Okay for a given case given solution resistance we apply a dc bias and on top of that we apply ac potential right.

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**NLEIS-  $R_{sol}$  Effects. Electron Transfer Reaction**

Apply sinusoidal potential: Visualize  $E_2$  as a sum of DC and AC (not necessarily at  $\omega$ )

$$E_2 = E_{2dc} + E_{2ac}$$

$$C_d \frac{d(E_{2dc} + E_{2ac})}{dt} + Fk_{10} e^{A(E_{2dc} + E_{2ac})} = \frac{E_{2dc} + E_{2ac} \sin(\omega t) - (E_{2dc} + E_{2ac})}{R_{sol}}$$



$$C_d \frac{d(E_{2ac})}{dt} + F \left[ k_{10} e^{A E_{2dc}} \right] e^{A E_{2ac}} = \left( \frac{E_{2dc} - E_{2ac}}{R_{sol}} \right) + \frac{E_{2ac} \sin(\omega t) - E_{2ac}}{R_{sol}}$$

Nonlinear equation. Can be linearized for small  $E_{2ac}$  and solved analytically

$$\frac{d(E_{2ac})}{dt} + \frac{Fk_{1dc}}{C_d} e^{A E_{2dc}} + \frac{E_{2ac}}{C_d R_{sol}} = \frac{E_{2dc} - E_{2ac}}{C_d R_{sol}} + \frac{E_{2ac} \sin(\omega t)}{C_d R_{sol}}$$

Arbitrary  $E_{2dc}$ : Numerical integration

$M \xrightarrow{k} M_{sol}^+ + e^-$

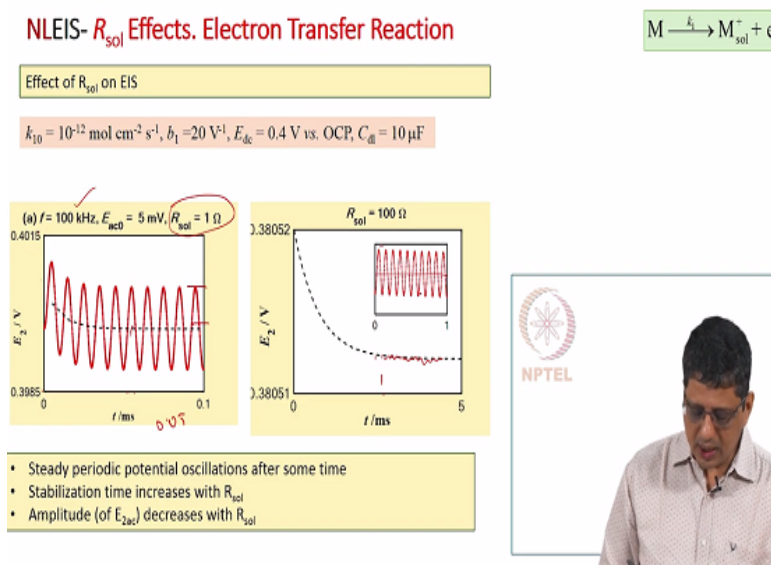
Now we have to visualize the E2 the potential drop across the interface as a sum of dc and ac but here I use the term ac in a loose fashion. It is not necessarily sinusoidal oscillation at frequency of omega. So you can write the equation instead of E2 you can write it as E2 dc+ E2 ac and this is what you would get. This is the potential that we apply a dc bias + an ac potential and that is at sinusoidal frequency omega we control that.

Of course by definition E2 dc does not vary with time therefore that goes away the remaining terms can be grouped k10 b1 E2dc can be written as k1 dc this Edc and E2dc can be grouped here sin omega t and E2ac can be grouped here. It is essentially a non linear differential

equation. You can linearize this for a small  $e_2ac$  and solve it analytically, but practically we want to solve it from the cases where we have large amplitude perturbation so you have to use numerical method.

So if you have an arbitrary value that is possibly a large value of  $E_2ac$  you have to use numerical integration.

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Now for a given set of kinetic parameter right if I apply a frequency of 100 kilohertz for a solution resistance of 1 ohm for a simple electron transfer reaction the potential oscillation stabilize after sometime. Earlier we have seen that when you have adsorbed intermediate the current oscillation will stabilize after sometime right.

