

Microscale Transport Processes
Prof. Somenath Ganguly
Department of Chemical Engineering
Indian Institute of Technology, Kharagpur

Lecture No. # 38
Electrohydrodynamic Atomization

I welcome you to this lecture of electrohydrodynamic atomization, which is also commonly referred as electro spray.

(Refer Slide Time: 00:35)

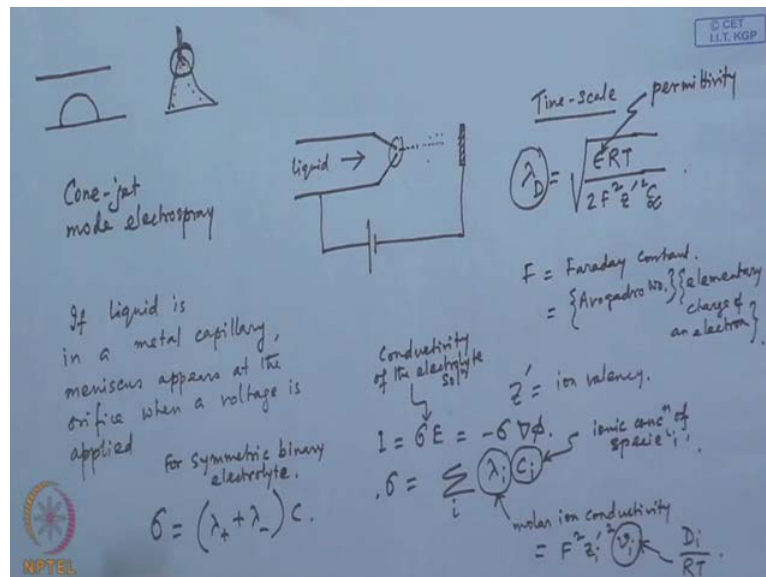
Electrohydrodynamic atomization

- Deformation of meniscus near a charged plate
- Counter-ions in double layer driven along the interface towards the meniscus tip
- High charge density at the meniscus tip
- Charge repulsive force > Local surface tension
=> Coulombic fission and draining of charge by extrusion of liquid jet

Use of

- DC electric field
- AC electric field

(Refer Slide Time: 00:48)



What we have here in this slide is, there is a deformation of meniscus near a charged plate. What we mean by this is, if there is a liquid sitting there, and if you bring a charged plate next to it, you see that the meniscus is deformed, by that what we mean is that you will see that the meniscus takes the shape like. This deformation of meniscus near a charged plate that happens, because counter ions in double layer driven along the interface towards the meniscus tip. I mean the this one immediate event that happens, because of this deformation of meniscus is that counter ions is, counter ions in double layer driven along the interface towards the meniscus tip.

And what eventually, this will lead to is a high charge density near this meniscus tip, high charge density at this meniscus tip. When this charge repulsive force, when this is greater, when this charge, if this repulsive force if I look at the slide, when this charge repulsive force is greater than the local surface tension, then we generally see something called a columbic fission and draining of charge by extrusion of liquid jet. So, what eventually, what this means is that you have a droplet, and that has because of the presence of this electric field, there is a high charge density at the meniscus tip.

And then this meniscus tip, there would be this charge repulsive force within this tip when it is higher than the local surface tension this droplet cannot hold that portion. There would be a Columbic fission and the only way, the charge can drain is, if there is a liquid

jet that excludes from this tip. Now, this idea has been used in something called a cone jet mode electro spray. What is done in this cone jet mode electro spray is, there is a metal capillary, this is the end of the cone and then there is a grounded plate. Now, you have here, this is the electric; this is the circuit that you have. So, what you see here is that the liquid jet emanates from this tip and this liquid jet breaks down into small droplets.

So, what you have here. So, what we are trying to do here is; what we are saying here is if liquid is coming from this side, if liquid is in a metal capillary, then a conical meniscus appears at the orifice I mean to say this place. At the orifice, when a voltage is applied across this point and this point. So, then automatically there would be a meniscus appearing. So, what we see here is that; what we will see here at the, this point is that there will be a double layer forming here. And what is important at this point is, something called time scale, in this operation there it is important to look into the time scale.

