

Mass Transfer Operation-2
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Lecture 9

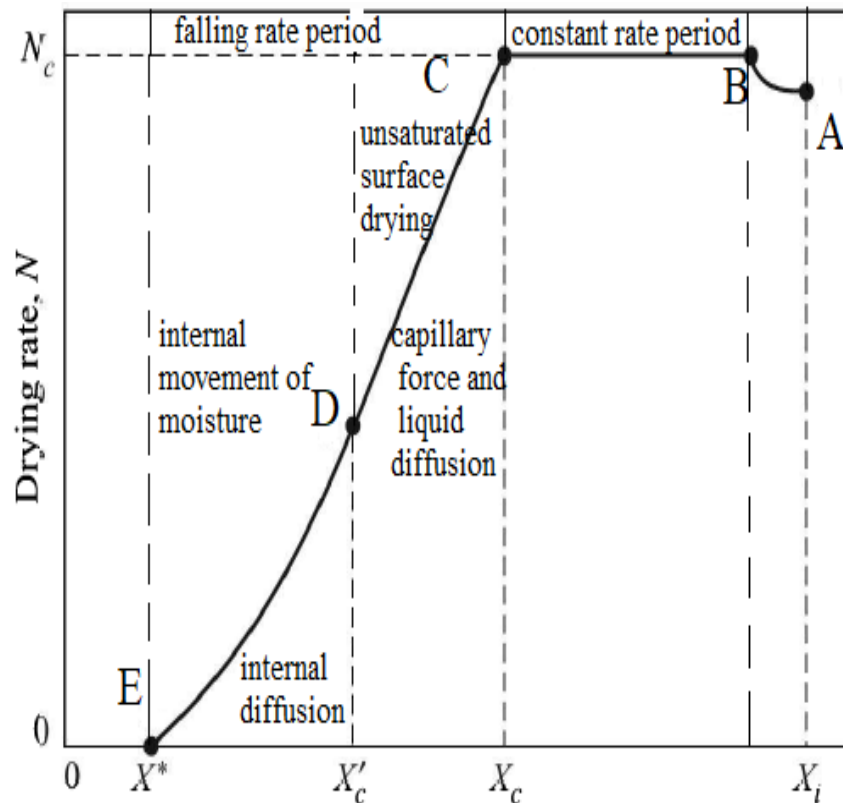
Drying Time Calculation from Drying Time Curve

Welcome back to mass transfer operation-2 course. In the last class we discussed the details about the continuous dryer and in this class we will be discussing the drying time calculation from drying time curve.

Movement of moisture within solid:

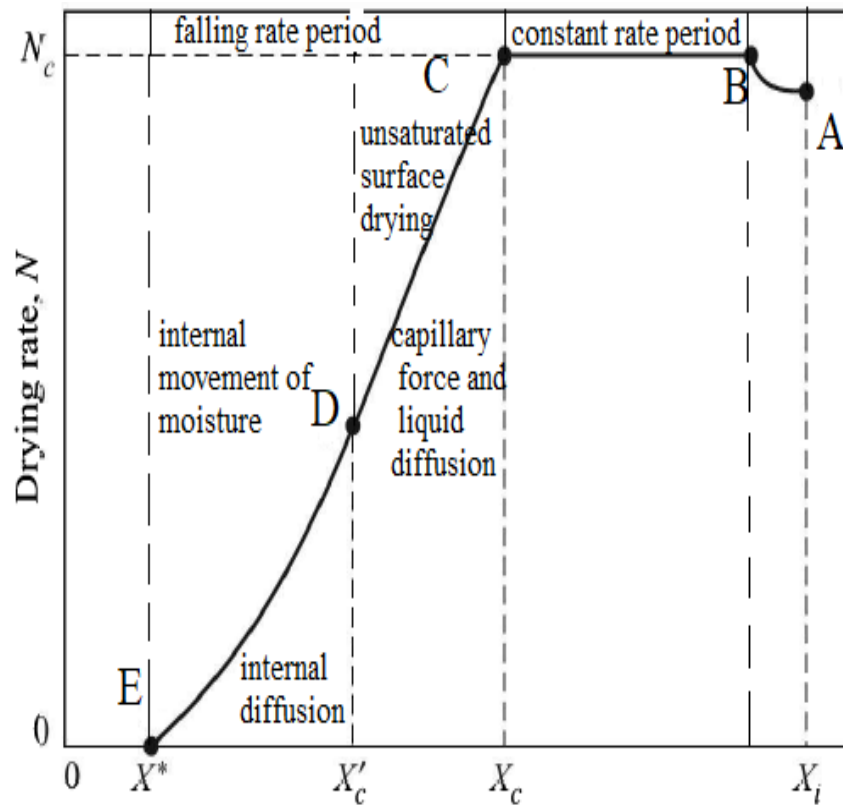
- Liquid Diffusion: Due to concentration gradient between depths and surface falling rate period. Diagram.
- Capillary movement: Unbound moisture in granular and porous solids moves through the capillaries → surface tension.
- Constant drying rate period: As drying proceeds, at first, moisture moves by capillary to the surface rapidly enough to maintain a uniformly wetted surface and drying rate is constant.
- Vapour Diffusion: If heat is supplied to one surface of a solid while drying proceeds from another, moisture may be evaporated beneath the surface and diffuse outward as vapour.
- $N = - \frac{S_s}{A} \frac{dX}{dt}$
- Let X_i is initial moisture content of solid,
- X_f is final moisture content

Calculation of drying time: Constant drying conditions



Movement of moisture within solid:

- **Liquid Diffusion:** Due to concentration gradient between depths and surface falling rate period.
- **Capillary movement:** Unbound moisture in granular and porous solids moves through the capillaries \rightarrow surface tension.
- **Constant drying rate period:** As drying proceeds, at first, moisture moves by capillary to the surface rapidly enough to maintain a uniformly wetted surface and drying rate is constant.
- **Vapour Diffusion:** If heat is supplied to one surface of a solid while drying proceeds from another, moisture may be evaporated beneath the surface and diffuse outward as vapour.



$$\therefore \int_0^t dt = - \int_{X_i}^{X_f} \frac{S_S}{A} \frac{dX}{N} \rightarrow t = - \frac{S_S}{A} \int_{X_i}^{X_f} \frac{dX}{N} = \frac{S_S}{A} \int_{X_f}^{X_i} \frac{dX}{N}$$

$$t = \frac{S_S}{A} \int_{X_f}^{X_c} \frac{dX}{N} + \frac{S_S}{A} \int_{X_c}^{X_i} \frac{dX}{N}$$

- * Drying rate remains constant at N_c till X_c from X_i .