Out here even for simple electron transfer reaction if we have a significant solution resistance when you apply a dc potential+ an ac potential you will not get as steady periodic current response right away. It will drift a little and then stabilize after sometime okay. In case of 1 ohm solution resistance it takes maybe 0.05 millisecond to stabilize. In case of 100 volt solution resistance it takes maybe 3 or 4 milliseconds to stabilize for this set of kinetic parameter at this dc potential.

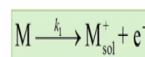
What that means is how long does it take to stabilize depends on kinetic parameter, it depends on dc potential, it also depends on the solution resistance value. It typically increases with solution resistance value and if I have large solution resistance for same set of kinetic parameter that means for the same reaction this magnitude if I just see this 100 kilohertz here

as well as 100 kilohertz here.

Although we can write it here it is the same 5 millivolt ac amplitude and 100 kilohertz frequency. This oscillation is going from 0.385 to 4015 this goes on 38051 to 38052 it means in fact it much, much smaller than this scale. I just shown you the average and the oscillation are shown in the inset. So when you increase a solution resistance the current oscillations are smaller that is expected also.

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### NLEIS- $R_{sol}$ Effects. Electron Transfer Reaction



Stabilization time

- is independent of frequency
- Decreases slightly with  $E_{ac0}$
- Strongly depends on kinetic parameter and  $C_{dl}$  values also  $E_{1/2}$

How much of  $E_{ac0}$  is applied across the interface? i.e.  $(|E_{2ac}|/E_{ac0}) = ?$

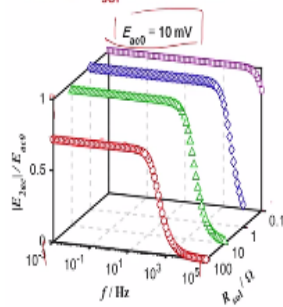


Now how long does it take to stabilize it is independent of frequency. It does vary with  $E_{ac0}$  and it is strongly depends on the kinetic parameter values, double layer capacitance values and the dc potential at which we acquire the data. Now if I apply an ac potential not all of that falls across the interface a part of it falls across the solution. Now how much of the  $E_{ac0}$  applied actually falls across the interface.

So if I take the magnitude of  $E_{2ac}$  and take the ratio of that magnitude to  $E_{ac0}$  ideally I should get 1 all the potential that we apply should fall across the interface and that will happen only when solution resistance is 0.

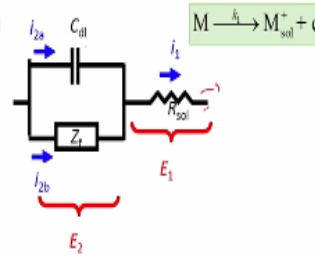
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### NLEIS- $R_{sol}$ Effects. Electron Transfer Reaction



How much of  $E_{ac0}$  is applied across the interface? i.e.  $(|Z_{ac}|/E_{ac}) = ?$

- Depends on frequency
- At high frequencies, current mostly flows through double layer capacitor and not via Faradaic process
- Double layer offers very little impedance. Total impedance is roughly equal to solution resistance
- At low frequencies, current mostly flows via Faraday process. Total impedance is equal to Faradaic impedance + solution resistance



Now we can do the calculation for a variety of solution resistance values and for various frequencies okay for one perturbation amplitude 10 millivolt this is what we see. At low frequencies it is more independent after frequency at high frequency it is very small and at intermediate frequency of course it falls from high value to a low value and ideal value is of course 1. When solution resistance is small pretty across this frequency region it is more or less 1.

When solution resistance increase less and less applied potentials falls across the interface and more of it falls across the solutions. So what fractions of it occurs across the interface that depends on the frequency. At high frequencies what happens is this pretty much all the current goes through the double layer capacitance because it offers negligible impedance. When this impedance is negligible.