What I mean by this is that, this double layer it has certain time to form there would be there would be something called a charge relaxation time. I mean you have to; when you apply the voltage there would be the charge, they have to align themselves next to the wall that requires certain amount of time and that time is different. If you expect an electron to align or if you expect an ion in an electrolyte to align or if you expect a polymer, big polymer molecule to align that time is different.

So, it requires certain time and at the same time you have something called an hydrodynamic timescale, where I mean the liquid is flowing. So, how much time it takes for the liquid to flow through this capillary? So, if these timescales are not matched properly, then you cannot expect there to; there, then equilibrium exist in that double layer and it is important to have that equilibrium existing in the double layer. Now, if you try to; if we try to look at the time the, this relaxation time that, I have been talking about if I look into it further.

If, what I see here is that this electric double layer I think by now, you all know that this electric double layer that has something called a Debye length. That is given by λ_D is equal to square root of $\epsilon R T$ divided by $2 F^2 Z^2 C$ infinity. And you know, what are these terms? Because we have already discussed in connection with

electroosmosis, F is here the faraday constant, which is nothing but the product of Avogadro number and elementary charge of an electron.

So, these two product of these two quantities, that is F and this Z prime is the ion valency, that is how we look at it and. So, in this case we have a λ_D , which is called a Debye length. Now, I expect the ion to travel over these λ_D , I mean if we; if this equilibration of double layer. If that has to happen, we ion has to travel over this distance. So, what we see on the other hand is that, a current can be written as σ into E which is nothing but this quantity where σ is given as sum over all components $\lambda_i C_i$.

What is σ here? σ is, let me point out here we have to note two terms which are important one is epsilon, which is permittivity that we have already studied in connection with electroosmosis. On the other hand this σ is basically the conductivity. This σ is conductivity of the electrolyte solution. So, this is permittivity and this σ is the conductivity. So, conductivity is reciprocal of the resistance. So, you can see, i is equal to σ into E and E is $\text{del of } \phi$.

So, this we understand and then σ ; I mean how? Where the conductivity is coming from? If we look at it, then this σ is basically the molar ion conductivity. That λ_i is equal to; λ_i is molar ion conductivity as far as this i -th, ion is concerned and this C_i is ionic concentration of species i . Now, if we are looking at a symmetric binary electrolyte with ion concentration of C . So, ion concentration of C , if you are looking at then, you write σ as equal to $\lambda_+ + \lambda_-$ into C .

So, this is for symmetric binary electrolyte with ion concentration C . Now, here we have another term that is λ_i . So, we understand σ , we understand C_i . Now, λ_i or in other here; we are talking about λ_+ and λ_- . This λ_i again intern, one writes λ_i as I said is molar ion conductivity that is equal to $F^2 Z_i^2 v_i$, where this v_i is, intern there related to the diffusivity of the ion v_i is equal to $D_i / R T$ so, this molar ion conductivity. So, that will have a faraday's constant and the valency that part is taken care of and then a velocity, which is arising from the diffusivity of that particular ion.

(Refer Slide Time: 13:47)

For fully dissociated symmetric binary electrolyte

$$D = D_+ = D_-$$

Time required for an ion to diffuse through Debye length.

$$t_D \sim \frac{\lambda_D^2}{D_i} = \frac{\epsilon}{\sigma}$$

$$\lambda_D = \sqrt{\frac{\epsilon R T}{2 F^2 \sum_i z_i^2 C_i}}$$

$$\sigma = f(\lambda_D) = f(F^2 \sum_i z_i^2 \cdot D_i)$$

$\approx 10^{-6}$ second, 10^5 to 10^7 ($10 \mu s - 0.1 \mu s$)

Viscous diffusion time scale $\sim \frac{R^2}{\mu/\rho}$

Now, for a fully symmetric, for fully dissociated symmetric binary electrolyte, if I write or if I go to the next page, for a fully symmetric for sorry for fully dissociated symmetric binary electrolyte D is equal to D plus is equal to D minus. We are talking about D iso, here, since we have brought in this λ_D plus and λ_D minus for symmetric binary electrolyte. So, by the same token, you will have D plus and D minus and both would be equal to D for a fully dissociated symmetric binary electrolyte. So, we given this information now, if somebody tries to find out, what is the time required? If somebody wants to find out, what is the time required for an ion to diffuse through Debye length?