1. Constant rate period:

$$t_C = \frac{S_S}{A} \int_{X_c}^{X_i} \frac{dX}{N_c} = \frac{S_S(X_i - X_c)}{AN_c}$$

2. First falling rate period:

- integral can be evaluated graphically ($\frac{1}{N}$ vs X)

Special case 1:

- $N = mX_c + b$
- $N_c = mX_c + b$
- $N_f = mX_f + b$

- $m = \frac{N_C - N_f}{X_C - X_f}$
- $t_f = \frac{S_S}{A} \int_{X_f}^{X_C} \frac{dX}{N}$
- $= \frac{S_S}{A} \int_{X_f}^{X_C} \frac{dX}{mX + b} = \frac{S_S}{Am} \ln\left(\frac{mX_C + b}{mX_f + b}\right)$
- $= \frac{S_S}{A} \frac{X_C - X_f}{N_C - N_f} \ln\left(\frac{N_C}{N_f}\right)$
- Drying rate at $N = 0$ at equilibrium moisture, $X = X^*$
- $\therefore 0 = m X^* + b \rightarrow b = -m X^*$
- $\therefore \frac{N_C}{N_f} = \frac{mX_C - mX^*}{mX_f - mX^*} = \frac{X_C - X^*}{X_f - X^*}$
- $t_f = \frac{S_S}{A} \frac{X_C - X_f}{N_C - N_f} \ln\left(\frac{X_C - X^*}{X_f - X^*}\right)$
- $t_f = \frac{S_S}{A} \frac{X_C - X^*}{N_C} \ln\left(\frac{X_C - X^*}{X_f - X^*}\right)$
- $t = t_C + t_f = \frac{S_S(X_i - X_C)}{AN_C} + \frac{S_S}{A} \frac{(X_C - X^*)}{N_C} \ln\left(\frac{X_C - X^*}{X_f - X^*}\right)$
- **Special case 2:**
- $N = bX^2 \rightarrow N_C = bX_C^2 \rightarrow b = \frac{N_C}{X_C^2}$
- $t_f = \frac{S_S}{A} \int_{X_f}^{X_C} \frac{dX}{bX^2} = \frac{S_S}{Ab} \int_{X_f}^{X_C} X^{-2} dX = \frac{S_S X_C^2}{AN_C} \left[\frac{1}{X_f} - \frac{1}{X_C} \right]$
- $t = t_C + t_f = \frac{S_S(X_i - X_C)}{AN_C} + \frac{S_S X_C^2}{AN_C} \left[\frac{1}{X_f} - \frac{1}{X_C} \right]$

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Calculation of drying time : Constant drying conditions

$$N - \frac{S_S}{A} \frac{dX}{dt}$$

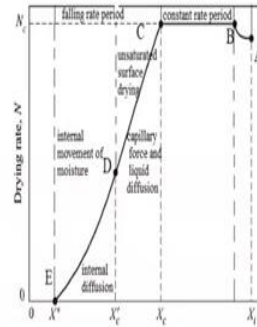
Let X_i is initial moisture content of solid,

X_f is final moisture content

$$\therefore \int_0^t dt = - \int_{X_i}^{X_f} \frac{S_S}{A} \frac{dX}{N} \rightarrow t = - \frac{S_S}{A} \int_{X_i}^{X_f} \frac{dX}{N} = \frac{S_S}{A} \int_{X_f}^{X_i} \frac{dX}{N}$$

$$t = \frac{S_S}{A} \int_{X_f}^{X_c} \frac{dX}{N} + \frac{S_S}{A} \int_{X_c}^{X_i} \frac{dX}{N}$$

* Drying rate remains constant at N_c till X_c from X_i .



And calculation of drying time like a constant drying condition we have already discussed this one still we are recapturing those like this $\frac{S_S}{A}$ is equal to say minus S by A into dX by dt where x is the initial moisture content and say we can say this one X to p be the final moisture content. Then whenever integrating this equation from 0 to t with respect to minus S by A dX by N will be getting this t in terms of we can say S by A and this one in terms of this N , and drying time remains constant at N_c till X_c this one from X_i like this from X_i means they say we will start from here and X_c is we can say this one when constant drying rate stops.

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Calculation of drying time : Constant drying conditions

1. Constant rate period:

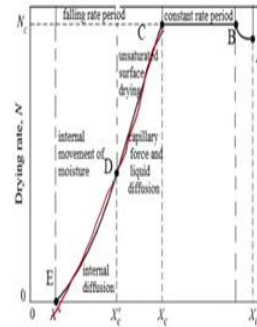
$$t_C = \frac{S_S}{A} \int_{X_C}^{X_i} \frac{dX}{N_C} = \frac{S_S(X_i - X_C)}{AN_C}$$

2. First falling rate period:

integral can be evaluated graphically ($\frac{1}{N}$ vs X)

Special case 1:

$$\begin{aligned} N - mX_C + b &= t_f - \frac{S_S}{A} \int_{X_f}^{X_C} \frac{dX}{N} \\ N_C - mX_C + b &= -\frac{S_S}{A} \int_{X_f}^{X_C} \frac{dX}{mX + b} = \frac{S_S}{Am} \ln\left(\frac{mX_C + b}{mX_f + b}\right) \\ N_f - mX_f + b &= \\ m \frac{N_C - N_f}{X_C - X_f} &= \frac{S_S}{A} \frac{X_C - X_f}{N_C - N_f} \ln\left(\frac{N_C}{N_f}\right) \end{aligned}$$



Now we will be discussing about the constant rate period time. The t_C is equal to S_S by A N_C into X_i minus X_C and for falling rate period we can say this one there may be two falling rate like First falling rate period than second falling rate period like this. Say we can say this one they may be first falling rate period maybe we can say drying rate is proportional to moisture content then this first falling rate period this T_f will be like this S_S by A into X_C minus X_f by N_C minus N_f into $\ln N_C$ by N_f . So that will be we can say this one time for first falling rate period.

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Drying rate at $N = 0$ at equilibrium moisture, $X - X^*$

$$\therefore 0 - mX^* + b = b - mX^*$$

$$\therefore \frac{N_C}{N_f} = \frac{mX_C - mX^*}{mX_f - mX^*} = \frac{X_C - X^*}{X_f - X^*}$$

$$t_f = \frac{S_S}{A} \frac{X_C - X_f}{N_C - N_f} \ln\left(\frac{X_C - X^*}{X_f - X^*}\right)$$

$$t_f = \frac{S_S}{A} \frac{X_C - X^*}{N_C} \ln\left(\frac{X_C - X^*}{X_f - X^*}\right)$$

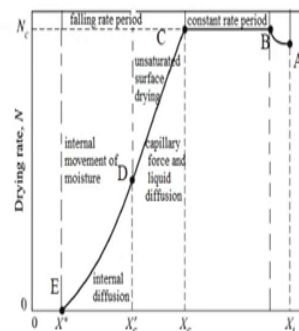
$$t = t_C + t_f = \frac{S_S(X_i - X_C)}{AN_C} + \frac{S_S}{A} \frac{(X_C - X^*)}{N_C} \ln\left(\frac{X_C - X^*}{X_f - X^*}\right)$$

Special case 2:

$$N - \beta X^2 \rightarrow N_C = \beta X_C^2 \rightarrow \beta = \frac{N_C}{X_C^2}$$

$$t_f - \frac{S_S}{A} \int_{X_f}^{X_C} \frac{dX}{\beta X^2} = \frac{S_S}{A\beta} \int_{X_f}^{X_C} X^{-2} dX = \frac{S_S X_C^2}{AN_C} \left[\frac{1}{X_f} - \frac{1}{X_C} \right]$$

$$t = t_C + t_f = \frac{S_S(X_i - X_C)}{AN_C} + \frac{S_S X_C^2}{AN_C} \left[\frac{1}{X_f} - \frac{1}{X_C} \right]$$



If the falling rate period is square of this we can say this one moisture content then it will be like this say t_f will be this S_S by A into N_C into X_C square into 1 by X_f minus into 1 by X_C .

Then we can say that total time for this drawing will be like t constant rate period and t final rate period that is we can say this one t_f so that will be S_s by AN_c into X_i minus X_c plus S_s by AN_c into X_c square into 1 by X_f minus 1 by X_c . So these are the, we can say this are the regular cases.