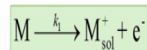
And this is anyway a finite number you can say total impedance is more or less solution resistance that means more or less all the applied potential will fall across the solution resistance. When you go to low frequencies this double layer capacitance offers very large impedance. So most of the current will flow through this faradaic process. Faradaic process of course the impedance value depends on the kinetic parameter, depends on the mechanism, depends on the dc potential dc bias.

And this is comparable or larger than this you would see most of the potential drop across the interface. So that is why at low frequencies more of this applied potential falls across the interface at high frequencies it mostly falls across the solution.



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### NLEIS- $R_{sol}$ Effects. Electron Transfer Reaction



Effect of  $R_{sol}$  on EIS

Wait for sufficient time, then record  $E_2$  as a function of time

Calculate current  $\rightarrow$

Apply FFT

Record current @  $\omega$  (and @ 'DC',  $2\omega$  etc.)

Magnitude of impedance =  $E_{ac0}/|i_\omega|$

Phase of impedance =  $0 - \text{phase}(i_\omega)$

$$i_1 = \frac{E_1}{R_{sol}} = \frac{E - E_2}{R_{sol}} = \frac{E_{dc} + E_{ac0} \sin(\omega t) - E_2}{R_{sol}}$$

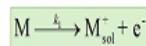


So the way we do this is apply a ac potential wait for some time to get steady periodic response and then in simulations we can record  $E_2$  as a function of time. Once you record  $E_2$  you can calculate  $E_1$  and then calculate the  $i_1$  that is basically calculate the current and then apply Fourier Transform mark the current at omega fundamental as well as at 2 omega, 3 omega dc any other frequency you want you can calculate that and record that.

You can calculate the magnitude of impedance as  $E_{ac0}/I$  omega here meaning that current at the fundamental omega and the phase of course 0-0 is for the potential. This is the phase of I omega you can calculate the impedance. This essentially source the equation relating  $E_2$  to the  $i_1$ .

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### NLEIS- $R_{sol}$ Effects. Electron Transfer Reaction



Matlab ® code. Set parameter values

```
F = 96485;

Rsol = 100; % Ohm
Eac0 = 0.1; %V
Wait_time = 0.01;%seconds

Edc = 0.5; % 0.1 means, 0.1 volt above OCP

k10 = 1e-12; b1 = 20;
Cdl = 10e-6; % 10e-6 means 10 microFarads

frequencyset = 10.^(5:-0.1:-3);

p1(1)=k10;
p1(2)=b1;
p1(3)=Edc;
p1(4)=Rsol;
```



So as an example 100 ohm solution resistance  $E_{ac0}$  is 100 millivolt you can say (0) (15:51) is 10 milliseconds we apply a dc potential we set the kinetic parameters values and set the frequency range. We want to pass the kinetic parameter values as vector.

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**NLEIS-  $R_{sol}$  Effects. Electron Transfer Reaction**

Matlab ® code. Find  $E_{2dc}$

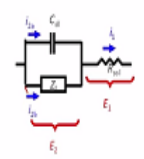
```
options = optimset('TolX',1e-12,'Display','off');
[E2dc,fval]=fminbnd(@(x)FindE2dcMech0(x,p1),0,Edc);
```

```
function residuel = FindE2dcMech0(x,p1)
F=96485;
k10=p1(1);
b1=p1(2);
Edc=p1(3);
Rsol=p1(4);


E2dc = x;

k1DC = k10*exp(b1*E2dc);
i2b = F * k1DC ;
i1 = (Edc - E2dc)/Rsol;
y = (i1-i2b);
residuel = y^2;
end
```

$M \xrightarrow{k_1} M_{sol}^+ + e^-$



$(i_1 - i_2) = 0$

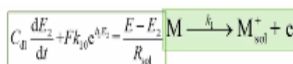


And then first find the dc potential occurring across this interface and we need to give that as the initial value for the integration okay. So in this case I have used the function called fminbnd. So essentially what we do is to take the dc equation which is  $i_1 - i_2a - i_2b = 0$ .  $i_1$  is  $E_{dc} - E_{2dc} / R$  solution  $i_{1b}$  faradaic current is  $F k_{1DC}$  and  $i_{2a}$  is 0 because when you apply dc potential no current passes through the double layer capacitance.