So, that can be given by in that case, t_D let us say, that would be close to λ_D^2 divided by D_i . This D_i is the diffusivity of component of species i , of that; ion diffusivity of that, through that D by layer and λ_D is the length. I mean basically we are talking about a length scale and the velocity at which, that will go through that length scale and from that you are trying to find out; we are trying to find out, what would be the time? What would be the order of time required for an ion, to go through this distance? So, that is basically λ_D^2 by D_i and if, somebody works with these I mean; I have already we have talked about λ_D here, what is λ_D ?

λ_D is square root of this quantity that we have already; we have studied this earlier as well and this part is the σ . So, if somebody works with these and this σ

intern is related to the D . So, if somebody works now, we to find out, what would be the λD^2 by D in this case? You will find that λD^2 means, you had λD as square root of $\epsilon R T / 2 F^2 Z'^2 C_\infty$. So, this λD^2 means, this square root would be gone and on the other hand if you look at, what you have in σ ? You will find that σ contains, those other terms, there as well σ contains λ .

σ is a function of λ and then λ , this is the molar ion conductivity and that contains these $F^2 Z'^2$ etcetera and then you have D . So, if somebody works through this, we will find out that, this all these terms will cancel out. This $F^2 Z'^2$ and here, you have σ plus λ plus and λ minus for definition of σ . So, you will find that, this will cancel out and you will have this $t D$, left you are with $t D$ is equal, this quantity is equal to ϵ / σ . That means, this is the ratio of permittivity of the electrolyte solution divided by the conductivity of electrolyte solution.

Now, if somebody looks into what are the typical permittivity and typical conductivity of the materials that one works with, you will invariably find that, this is; this comes to about ten to the power of minus six second. I mean of course, this is of the order I mean ten to the power minus six means, it can be between ten to the power minus five to ten to the power minus seven. That means, this is basically ten micro second to point one micro second. So, this is the time required for equilibration of anion, if you through this Debye length.

Now, here you have another time scale see, what you are playing with? If you have to atomize this way; that means, you are having a liquid flow and at the same time you are developing some charges within this solution. So, if you expect that these two; expect that the; that these equilibration within the Debye length happens, much faster than I mean much before the liquid is flown out of it. So, then in that case this time scale has to be compared with, what you call? The viscous time scale, viscous or that complete term would be viscous diffusion time scale which typically referred as R^2 / μ by ρ where R is the capillary dimension. So, this is typically a viscous diffusion time scale or if somebody wants to work with a, what you call?

(Refer Slide Time: 22:15)

For fully dissociated symmetric binary electrolyte
 $D = D_+ = D_-$

Time required for an ion to diffuse through Debye length.

$$t_D \sim \frac{\lambda_D^2}{D_i} = \frac{\epsilon}{\sigma}$$

$$\lambda_D = \sqrt{\frac{\epsilon R T}{2 F^2 z_i^2 c_i}}$$

$$\sigma = f(\lambda) = f(F^2 z_i^2 \cdot D_i)$$

Permittivity of the electrolyte solⁿ
 Conductivity of electrolyte solⁿ

$\approx \frac{10^{-6} \text{ second}}{10^{-5} \text{ to } 10^{-7}} \approx 10 \mu\text{s} - 0.1 \mu\text{s}$

Viscous diffusion time scale $\sim \frac{R^2}{\mu/\rho} \approx 10^4 \mu\text{s}$ (for $R = 0.1 \text{ mm}$)

Hydrodynamic time scale $\sim \frac{L}{v} \sim \frac{R L}{Q}$

$t_D \ll t_{\text{viscous diffusion}}$

- double layer exists in equilibrium
- sufficient residence time to charge double layer when jet moves out of capillary

Hydrodynamic time scale, one can write it this way as well. Hydrodynamic time scale which would be equal to L/v is the length divided by v , which is the velocity, which is something like $R^2 L / Q$. Q is the flow rate, R is the capillary dimension. So, that also possible, this hydrodynamic time scale and what you have to ensure is that, this viscous diffusion time scale, has to be greater than this. That means, this should be shorter, this t_D the time required for these ions to relax. Within this Debye length that has to be shorter by, at least couple of orders of magnitude because then in; that case, the flow will happen and the equilibration will happen instantaneously.