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Problem 4

A wet solid having 32% moisture (dry basis) is to be dried on a tray dryer to a final moisture of 1%. The solid loading is 30 kg dry solid per m^2 tray area. There are two critical moisture values $X_{c1} = 0.183$, and $X_{c2} = 0.097$. A laboratory test gives a drying rate of 4 kg/m^2 h in the constant rate period. In the first falling rate period, the drying flux is linear in the moisture content; in the second falling rate, the drying flux varies as the square of the moisture content. The equilibrium moisture is negligible. Calculate the drying time if the drying conditions are the same as in the laboratory test. Mention any assumption made.

Now we will be discussing one case where the 2 drying rate this one falling rate periods are there with first falling rate will be proportional to X and second falling rate will be proportional to X Square. So the problem is like that a wet solid having 32 percent moisture that is on dry basis is to be dried on tray dryer to a final moisture of 1 percent. So this is the requirement of the system like final moisture content will be 1 percent.

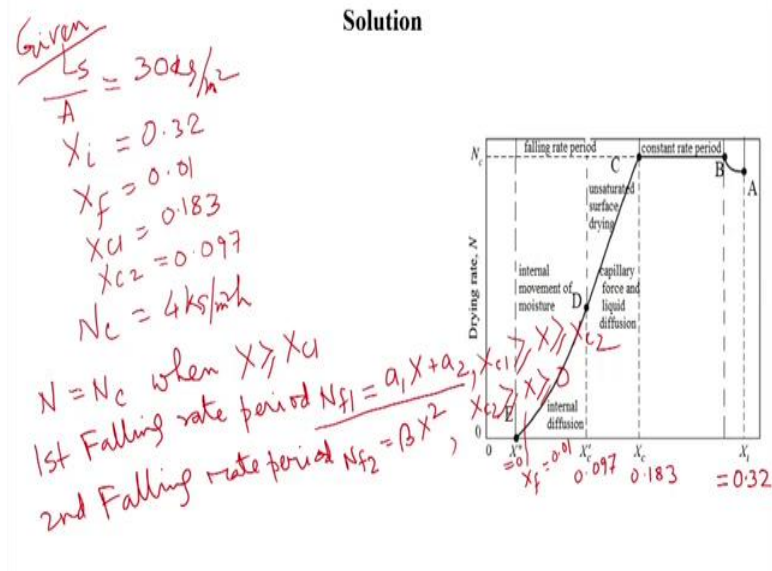
The solid loading is 30 kg dry solid per meter square area that is we can say S_s by A is equal to 30 kg per meter square. There are two critical moisture values that is X_{c1} that is equals to 0.183 and X_{c2} is equal to 0.097. A laboratory test gives a drying rate of 4 kg per meter square hour. In the first falling rate period the drying rate is linear in the moisture content. In the second falling rate the drying Flux varies as the square of the moisture.

The equilibrium moisture is negligible. Means we can say this one X equal to 0. We need to calculate the drying time if the drying conditions are the same as in the laboratory test so we need to mention any assumption if we have made.

So we will be doing this one step by step calculation but here we will be doing one thing that say the falling rate period are here two different falling rate period. Like the fast falling rate

period will be following this linearity and second falling rate period will be proportional to X Square.

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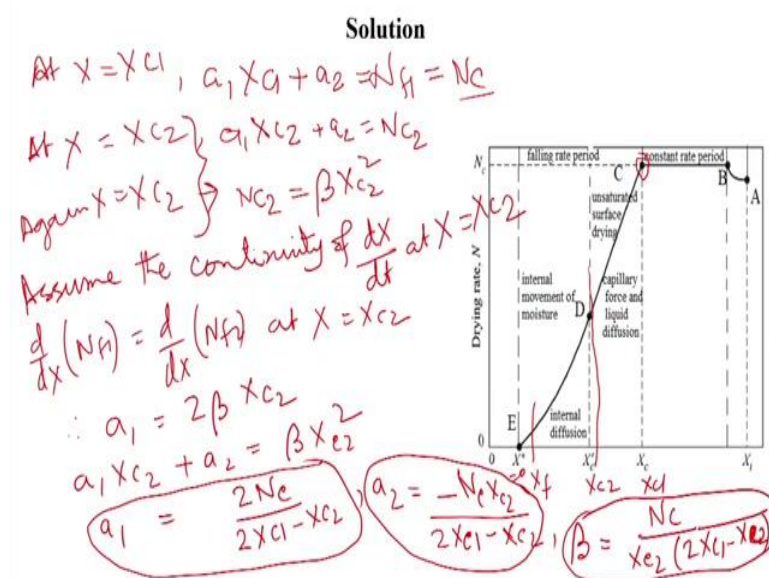
Let us start with the solving the problem like this we can say, we will be writing this one this L_s by A that is we have this one 30 kg per meter square and X_i that is given as 0.32 and X_f that is given as 0.01 and X_{c1} that is 0.183. These are all given and X_{c2} is equal to 0.097 and N_c that is also supplied that is 4 kg per meter square hour. So these are all given in the problem itself.

Now we need to get the co-relations having all this X values along with the flux values or we can say the drying rate values like this this N is equals to we can say this one N_c when we can say X is greater than equal to X_{c1} and first falling rate period say this one first falling rate period. Say we have this N_{f1} that is equals to you can say this is linearly proportional to the X values so we can say is equals to say a_1 into X plus a_2 .

So this X_1 that is given as 0.32. So X_{c1} that is given as a 0.183. So X_{c2} is given as 0.097. So somewhere this X_f actually is given as 0.01 and this X^* is equals to 0. This is equals to 0 it is given in the problem. For second falling, falling rate period N_{f2} is equals to say you can say beta into X square.

So for first falling rate period it is mentioned that for this case say we have say X_{c1} that is greater than equal to X greater than equal to X_{c2} . For second rate falling period we have this X_{c2} greater than equal to X greater than equal to 0. So these are the conditions these are given. So we will be mentioning the conditions like this.

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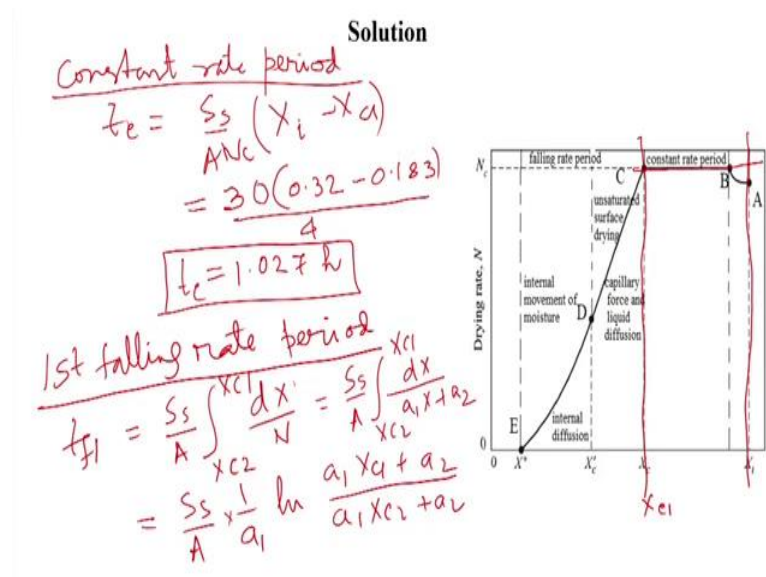


At X is equals to X_{c1} we have this $a_1 X_{c1}$ plus a_2 that is equals to N_{f1} and that is also equals to N_c because that is we can say this one here it is it is the this one at the beginning so we can say it is nothing but N_c . And X is equals to we can say at X is equals to X_{c2} where we can say this one X_{c2} and this one as X_{c2} a_1 into X_{c2} plus a_2 is equals to N_{c2} . Again at X is equals to this in this again at X is equals to X_{c2} here also we can say we have this N_{c2} that is square of this moisture content then we can say this is equal to β into X_{c2} square.