Ideally this should be 0  $i_{2b}$  so numerically we want to calculate the residual and then take the square of that, that should become 0 or minimum here.

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### NLEIS- $R_{sol}$ Effects. Electron Transfer Reaction



Matlab ® code. Integrate to find  $E_2$ .

Apply  $E_{dc} + E_{ac} \sin(\omega t)$ , wait for sufficient time (i.e. integrate for sufficient time) and then continue integration

```
options = odeset('RelTol',1e-12,'AbsTol',1e-12);
[tnew_row,E2_row] = ode15s(@(x,y) dE2Mech0(x,y,p),t,E2_init,options);
```

```
function dE2 = dE2Mech0B(t,E2,p1)
... parameters are retrieved from vector p1
dE2 = zeros(1,1);
k1 = k10 * exp(b1 * E2);
omega = 2*pi*frequency;
E = Edc + Eac0 * sin(omega*t);
part1 = (E-E2) / Rsol / Cdl;
part2 = -F * k1 / Cdl;
dE2(1) = part1 + part2;
end
```

Note: The equations are stiff, require ode15s instead of ode45

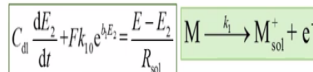


$$\frac{dE_2}{dt} = \left( \frac{E - E_2}{R_{sol}} - \frac{F k_1}{C_{dl}} \right) \frac{1}{C_{dl}}$$

And once you find the  $E_{2dc}$  use that as initial value and you have integrate this equation which is basically take this equation rearrange as  $dE_2/dt$  as  $E - E_2/R$  solution-  $F k_1$  the entire thing/ $C_{dl}$ . So we take the parameter we need to use ode15s because it is a stiff equation we find the value of  $k_1$ , we write the value of  $E$  the first part we write it as  $E - E_2/R$  solution/ $C_{dl}$  second part we write it as  $F k_1/C_{dl}$  and this is the derivative.

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### NLEIS- $R_{sol}$ Effects. Electron Transfer Reaction



Matlab ® code. Calculate current, analyze with FFT

```
E1 = Edc + Eac0 * sin(2*pi*frequency*t) - E2;
current = E1/Rsol;
```

```
n = length(current);
current = current(1:(n-1));
[YfreqDomain,frequencyRange] = positiveFFT(current,Fs);
```

```
LookatIndex = 2;
valFFT = YfreqDomain(NumLoops * (LookatIndex-1)+1); iac0 =
abs(valFFT)*2; % calculate current magnitude
phiac0 = angle(valFFT)+(pi/2); % and current phase
absImpedance = Eac0/iac0; % calculate impedance magnitude
phaseImpedance = -phiac0; % and phase
```

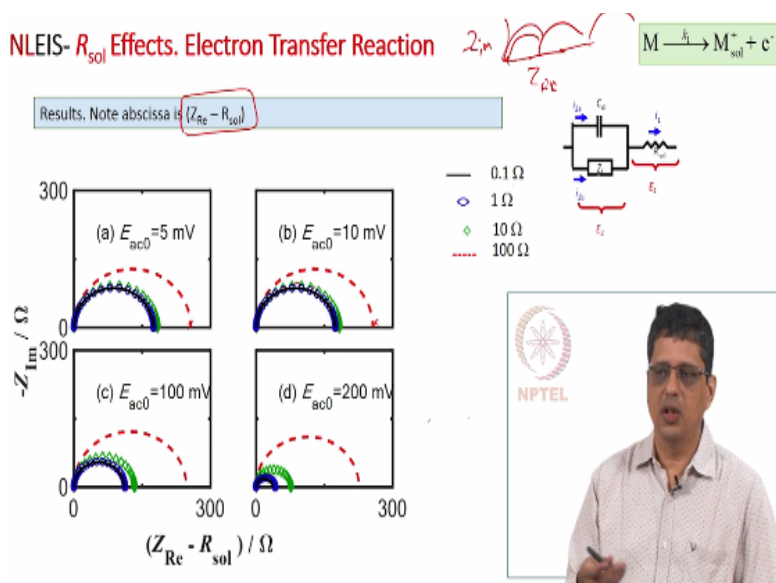


$$E_1 = E - E_2$$

$$i = i_1 = E_1 / R_{sol}$$

And after you integrate this you can estimate the value of  $E_1$  saying  $E_1$  is  $E - E_2$  current of course is  $I$  that is  $i_1 = E_1/R$  solution and then the standard procedure of applying Fourier Transform. So we calculate the impedance at 1 frequency if we repeat it for  $n$  number of frequencies we can get the spectral.