So, you do not have any issue with it, has not been equilibrated ions have not formed properly, but the, you are forcing the flow to go out I mean it is not that way. So, then you can use the full potential of this mechanism. So, these time scales are extremely important I mean, if I try to articulate this further, if this time scale. If, this t_D is much less than this viscous diffusion, if this is much less, what you can conclude from here is that, the double layer exists in equilibrium.

In other words the; so, what you conclude here is that, sufficient time sufficient I would say, residence time to charge double layer, when jet moves out of capillary. So, this is ensured these issues are addressed, if you have this time scale. This is less than the viscous diffusion, fortunately this viscous diffusion typically. People have for say, for R is equal to say, if you are looking at 0.1 millimeter, then these will come to the, these time scale. This viscous diffusion time scale typically will for conventional, for common materials that we handle this will come to about ten to the power four microsecond.

So, we were looking at 10 to 0.1 or may be 1 micro second. Ten to the power minus six second for that time required for an ion to diffuse whereas, the viscous time scale is much higher. So, when this is much higher so; that means, the equilibration of double layer will have I mean this time scale appears almost instantaneous compared to the flow of this liquid. The viscous as far as the, viscous effect is concerned and this is extremely important if these two time scales are not adjusted then you cannot leverage this idea.

So, as a matter of fact, I mean since we are into this relaxation time scale I mean, we need to understand this time scale or rather we will address this time scale, once again at a later time. Because these we have; so far, assumed that this is a DC electric field. Now, if one has this AC if the; if there is AC field. If, there is another time scale involved, within the time scale the voltage changes. So, then that time scale also will become important for this mechanism. So, this AC time scale, an AC time scale, what defines AC time scale is a frequency? How many hertz?

So, what frequency you have? This the that is, that becomes important and then what we will try to do at this, at that time is, we are trying to find out, what is the R c time scale? Because see, this Debye, the time required for a diffuse to Debye length. So, this double layer, that can be expressed as A Circuit like, a resistance and capacitance for example, a resistance and capacitance they are acting parallel. Because this electric double layer that has its own resistance, arising since you have seen, there is a sigma involved and at the same time there would be some amount of capacitance as well.


So, you bring in that resistance capacitance type model I mean instead of looking at the double layer, you simulate that with a R and c and find out, what is the time scale for that R C circuit? And that is referred as a R C time scale so, this R C time scale will be compared I mean, if you had an AC field operating then the that has its own time scale and how that? Where that stands with reference to the time scale of this diffusion of this, ions? So, that is another aspect, another issue that will come up.

So, this... So, you... So, basically if we go back to I mean, where we started? We started this, we started on these, formation of we are trying to find a mechanism by which the electrospray will happen and we found that, there is an, there is a issue of this time scale we I mean that is appreciated.

(Refer Slide Time: 28:40)