So we have this one and we need to assume this one that continuity this one we assume, we assume the continuity of dx/dt . So at X is equals to X_{c2} like this some X_{c2} that is X_{c2} that is X_{c1} that is X_i then somewhere it is X_f that is X_{star} equals to 0. So then we can say this one dx into N_{f1} is equals to $d dx$ of N_{f2} at X is equals to X_{c2} .

So from there we will be getting like this a_1 is equals to 2 into β into X_{c2} and a_1 into X_{c2} from the above equation plus a_2 is equals to you can say this one β into X_{c2} square or a_1 will be equals to $2N_c$ that from the above equation we will be just by manipulation we will be getting $2X_{c1}$ minus X_{c2} and a_2 will be equal to minus N_c into X_{c2} divided $2X_{c1}$ minus X_{c2} and β will be equal to N_c divided by X_{c2} into $2X_{c1}$ minus X_{c2} . So now we have this a_1 , a_2 and β . So we have all this three values.

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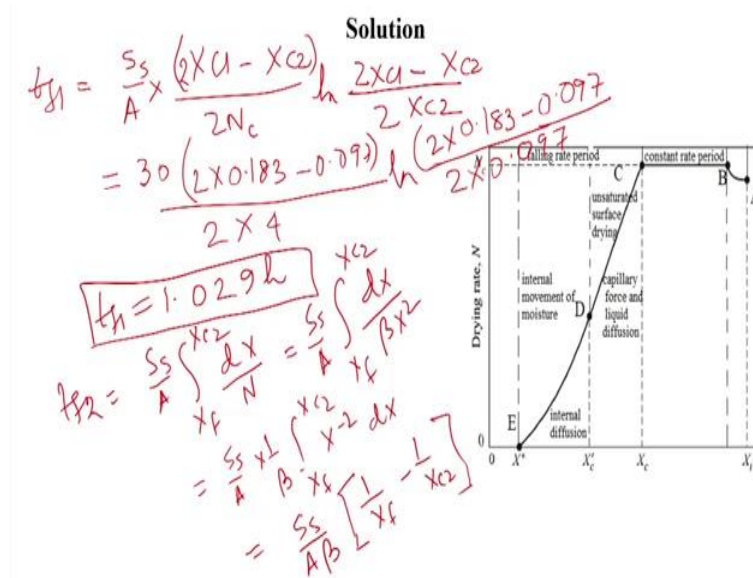


Now we need to calculate the drying time for different period like this we will start with constant rate period. So drying time calculation for constant rate period that is very easy to calculate constant rate period, first for this rate period we can say from X_i to X_{c1} So X_{c1} from X_i to X_{c1} for drying time calculation like this t_c will be, t_c is equals to say S_s by $A N_c$ into X_i minus X_{c1} .

So that will be like this S_s by A we know this one this is 30 and X_i is equals to 0.32 and X_{c1} is equals to 0.183 divided by N_c is equals to 4 so it will be becoming as in terms of hour that is 1.027 hour. So this is we have got, obtained as t_c is equals to 1.027 hour. Now we have 2 falling rate periods, so for first falling rate period t_{f1} that is equals to S_s by A into integration X_{c2} to X_{c1} dx by N though you have this one the minus S_s by A into integration X_{c1} to X_{c2} but the, to make this positive we have changed the limits that will be nothing but we can say is equals to S_s by A into X_{c2} to X_{c1} dx by $a_1 X$ plus a_2 .

So from there we will be getting that is equals to S_s by A into 1 by a_1 into $\ln \frac{a_1 X_{c1} + a_2}{a_1 X_{c2} + a_2}$ divided by $a_1 X_{c2} + a_2$. Just by manipulating this a_1 value and a_2 value in this equation we will be getting this a final form of this t_{f1} will be like this.

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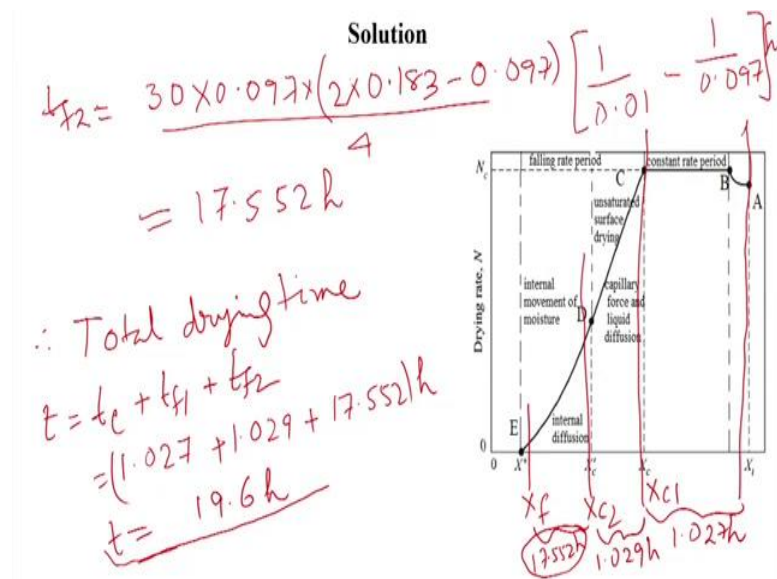


So t_{f1} will be equal to just by putting this a_1 , a_2 values in this equation we have this S_s by A into $2X_{c1}$ minus X_{c2} divided by $2N_c$ into $\ln 2X_{c1}$ minus X_{c2} divided by $2X_{c2}$ and all the values are now known to us we will be putting this value to get the first falling rate time. So it is S_s by A is equals to 30 then X_{c1} is say that is $2X_{c1}$ 2 into 0.183 minus X_{c2} is 0.097 divided by 2 into N_c is equals to 4 into $\ln 2$ into 0.183 minus X_{c2} is equals to 0.097 divided by 2 into 0.097.

So it is coming out as 1.029 hour. So first falling time is 1.029 hour. Now we need to calculate the second falling rate period time. So for that also we need to do one, we need to get the expression for this one but in this case you should this initial condition will be the final condition of the first falling rate period. So it will start from X_{c1} to X_{c2} , So that is why we can say t_{f2} will be like this S_s by A into integration X_f to X_{c2} that is dx by N we told that this is the drying rate is proportional to the square of this moisture content β into X square.

So we will be putting this N is in place of N we will putting βX square so we will be writing this one S_s by A this integration X_f to X_{c2} dx by β into X square. So that will be like this is equals to we can say S_s by A into 1 by β and integration X_f to X_{c2} , X $((\frac{1}{X}))$ minus 2 into dx .

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That is coming out as a, Ss by A into beta and into say 1 X_f minus 1 by X_{c2} . That is we can say this t_{f2} if we put all this values we will be getting like this, so t_{f2} will be say 30 into 0.097 into 2 into 0.183 minus 0.097 divided by 4 into 1 by the final one is 0.01 minus X_{c2} is equals to 0.097 that is in hour. So that will be coming out to be 17.552 hours.

So the total time there for we can say the total drying time t is equals to constant drying period then first falling rate then second falling rate. So that is coming out to be 1.027 plus 1.029 plus 17.552 this hours that is coming out to be 19.6 hours. So this total drying time is 19.6 hours. So here we say out of this three drying times say the first drying time we can say this one from X_i to X_{c1} then from X_{c1} to X_{c2} then from X_{c2} to X_f .