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So when you plot it I can vary the  $R_{sol}$  I have tried 0.1 ohm centimeter square after that 1 ohm, 10 ohm, 100 ohm actually this should be ohm centimeter square all right. When we plot normally we will plot  $-z$  imaginary/ $z$  real that means 0.1 ohm would probably would start here 1 ohm would start somewhere here 10 ohm would be much later. 100 ohm is going to be much, much later. In order to compare that what I have done is after calculating all the impedance values I have subtracted the  $R_{sol}$  and then plot it that in the abscissa.

Ordinate of course I keep  $-z$  imaginary. So what you see is the following you can calculate this for small or large amplitude perturbation using this methodology and you can calculate it for small solution resistance or a large solution resistance value. Now when I increase the solution resistance value obviously the measured impedance increases. What that means is if I have a low solution resistance I have a spectrum like this if I have a large solution resistance.

In the beginning we said it is just going to shift to the right that is correct when you have an electrical circuit. When you have a reacting system it is not just going to shift to the right it is also going to change in magnitude. It is showing an increase here it may not always be an increase. So what it means if solution resistance is not just going to shift the impedance spectrum it is also going to make changes to the values other than just adding  $R_{sol}$ .

But then it takes quite some time to do the integration and get the results and it has to be done pretty much by numerical methods for most of the reactions. There may be 1 or 2 exceptions but by and large you will have to use numerical methods which means give a set of kinetic

parameters, given a solution resistance value to generate the spectrum if we have an analytical expression in a computer.

We can generate that in a second may be much less than that. If we use the numerical method, it will take probably half an hour or so. So if you want to change the parameter and optimize it is not practical to do in this method. This is the right method but it is very time consuming. So what we end up doing is we generate the spectrum assuming that the solution resistance is negligible and then add the solution resistance.

Although we know that is not the best way practically that is probably the only way we can do right now. If you are able to get an analytical expression, then we can use the correct expression or if we get faster code or faster computer maybe we can try that.

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**NLEIS-  $R_{sol}$  Effects. Two step reaction with adsorbed intermediate**

Example with two step reaction. Basic equations

Faradaic current

$$i_{2b} = Fk_1(1 - \theta_1)$$

$$k_1 = k_{10}e^{k_1E_1}$$

Mass balance equation

$$\Gamma \frac{d\theta_1}{dt} = k_1(1 - \theta_1) - k_2\theta_1$$

Total current balance

$$C_{dl} \frac{dE_2}{dt} + [Fk_{10}e^{k_1E_1}(1 - \theta_1)] = \frac{E - E_2}{R_{sol}}$$

Chemical reactions:

$$M \xrightarrow{k_1} M_{ads}^+ + e^-$$

$$M_{ads}^+ \xrightarrow{k_2} M_{sol}^+$$

Handwritten notes:

$$\theta = \theta_1 + \theta_2$$

$$E = E_1 + E_2$$

Now what happens when you have an intermediate species it is an example two step reaction with an adsorbed intermediate. So the same electrical circuit equivalent circuit the faradaic current I would write it as  $Fk_1(1 - \theta)$  for this  $\theta$  is the fractional surface coverage of the adsorbed intermediate and  $k_1$  is related to the potential as  $k_{10}e^{k_1E_1}$  it is not  $k_{10}E_1$ . Mass balance equation of course we are seeing this before it is  $\Gamma \frac{d\theta}{dt} = k_1(1 - \theta) - k_2\theta$ .

The first step forms  $\theta$  second step removes  $\theta$  so we have  $k_1$  and  $-k_2$  and the total current  $i_{2a} + i_{2b}$  is  $i_1$ . So earlier we have written this  $i_{2a}$  as  $C_{dl} \frac{dE_2}{dt}$  we have written that. The second expression because the reaction is different here it is going to  $Fk_1(1 - \theta)$  and on the right hand side of course  $i_1$  is related to the  $E_1$  as  $\frac{E - E_2}{R_{sol}}$ . Now you can

understand you can write the dc expression for this solve it get the steady state value.