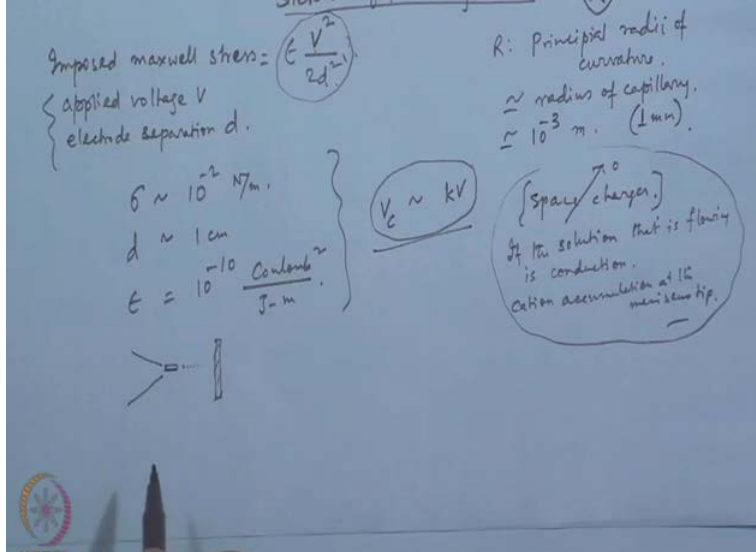
...Contd. from previous slide

- Critical voltage
- Subsequent break-up of jet
 - Rayleigh Capillary instability
 - Coulombic instability
- Evaporation within drop => decrease in drop size=> increase in charge density=> coulombic fission
- Spray mode (decided by the applied voltage and liquid flow rate through the capillary)
 - Pulsating:
 - Continuous



If, there could be and. So, if I look at the slide, there could be the DC electric field or an AC electric field either way, it is possible. For DC electric field whatever, we talked about is fine. For AC electric field, there is another time scale coming in because of the frequency. Now, what we must understand here is there exists something called A Critical voltage I mean of course, we are in this DC regime. I mean we first, we are talking about the DC field, not the AC field we will look into some of the aspects of these later.

(Refer Slide Time: 29:15)



Imposed maxwell stress = $\epsilon \frac{V^2}{2d^2}$


applied voltage V
electrode separation d.

$\sigma \sim 10^{-2} \text{ N/m.}$
 $d \sim 1 \text{ cm}$
 $\epsilon = 10^{-10} \frac{\text{Coulomb}^2}{\text{J-m}}$

$V_c \sim \text{kV}$

R: Principal radii of curvatures.
~ radius of capillary.
~ 10^{-3} m. (1mm) .

[Spray charges.]
If the solution that is flowing is conducting.
Action accumulation at the meniscus tip.



What we have here, is something called A Critical voltage. The critical voltage is, if I try to derive it here, the critical voltage see this, the liquid. If we look at the liquid meniscus, that protrudes from the orifice. If you assume that, this is basically of the shape of a spherical cap. If this is considered as spherical cap, then this is equal to, then this will be σ by R . So, then the stress arising from surface tension that is, equal to σ by R . Where R is basically you know principle. I mean you understand what this is? This is of the same order as radius of the capillary.

We are doing just an order analysis here, I mean at what time well we need ten to the power five volt or ten to the power eight volt. That kind of analysis we are doing, it is not a rigorous analysis either and. So, this is of the order of, if we look; try to look at the order, this would be about ten to the power minus three meter. That means, we are talking about say 1 millimeter and the if, somebody looks into the imposed Maxwell stress, then for applied voltage V and electrode separation D , Maxwell stress imposed. Maxwell stress would be ϵ into V square by D square.

So, if this if, the stress arising from surface tension is greater than the Maxwell stress, then that droplet will not come out. So, this Maxwell stress this has to be greater than this quantity has to be greater than this σ by R , then only the I mean we are talking about the order by order of magnitude. This has to be greater than this then, only you can expect the droplet to come out. So, if that is. So, then you can come up with a, what would be the critical voltage? I mean what is the minimum voltage that you should have?

So, if we equate this to and if, you remove two I mean if you. So, what if we try to find out the order? You can take σ as ten to the power minus two Newton per meter. d is the distance between the two can be taken as say, one centimeter that means, what distance we are talking about? We are talking about distance between the two electrodes, we are talking this distance. So, this distance is let us say, 1 centimeter and then ϵ is commonly taken as ten to the power minus ten coulomb square by joule meter. So, in that case, you will find this critical voltage. This critical voltage will come to the order of kilovolt.

Say it can, this is of the order of kilovolt, not simple volt or not megavolt. It will be of the order of kilovolt. So, that is something. So, there exists such critical voltage. So, you. So, if somebody wants to know for conventional applications, for conventional the

materials that we have, if we want to create droplet by this mechanism. What kind of voltage we have to create between the electrodes? And this is something, which we have there. So, there exists A Concept of critical voltage.

Of course this is, but just an order analysis one important thing you must understand I mean if you do such, if you go by this kind of the route. What we are neglecting here is the space charges, because there would be if, the solution itself. If the solution that is flowing if the... So, this part is, this is neglected. So, if the solution that is flowing is conducting if the solution, that is flowing is conducting, that means there would be a charge accumulation at the meniscus tip.