So it takes only 1.027 hour this takes 1.029 hours but his X_{c2} to X_f it took 17.552 hours. So that is I say at end the drying rate actually becomes very slow due to we can say this one the less drying force it takes more time. So where as the total time is 19.6 hours.

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Problem 5

A wet solid having 25% moisture is to be dried at a rate of 1000 kg/h to 1% moisture in a continuous counter current dryer. The drying air enters at 100°C at a rate of 12,000 kg/h (dry basis) with a humidity of 0.025 kg per kg dry air and the dry solid leaves at 60°C. The temperature of the wet solid entering the dryer is nearly the same as the adiabatic saturation temperature of the air leaving the dryer.

Following data and information are available:

Gas-phase mass transfer coefficient for drying of the solid, $k_y = 150 \text{ kg/m}^2 \text{ h}$; effective surface area of the solid = $0.065 \text{ m}^2/\text{kg dry solid}$; specific heat of the solid, $c_{ps} = 0.96 \text{ kJ/kg K}$; critical moisture of the solid is 8%; the equilibrium moisture is negligible. All moistures are on wet basis. Calculate the drying time.

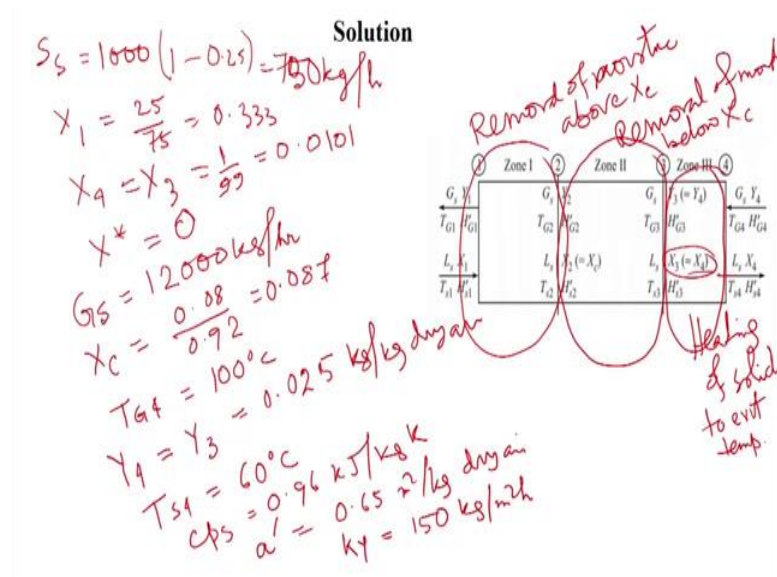
Now we will be discussing about another problem that say for this constant continues dryer the drying time calculation we will be doing this one and this is a very interesting problem also. Wet solid having 25 percent moisture is to be dried at a rate of 1000 kg per hour to 1 percent moisture in a continues counter current dryer. The drying air intercept 100 degree Celsius at a rate of 12,000 kg per hour dry basis with a humidity of 0.025 kg per kg dry air and the dry solid leaves at 60 degree Celsius.

The temperature of the wet solid entering the dryer is nearly the same as the adiabatic saturation temperature of the air leaving the dryer. That is we can say that is we can assume also. The following data and information are available for this drawing operation. The gas-phase mass transfer coefficient for drawing of the solid k_y that is a 150 kg per meter square hour, effective surface area that is a star a prime is equal to 0.065 per meter square per kg. Specific heat the cps of the solid is 0.96 kg per kg Kelvin and critical moisture of the solid is 8 percent the equilibrium moisture is negligible.

All moisture are on wet basis we need to calculate the drying time. So in this case as this is continues counter current dryer so definitely we have different drying zone here also from the problem we assume that the drawing there are three drying zone that is one either zone 1 and zone 2 and zone 3. Where zone 3 is the we can say this one only heating of the final dried product to the, we can say desired temperature.

And in zone 1 the, we can say this one the initial drying starts and in zone 2 it is we can say this one temperature remains almost constant throughout the drying period. So we will be solving this one systematically.

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So whatever the data points we have just will be first jot down those. Like this we can say this one is this zone 1 that is we can say this one revolve of moisture above X_c that is we can say this one revolve of moisture above X_c and in zone 2 that is we can say this one revolve of moisture below X_c and in zone 3 here we can say this one only heating of the solid to exit temperature.

So this three zones, so whatever the given parameter we have in the problem we will now first jot down all this S_s is actually given as say 1000 is the flow rate of this wet solid with this we can say this 25 percent moisture content. So it is 1 minus .25 so we can say it is 750 kg that is per hour. Then X_1 will be definitely say 25 by 75 that is will be 0.333 and X_4 where it will be X_4 or X_3 both are same when it is executive actually from their because you see after that the moisture content does not changes. So that is X_3 and X_4 both are same.

So that is we can say this one that is equals to X_3 also is equals to so that is 0.0101 like this and X^* for this means it is equilibrium moisture content that is 0 almost 0. So we can assume this one as zero and gas flow rate this drying gas flow rate is given as 12,000 kg per hour and X_c is given as 0.08 by 0.92 that is equal to 0.087 and T_{G4} that is also given as we can say 100 degree Celsius and y_4 that is equal to y_3 also is equal to 0.025 kg dryer and T_{S4} is equal to 60 degree Celsius and c_{ps} is equals to 0.96 kilo joule per kg kelvin.

And a prime where specific area is given as 0.65 meter square per kg dryer and mass transfer cooperation value k_y is given as 150 kg per meter square hour. These are the parameters given in this problem.

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Solution

Overall moisture balance

$$S_s (X_1 - X_4) = G_s (Y_1 - Y_4)$$

$$750 (0.333 - 0.0101) = 12000 (Y_1 - 0.025)$$

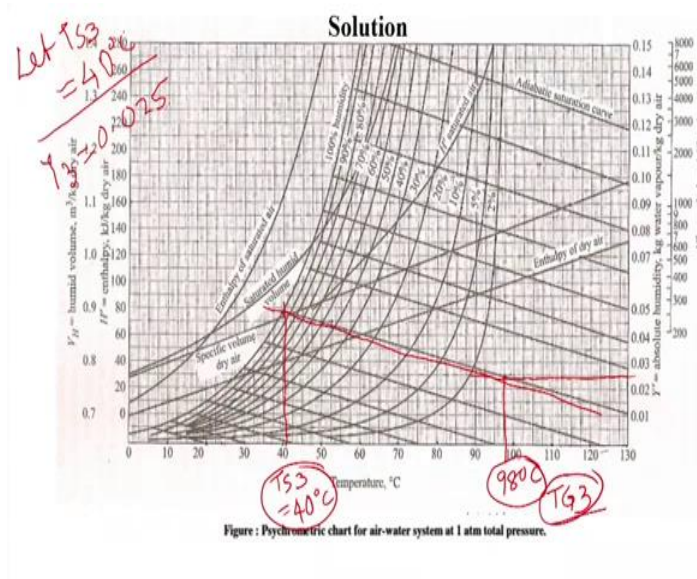
$$Y_1 = 0.0452 \text{ kg/kg dry air}$$

$$C_H = 1.005 + 1.88 Y_4 = 1.005 + 1.88 \times 0.025 = 1.052 \text{ kJ/kg K}$$

So we will be doing now overall moisture balance to get this y_1 value that is first that is unknown parameter we will be calculating this one from this if we do the warm moisture balance across this enter dryer. So we will be doing this warm moisture balance like this. So we will be taking this S_s into X_1 minus X_4 here S_s is both are same is equals to G_s into y_1 minus y_4 so that is we can say this one 750 into X_1 is equals to 0.333 minus X_4 is equals to 0.0101 and G_s is given as 12,000 y_1 we do not know minus 4 is given as 0.025 so from here we will be getting this y_1 is equal to 0.0452 kg per kg dryer.