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**NLEIS-  $R_{sol}$  Effects. Two step reaction with adsorbed intermediate**

$$M \xrightarrow{k_1} M_{ad}^+ + e^-$$

$$M_{ad}^+ \xrightarrow{k_2} M_{ox}^+$$

$$\Gamma \frac{d\theta_1}{dt} = k_1 (1 - \theta_1) - k_2 \theta_1$$

$$C_a \frac{dE_{dc}}{dt} + [Fk_1 e^{\delta E_{dc}} (1 - \theta_1)] = \frac{E - E_{1/2}}{R_{sol}}$$

Steady state values of  $E_{dc}$  and  $\theta_{ss}$ . Solve the following nonlinear algebraic equations simultaneously

$$\theta_{ss} = \frac{k_1}{k_1 + k_2}$$


$$\frac{E_{dc} - E_{1/2}}{R_{sol}} = Fk_1 e^{\delta E_{dc}} (1 - \theta_{ss})$$

To get impedance: Apply sinusoidal wave,  $E = E_{dc} + E_{ac} \sin(\omega t)$

Solve mass and charge balance equations simultaneously

Wait for sufficient time, and then record steady periodic results

Calculate current, use FFT,...



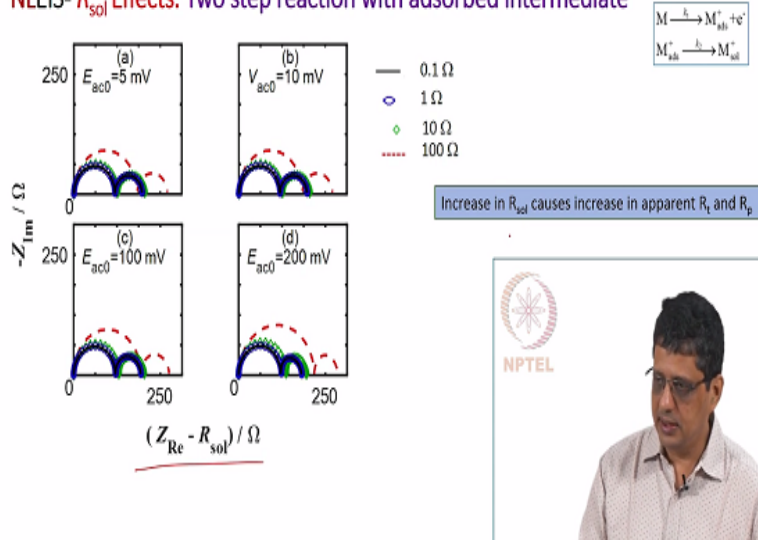
In order to get the steady state value, we say this goes to 0, this goes to 0. We have 2 equations here  $k_1$  depends on  $E_{dc}$ . So we have 2 equations to our notes we do not know  $E_{dc}$  and we do not know steady state value of  $\theta$ . You can solve this. These are 2 algebraic equation non linear equation, but you can solve them and then get the values of  $E_{dc}$  and  $\theta_{ss}$  use that as the starting point or initial value and integrate these 2 equation simultaneously.

So we have independent value as time we have  $E_2$  and  $\theta_1$  as the dependent variable. We have  $\theta_{ss}$  and  $E_{dc}$  as the initial points. So we have to charge mass and charge balance equation simultaneously. Again you have to see how long it takes for it to stabilize wait for sufficient time and then takes steady periodic results and record you get a value of  $E_2$  as a function of time  $\theta$  as a function of time.

We need only  $E_2$  to calculate  $E_1$  and we need only  $E_1$  to calculate current we do not need to use the  $\theta$ , but we will have to integrate them together to get the value of  $E_2$  and then we go through the same procedure right get the current value apply FFT get the ratio of potential to current with the impedance.