So, there would be some difference, there would be some change to this critical voltage. So, for A Conducting solution there if the solution that is flowing is, it has its own space charge. So, that has to be. So, that will effect this analysis. Now, once the jet comes out, once the liquid meniscus come out here, we have another issue here. Subsequent breakup of jet, which is basically there are two ways, this jet can breakup. One is by Rayleigh capillary instability and the other is coulombic instability. Let us try to understand, do you have? This is the cone and the, this is from which the liquid is coming and this is the ground plate and a jet comes out of this. Now, this jet subsequently breakup into small droplets.

(Refer Slide Time: 36:49)

Rayleigh Instability.
 charging of jet is not ~~substantive~~ excessive.
~~the charge on~~ drop diameter $\sim 1.9 \left\{ \text{Jet diameter} \right\}$.
 drops are monodisperse.

Coulombic instability
 for excessively charged jet \Rightarrow jet goes through bending/whipping.
 drops are polydisperse
 drops are significantly smaller in size.

Diagram showing a jet being deflected by a cone. The cone angle is labeled as θ . The jet diameter is labeled as d .

Cone angle approaches to 49.3° .
 for perfectly conducting drop
 the liquid to gas permittivity ratio $\rightarrow \infty$.
 Cone angle = f (liquid to gas permittivity ratio)

© CET I.I.T. KGP

NPTEL

This breakup is possible by two mechanisms one I say one we see here is that, this Rayleigh capillary instability other is the coulombic instability. The first type of instability, we have seen already. We have studied herein the previous occasion this instability will come into play, when the charging of jet is not not not substantial, not not excessive. So, this is the type of breakup of jet. We have talked about it is exactly the same one. Basically you assume that the charge on the surface of that jet, does not contribute significantly to this instability. Here, in this type of instability the drop diameter is close to 1.9 times the jet diameter and drops are monodisperse.

So, basically that the Rayleigh instability that was, that is considered for breaking of a jet, that has been modified for the presence of charge and that the that. So, people have done analysis in that line and that type of breaking will happen, when the charging of jet is not very high. So, in that case, there is monodispersed drops are obtained by drop diameter is close to 1.9 times jet diameter. You know, we have already talked about these Rayleigh instability. How the jet diameter? How the drop diameter relates to jet diameter? There would be a perturbation, that perturbation grows and there is a from breaking of jet into smaller droplets. So, this we have done similar analysis, only the presence of charge has to be considered in that analysis nothing else.

The other type of breaking is referred as coulombic instability this is for excessively charged jet. So, this is, this. When the jets are, jet is excessively charged that time the such coulombic instability will come in. And these in this case the drops are polydispersed and a jet undergoes basically. In this case jet goes through bending or whipping jet is, jet goes through these processes and a drops are polydisperse. However, here, the drops are significantly smaller in size another thing you must appreciate here is that evaporation within drop. See, when the jet is, jet has come out of the capillary tip there will be evaporation within the drop and therefore, there will be decrease in drop size.

If, there is evaporation then automatically there would be loss, evaporative loss and the drop size decreases. But charge does not decrease. So, there would be increase in charge density and as a result, if there is increase in charge density, there would be more possibility of this coulombic fission. So, once the drop forms, it could be that you the drop has formed by this the first one, first type of instability. But with evaporation there is decrease in drop and then there would be increase in charge density. So, there would be


further coulombic fission; that means, further breaking of drops. So, that. So, this evaporation is because here. So, that you have to make note of.

Now, when it comes to the spray mode, the spray of this there are two ways. It can be I mean two distinct features can be observed here, one is called a pulsating mode another is continuous mode. In pulsating, this is decided by the applied voltage and liquid flow rate through the capillary. So, these are primarily the two parameters, two variables that control the spray mode and. So, you have the two methods one is pulsating another is continuous.