So now we have got this y_1 that is we can say this one this moisture content of this drying air which is leaving the dryer. So we have got this y_1 now. Now humid heat actually of the air C_H in zone 3, we will be starting here actually. So we can say this one C_H , C_H is equals to say 1.005 plus 1.88 into y_4 that is we can say 1.005 plus 1.88 into that y_4 is equal to 0.025. That is coming out as 1.052 kJ/kg dry air per kelvin.

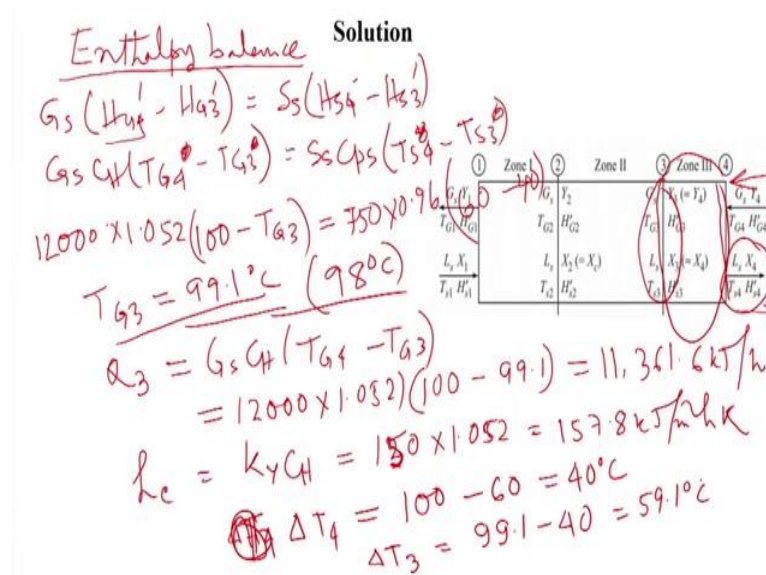
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Now actually we need to take the help of this psychologic chart to get this T_{g3} . So we will be doing like this we need to assume say for this one say let T_{s3} will be assume let T_{s3} that is we assume on 1 value actually we are assuming now like 40 degree Celsius. For the time being we are assuming this T_{s3} is equals to this 40 degree that is nothing but we can say this one automatic saturation temperature.

So we will be taking this one at 40 degree Celsius like this here and then we will be taking this we can say y_3 actually is given as point y_3 that is given as 0.025 that is already given 0.025 let us take this one this line and 0.025 this line. So here we see we can say this one let us take this is 98 degree Celsius. So that is we can say this one T_{g3} that is T_{g3} . So let us take this one is T_{g3} whether we are our assumption is correct or not we can get this one from the enthalpy balance also. It is for the, we can say for simplicity we have assume this one we assumed as this T_{s3} is equal to 40 degree it may be rare temperature also but it is Celsius for that we will be getting T_{g3} at as the 98 degree Celsius.

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For that let us take so this will be doing the enthalpy balance to Nco whether this T_{g3} is correct or not. Let us do the enthalpy balance very easily you can do this one. In this zone 3 we will be doing this one that G_s into H_{g4} prime minus H_{g3} prime is equal to S_s into H_{s4} prime minus H_{s3} prime. Here you see this except this we can say this one again we can say this one we can convert this into a in terms of H we can see we can write in the time the form of a Ch into temperature like this or G_s into Ch into T_{g4} prime minus T_{g3} prime is equals to S_s into c_{ps} into T_{s4} prime minus T_{s3} prime.

So here you see this G_s is equals to we can say this 12000 into Ch is equal to 1.052 into T_{g4} is equals to is given already as 100 degree Celsius T_{g3} prime will be getting T_{g3} say than it is equals to S_s is given as 750 c_{ps} is given as 0.96 and T_{s4} is given as 60 and T_{s3} is given as 40. So now from here we can say T_{g3} so these are not prime values actually so T_{g3} is equals to we can say this 99.1 degree Celsius and from this psychrometric chart we got 98 degree Celsius.

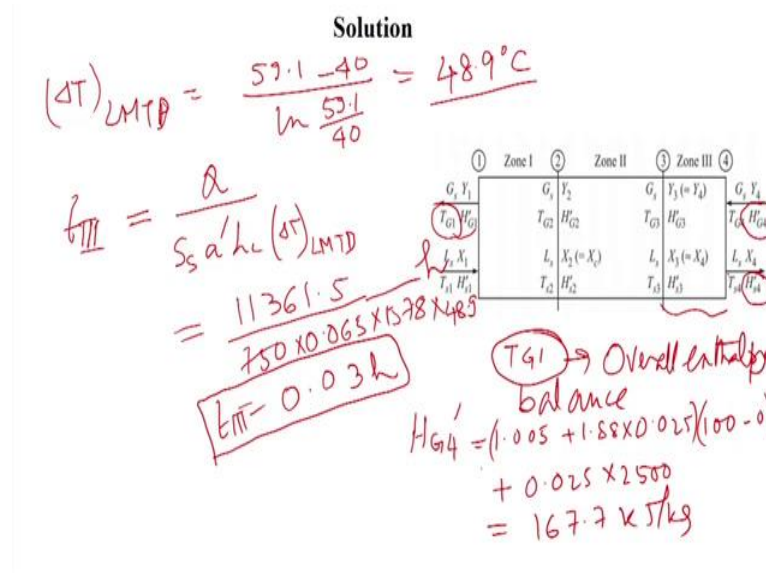
So we can say may be both are almost similar. So we can assume that the T_{g3} is equals to say 99.1 degree Celsius. Now in zone 3 so in zone 3 we can say this one Q_3 whatever the heat flow actually is there is equal to we can say this one G_s into Ch into T_{g4} minus T_{g3} , so that comes out as 12,000 into 1.052 into this is T_{g4} is equal to 100 minus T_{g3} 99.1.

So that is coming out as 11,361.6 kilo joules per hour and we have this L_c is equals to K_y into Ch . So that we have this one K_y is equal to 150 into Ch is equals to 1.052. So that is, so 157 this is 150 that is 157.8 so kilo joules per meter square per hour Kelvin. Now we need to get the temperature driving force so for that we have this one in this in this station 4 or

wherever we can say this one hot drying solid is exiting and drying gas actually is entering there we say ΔT_4 that is we can say is equals to say 100 minus 60.

So this is 40 degree Celsius and here in station 3 this is we can say this is on temperature driving force ΔT_3 is equal to say, we can say 99.1 minus 40 so it is 59.1 degree Celsius.

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So from here actually we will be getting that say LMTD ΔT LMTD will be like this 59.1 minus 40 divided by $\ln 59.1$ by 40. So this is coming out as 48.9 degree Celsius in this zone we can say this one temperature drying force is 49 degree Celsius and say we know this we have already this one this T_3 in the previous class we have already derived that is in T_3 this is Q by S_s into a prime into L_c into LMTD that is equals to Q is equals to 11,361.5 divided by 750 into a is 0.065 into L_c is given as 157.8 and LMTD is 48.9.

So that is coming out in hour that is we got this time as 0.03 hour. So this T_3 is 0.03 hour. Now we will be going to this zone 2. So in zone 2 first we need to get T_{G1} because this temperature is actually is not known we need to do by some this we can say indirect method we will say for this getting for this T_{G1} we need to do the overall enthalpy balance in zone 2 from there actually we will be getting like this before that actually we need to calculate this, suppose we need to get T_{G1} .