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### NLEIS- $R_{sol}$ Effects. Two step reaction with adsorbed intermediate

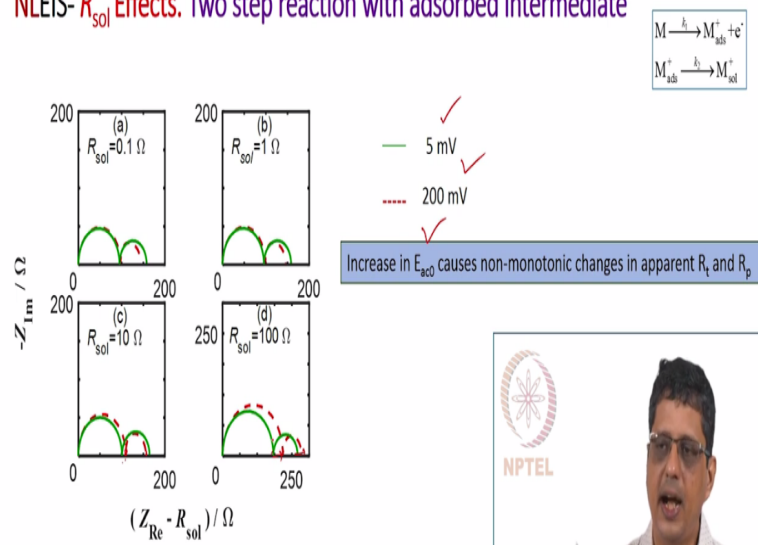


I am not showing you the code I am just showing you the final results this case for this given set of kinetic parameter you get 2 capacitive loops. When you change the solution resistance here it increases, but I also want you to note the following. I have plot a different perturbation amplitude here you can replot it for a given solution resistance different amplitudes and in all cases we have plotted real part of impedance after subtracting the solution resistance.

Now increase in  $R$  solution causes an increase in the apparent value of  $R_t$  and  $R_p$  at any potential amplitude.

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### NLEIS- $R_{sol}$ Effects. Two step reaction with adsorbed intermediate



Now if I fix the solution resistance and replot that so that I plot only 5 millivolt and 200 millivolt ac you can also plot the intermediate ones, but this is sufficient to illustrate. When you look here when I increase the ac potential perturbation charge transfer resistance seems



to increase for this. Polarization resistance seem to decrease. At 100 ohm charge transfer resistance increases polarization resistance increases.

So what that means is when you change the solution resistance, when you go to large amplitude perturbation you do not always have a monotonic trend okay. Increase in  $E_{ac0}$  causes non monotonic changes in apparent  $R_t$  and  $R_p$  values. Some cases it looks like it is more or less same some cases it looks like increases some cases it may show a decrease, but it will cause a change.

All that it means is you cannot just take a solution resistance and add it at the end.

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### NLEIS- $R_{sol}$ Effects. Summary

#### Potentiostatic measurements

- If  $R_{sol}$  is significant, the impedance response can be calculated using numerical integration for any mechanism
- Depending on the kinetic parameter values, solution resistance and applied dc bias, the spectra will be altered.
- Effect of  $E_{ac0}$  on  $R_p$  and  $R_t$  depend on the mechanism, kinetic parameters and  $R_{sol}$  value.

#### Galvanostatic measurements

- Modeling is simpler compared to that of potentiostatic measurements



So all this is done for n potentiostatic measurements right. When solution resistance is significant you can calculate the impedance response using numerical method. You have to use integration for pretty much any mechanism and depending on the dc bias, depending on the kinetic parameter values, depending on the mechanism, depending on the solution resistance the measured spectrum or simulated spectrum will change.

And it is not always monotonic change even for the limiting values of polarization resistance and charge transfer resistance. Whether it is linear or non linear regime (()) (27:34) it does not matter you cannot say that it is going to be monotonic okay. We have seen earlier that in case of galvanostatic measurement it is actually lot easier or simpler to model this you have to just model the interface and add the solution resistance a bit late.



So it is lot easier than doing it in the n potentiostatic measurement when we talk about simulations. In terms of experimental measurements there is no extra difficulty incorporating solution resistance effect. In fact, is harder to reduce a solution resistance in some cases okay. We will stop here today.