(Refer Slide Time: 42:41)

...Contd. from previous slide

- Pulsating spray mode: meniscus cone from the orifice oscillates during ejection of droplet from the tip
- Continuous spray mode: continuous jet disintegrates outside the capillary into droplets.
- Concerns in using DC electric field
 - Gas bubbles from electrolytic reactions to balance the charge induced by the applied voltage
 - Joule heating
 - Generation of other chemicals due to Faradaic reactions at the electrode



Now, what you have in pulsating spray mode is that, the meniscus cone from the orifice oscillates during ejection of droplet from the tip. So, and on the other hand the continuous in the case of continuous spray mode, continuous jet disintegrates outside the capillary into droplets. So, in one occasion we see, that if this is the cone, there is a meniscus coming out from the orifice and then this is released and then this one is meniscus goes back. So, meniscus cone from the orifice oscillates, meniscus cone comes out forms a drop and again it goes back again another drop comes in. So, these drops are formed directly from this meniscus, that is a possibility, the other possibility could be A Continuous jet coming out and then that jet disintegrates into droplets.

So, these; this one is referred as in this jar gun that they are referred as pulsating spray mode another is continuous spray mode. So, there is a difference between the two. In this

connection, you must understand one thing is that, there is a proposition here, is that. If, this is the capillary I mean we are talking about this spray mode we are talking about this pulsating mode. We have that means, you are forming A Cone here, Taylor has shown that, if this is the capillary, if say radius capital R and then it forms A Cone like this. So, this is the end of the capillary and then it forms A Cone like this and this being the cone angle.


So (O) had this, there is scientific literature, where it is shown that, this cone angle approaches this cone angle. So, we are talking about this for this pulsating mode, this cone angle approaches to 49.3 degree for perfectly conducting drop and the liquid to gas permittivity ratio into infinity. So, for these cases this was shown that, this cone angle will take a value of this. So, in fact, cone angle can be predicted from the liquid to gas permittivity ratio. So, so from. So, this there would be. So, if this is the end of the capillary tip, there would be forming A Cone and this cone angle will be a function of liquid to gas permittivity ratio. And these at the limit, this is 49.3 degree this has been shown by the researcher already.

So, this is what we have as far as. So, these are the two modes, that we talked about and this is all, we have as far as the two rather that, the use of DC electric field is concerned in forming the droplets. Now, there are certain concerns in using DC electric field because next what we would be talking about is the AC electrospray. And there are certain concerns in using this DC electric field. This, concerns are I can see number one is gas bubbles from electrolytic reactions to balance the charge induced by the applied voltage. So, that is one concern, which is already I mean we understand this, that is not that is understandable. Second thing is joule heating, because this is a DC field and generation of other chemicals due to Faradaic reactions at the electrode.

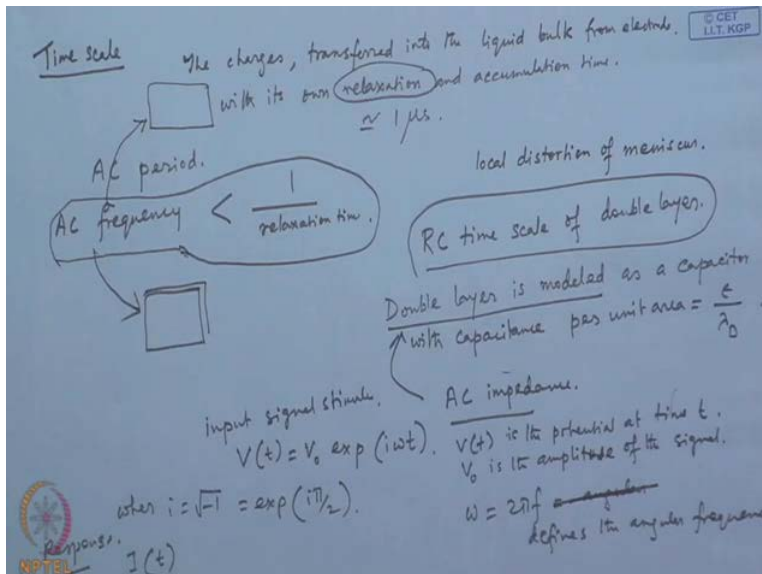
(Refer Slide Time: 48:27)

...Contd. from previous slide

- AC electropray
 - Importance of AC frequency
 - Superimposition of AC potential on DC potential
- DC electrospinning
 - Jet solidifies downstream due to solvent evaporation
 - Random beading / coiling / bending / winding / spiralling / looping / due to axi-symmetric or azimuthal Rayleigh instabilities
 - Ring electrodes around jet can control instability



(Refer Slide Time: 49:08)



Time scale

the charges, transferred into the liquid bulk from electrode, with its own relaxation and accumulation time. $\approx 1 \mu s$.