So calculation of T_{G1} for that we will be doing say overall enthalpy balance. We need to get this from this overall enthalpy balance before that we need to get this one we have already derived this one for enthalpy value for gases enthalpy values for solids. This H_{G4} prime that

will be like this 1.005 plus 1.88 into say y prime that is into 0.025 into temperature gradient this 100 minus T0 is equals to 0 plus this y prime 0.025 into 2500.

It is coming out as 167.7 kilo joules per kg. So that is Hg4 prime, however it is entering here.

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Solution

$$H_{g1} = (c_{ps} + X_1 c_{p1})(T_{s1} - T_0) = 94.2 \text{ kJ/kgK}$$

$$= (0.96 + 0.0101 \times 4.187)(60 - 0)$$

$$= 60.14 \text{ kJ/kgK}$$

$$H_{s1} = (c_{ps} + X_1 c_{p1})(T_{s1} - T_0)$$

$$= 94.2 \text{ kJ/kgK}$$

$$G_s (H_{g4} - H_{g1}) = S_s (H_{s4} - H_{s1})$$

$$12000 (167.7 - H_{g1}) = 750 (60.14 - 94.2)$$

$$H_{g1} = 169.8 \text{ kJ/kg}$$

$$\Rightarrow T_{g1} = 52^\circ\text{C}$$

Zone I		Zone II		Zone III	
G_1, Y_1	G_2, Y_2	$G_3, Y_3 (= Y_4)$	G_4, Y_4		
T_{G1}, H_{G1}	T_{G2}, H_{G2}	T_{G3}, H_{G3}	T_{G4}, H_{G4}		
L_1, X_1	$L_2, X_2 (= X_3)$	$L_3, X_3 (= X_4)$	L_4, X_4		
T_{L1}, H_{L1}	T_{L2}, H_{L2}	T_{L3}, H_{L3}	T_{L4}, H_{L4}		

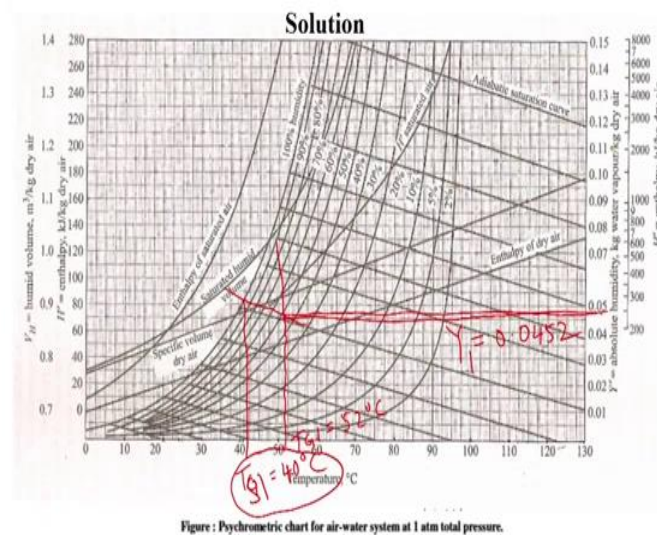
Similarly for getting this Hs4 prime we have another formula we will be using that for Hg for gas which is solid actually Hs4 prime that is whatever is coming out this we can say drying solid the enthalpy value will be like this cps plus X1 into Cpl into Ts1 minus T0 but doing this one 0.96 plus pint 96 plus 0.0101 into 4.187 into 60 minus 1, 0. So it is coming out as 60.14 kilo joules per kg Kelvin and for Hs1. So we have this cps plus that will be actually X4, X1 into cpl into Ts1 minus T0.

It is coming out as 94.2 kilo joules per kg Kelvin. Now we will be doing the overall enthalpy balance of course the entire dryer so we will be getting like this Gs into Hg4 prime minus Hg1 prime is equals to Ss into Hs4 prime minus Hs1 prime. So we will be putting all this values to get this Hg1 prime that is not actually available. So we will be doing this one 12,000 into Hg4 prime we have already got this value 167.7 minus Hg1 prime is equals to 750 into that is we have this 60.14 minus 94.2.

So from here we will be getting this Hg1 prime that we will be getting as 169.8 kilo jul per kg. So that is we can say this that will be equal to just using this formula 1.005 plus 1.88 into y1 prime that is we can say this one 0.4, 0452 into Tg1 minus Tg0 plus 2500 into y1 prime that is equals to 0.0452. That is equal to we can say this one we will be putting this is equal to, here you see T0 is equal to 0 degree Celsius.

So we will be putting here 0. So now from this equation 169.8 is equal to 1.005 plus 1 0.88 into 0.0452 into T_{g1} plus 2500 into 0.0452. So from there we will be getting T_{g1} is equal to you can say 52 degree Celsius. Now we see this gas temperature is now known this T_{g1} is known. What will be the T_{s1} for that actually we can say this and humidity this y_1 prime is already obtained as 0.0452.

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So we will be going to this one in psychrometric chart, so we have this T_{g1} is equals to gas temperature we can say this one 52 degree Celsius. So 52 degree Celsius at T_{g1} and 0.0452 like this we have this we have this one. So from here if we follow this adiabatic saturation line we will be getting we can say this one temperature as like this are, let us take it is say T_g this on T_{g1} , T_{s1} that will be equal to we can say let us say 40 degree Celsius.

So for 52 degree Celsius T_{g1} , T_{g1} is equals to 52 degree Celsius with y_1 is equal to 0.0452 we have T_{s1} is equal to just 40 degree Celsius.

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Solution

$$X_2 = X_c = 0.087$$

$$H_{s2}' = (C_{ps} + X_2 C_{pl})(T_{s2} - T_0) = (0.96 + 0.087 \times 4.187)(40 - 0)$$

$$= 52.97 \text{ kJ/kg K}$$

$$H_{s3}' = (C_{ps} + X_3 C_{pl})(T_{s3} - T_0)$$

$$= (0.96 + 0.0101 \times 4.187) \times 40$$

$$= 40 \text{ kJ/kg K}$$

Moisture balance

$$G_s(Y_2 - Y_3) = S_s(X_2 - X_3)$$

$$12000(Y_2 - 0.025) = 750(0.087 - 0.0101)$$

$$Y_2 = 0.0298 \text{ kg dry air}$$

$$H_{s3}' = (1.005 + 1.88 \times 0.0298)(99 - 0) + 2500 \times 0.0298$$

$$= 166.8 \text{ kJ/kg K}$$

Now we will be getting this X_2 values that is we can say this one X_2 that is nothing but this we can say this one X_c that is we have already this one obtained as 0.087. We have this value and from there actually we will be getting H_{s2}' that is that is we can say this c_{ps} plus X_2 into c_{pl} that is liquid water into T_{s2} minus T_0 that is we can say this one will be getting as 0.96 plus X_2 0.087 into 4.187 into T_{s2} that is assume this as 40 and T_{s0} is equal to 0, 40 minus 0.

So from here actually we will be getting H_{s2}' prime is equals to 52.97 kilo Joules per Kg kelvin, and similarly this H_{s3}' prime we will be calculating by this like c_{ps} plus X_2 into c_{pl} into T_{s3} minus T_0 so here we can say it is say c_{ps} that is 0.96 plus H_{s3} that is will be X_3 actually that is 0.0... and X_3 X_4 both are same 0.0101 into 4.187 into 40.

So that is coming out as 40 okay kilo Joules per kg Kelvin. Now we will be doing the moisture balance in this zone 2, in the moisture balance so we will be doing, moisture balance like G_s into y_2 minus y_3 is equal to S_s into X_2 minus X_3 . So this is 12,000 into y_2 minus 0.025 this is y_3 we have is equal to 750 into X_2 we have 0.087, 0.087 and X_3 that is X_4 both are same 0.0101.

So from here we will be getting y_2 that is equal to we can say 0.0298 kilo joule per kg dryer and now we can calculate this H_{g3}' prime that is equal to just by putting by this value this we will be getting 1.005 plus 1.88 into this y_3 that is 0.025 into 99 minus 0 that is y_3 that is minus 0 plus 2500 into y_3 0.025. So that is coming out as 166.8 kilo joules per kg Kelvin.

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Solution

Enthalpy balance

$$G_s(H_{G3}' - H_{G2}') = S_s(H_{S3}' - H_{S2}')$$

$$12000(166.8 - H_{G2}') = 750(40 - 52.97)$$

$$H_{G2}' = 167.6 \text{ KJ/kg K}$$

$$= (1.005 + 1.88 \times 0.0298)(T_{G2} - T_0) + 2500 \times 0.0298$$

$$T_{G2} = 87.5^\circ\text{C}$$

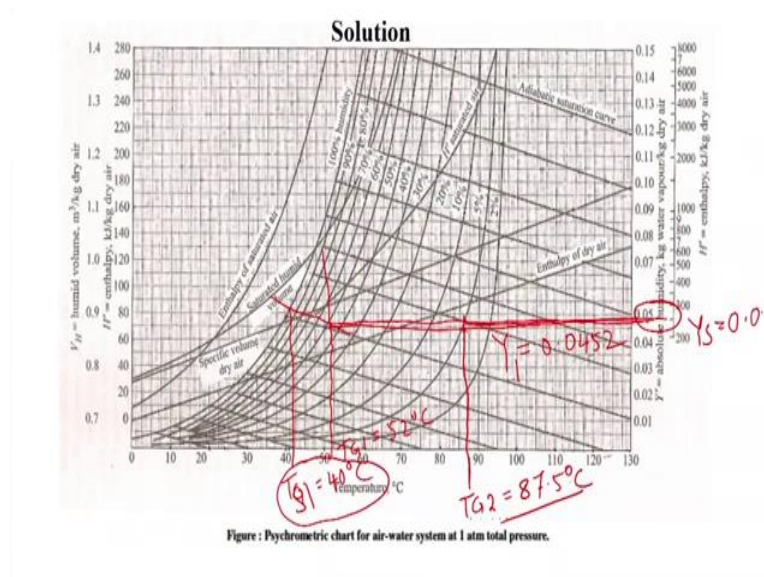
$$b_{II} = \frac{G_s X_C}{S_s a' k_y \left[X_4 + \frac{G_s}{S_s} (Y_5 - Y_4) \right]} \ln \frac{X_C (Y_5 - Y_3)}{X_4 (Y_5 - Y_2)}$$

So now we have and we need to get this one this T_{G2} that is not this one we do not have T_{G2} value we will be calculating this one. So we will be doing the enthalpy balance G_s into H_{G3} prime minus H_{G2} prime is equal to S_s into H_{S3} prime minus H_{S2} prime that is we can say 12,000 into H_{G3} prime we have just calculated 166.8 minus H_{G2} prime we do not know is equal to 750 into H_{S3} prime is 40 minus H_{S2} prime is 52.97.

From here actually we will be getting H_{G2} prime. So that is coming out at 167.6 kilo joules per kg Kelvin. So there actually we can say this one this is nothing but just by using the equation in terms of this temperature we will be getting like this that is equal to 1 point for gas actually 1.005 plus 1.88 into y_2 that is 0.0298 into T_{G2} we do not know T_{G2} minus T_0 that is 0 degree plus 2500 into y_2 that is 0.0298.

So from this equation 167.6 is equal to 1.005 plus 1.88 into 0.0298 into T_{G2} minus T_0 that is 0 plus 2500 into 0.0298 we will be getting the T_{G2} is equal to say 87.5 degree Celsius. So we have T_{G2} so whatever the T_s actually will be we can say this one this T_s that is we can say T_s is equal to 40 degree Celsius.

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We know suppose this T_{s1} is actually 40 degree Celsius and now we have T_{g2} is equal to 87.5 degree Celsius, so this 87.5 degree Celsius like here so we can say if we have this the T_{g2} is 80 T_{g2} is equal to say 87.5 degree Celsius and T_{s1} actually we have this 40 degree Celsius and say whatever will be the y_s . It will be like this so y_s will be is equal to say 0.05.

So now we need to get this t_2 whatever the time for this second joules time require to dry the second joules so t_2 we can say this one G_s into X_c by S_s into a prime into k_y into 1 by X_4 plus G_s by S_s into Y_s minus y_4 into $\ln X_c$ into Y_s minus y_3 by X_4 into y_s minus y_2 .

Now all the values are actually known to us we will be getting all this values and we will be putting then we will be getting.

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Solution

$$t_{II} = \frac{12000 \times 0.087}{750 \times 0.065 \times 150} \left[0.0101 + \frac{12000}{750} (0.05 - 0.025) \right] \ln \frac{0.087 (0.05 - 0.0278)}{0.0101 (0.05 - 0.0278)}$$

$$= 0.824 \text{ h}$$

Zone I

$$t_I = \frac{G_s}{S_s a' k_y} \ln \frac{Y_s - Y_2}{Y_1 - Y_1}$$

$$= \frac{12000}{750 \times 0.065 \times 150} \ln \frac{0.05 - 0.0298}{0.05 - 0.0452}$$

$$= 2.358 \text{ h}$$

Total drying time = $(0.03 + 0.824 + 2.358) \text{ h}$
 $= 3.21 \text{ h}$

	① Zone I	② Zone II	③ Zone III	④
G, Y	$G_1 Y_1$	$G_2 Y_2$	$G_3 Y_3 (= Y_4)$	$G_4 Y_4$
T, H	$T_{G1} H_{G1}$	$T_{G2} H_{G2}$	$T_{G3} H_{G3}$	$T_{G4} H_{G4}$
L, X	$L_1 X_1$	$L_2 X_2 (= X_3)$	$L_3 X_3 (= X_4)$	$L_4 X_4$
T, H	$T_{L1} H_{L1}$	$T_{L2} H_{L2}$	$T_{L3} H_{L3}$	$T_{L4} H_{L4}$

So the time required in this zone 2 t_2 will be 12,000 into 0.087 divided by 750 into 0.065 into 150 into 1 by 0.0101 plus 12,000 divided by 750 into 0.05 minus 0.025 into $\ln X_c$ that is 0.087 into y_s that is 0.05 minus y_3 0.025 divided by X_4 that is 0.0101 into y_s that is 0.05 minus y_2 that is 0.0298.

So this is coming out at 0.824 hour. Now we will be calculating the time for zone 1 so that is for zone 1 we have this total time T_1 is equal to G_s by S_s into a prime into k_y into $\ln y_s$ minus y_2 by y_s minus y_1 . So here G_s is equal to 12,000 by S_s means 750 into a prime is 0.065 into k_y is equal to 150 into $\ln y_s$ is equal to 0.05 minus 0.0298 divided by 0.05 minus 0.0452.

So it is coming out as 2.358 hour. So this is T_1 , so total time we can say this one total drying time will be total drying time is equal to we can say T_1 plus T_2 plus T_3 that is we have this 0.03 plus 0.824 plus 2.358. So that is total 3.21 hour. So this total drying time is obtained as 3.21 hour.

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So thank you we will start this next topic that is liquid-liquid extraction in the next class