AC period.

AC frequency $< \frac{1}{\text{relaxation time}}$.

local distortion of meniscus.

RC time scale of double layer.

Double layer is modeled as a capacitor with capacitance per unit area = $\frac{\epsilon}{\lambda_D}$.

input signal stimulus.

AC impedance.


$V(t) = V_0 \exp(i\omega t)$. $V(t)$ is the potential at time t . V_0 is the amplitude of the signal.

$\omega = 2\pi f$ defines the angular frequency.

where $i = \sqrt{-1} = \exp(i\pi/2)$.

Response, $I(t)$

© CET IIT KGP



So, these are some of the concerns of DC electric field for which there was an interest in using the AC field. Now, if we look at this AC electrospay, there are a couple of main issues, that I would be touching upon because otherwise this method is same and. You have this relaxation time and then there would be critical voltage and all everything would remain same. Only thing added on top here is the importance of AC frequency. So, importance of AC frequency, what we meant here is, that there exists another time scale.

So, that is this time scale that is another time scale, which is coming which is from the electric field itself. I mean if, I try to revisit that issue once again, the charges transferred into the liquid bulk from of course, from electrode. The charges transferred into the liquid bulk from electrode with its own relaxation and accumulation time. We showed that, this is of the order of 1 microsecond. Now, here you are imposing another AC period; that means, you are applying the voltage for certain time and for other times you are not applying I mean, you are going the other way.

So, if this AC frequency is less than 1 divided by this relaxation time, we are talking about. So, if the AC frequency is less, if this AC frequency is less see there is an issue, it could be that we are looking at a local distortion of meniscus. That means, there is I mean you are applying a voltage, but you are moving away from that. So, fast that simply the meniscus has distorted locally, but it has not gone to the culmination within that time frame. Culmination means forming of a complete drop and going out.

So, you have to ensure that, this voltage is applied for sufficient time. So, that the, you are not simply talking about the local distortion of the meniscus. So, this AC frequency culmination is linked with both these aspects. One is towards this AC frequency has to abide by the demand from that charge relaxation point of view and this AC frequency also has to be in tune with. What is the viscous timescale? Because the it is, if it is just a local distortion of meniscus that will not be sufficient.

So, what we would be looking at this point probably is something called the R C timescale of double layer that is something, which we have to look into. Now, if we try to understand, what is the R C timescale of double layer? Then what we have to do here, is we need to understand, how this double layer works here? Typically double layer is modeled as a capacitor with capacitance per unit area is equal to epsilon divided by lambda D, epsilon you know is permittivity lambda D is the thickness of this double layer.

Now, this double layer; so in that case the double layer I mean the ideally, I mean what they? What the way it is looked into is, if you consider this as an AC impedance that is the ideal way of handling double layer. Because in that case, you will; if you have an input signal something like this where I is equal to this quantity. So, if you give an input signal, where V naught is the amplitude of the signal, and V t is the potential at time t and omega here is equal to 2 pi F is the angular. I mean basically it is associated with the

angular frequency, in that case, what we will do? In fact, we will continue this analysis what we will do is, we will find out the corresponding response signal which is called a response signal.

So, $V(t)$ is an input signal stimulus and this is the response signal, we will calculate and then we will calculate, what is called the complex impedance. So, I mean this is a very standard exercise for electric circuits, what we are trying to find? We will try to find out, what would be the RC timescale of a double layer? And that has to relate to the AC frequency. I will try to at least touch upon the various issues in the next lecture and then, we will continue understanding this electrospray process and there is another very interesting phenomenon, which is I mean which is very close to this kind of, this use of electrode and producing flow out of it. So that, we will touch upon in the next lecture thank you very much.