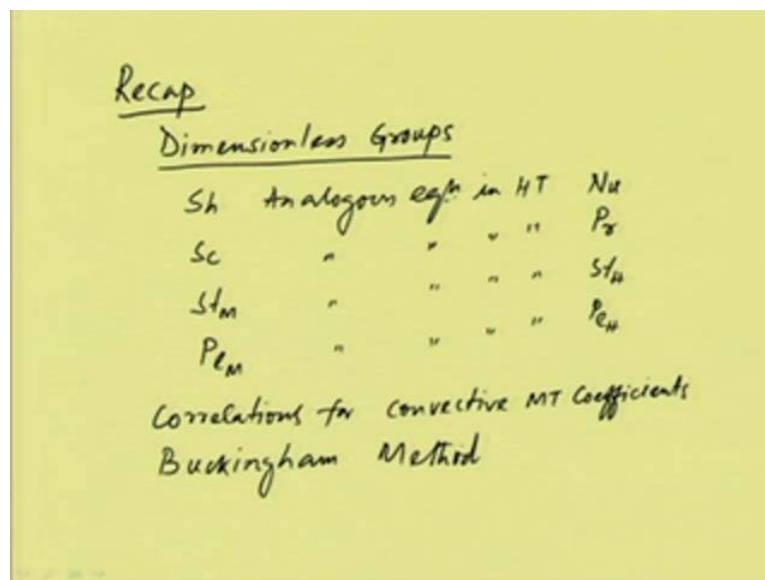


Mass Transfer Operations - I
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Module - 2
Mass Transfer Co-efficient
Lecture - 3
Mass transfer co-efficient in Laminar Flow Condition

Welcome to the third lecture of module 2. The module 2 is on mass transfer coefficients. In this lecture, we will consider the mass transfer coefficient in laminar flow conditions. So, before going to this lecture, let us have a quick recap of our previous lecture, what we have discussed.

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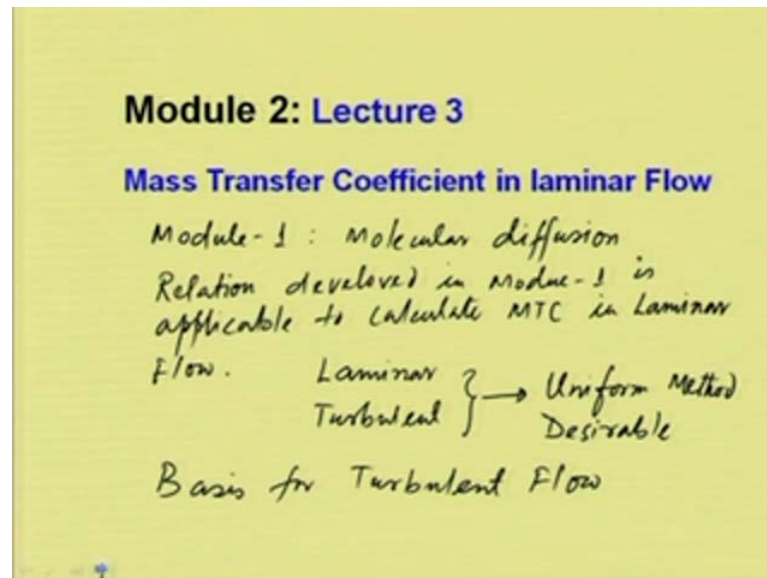


In our previous lecture, we have discussed dimensionless groups; in this, particularly, we have discussed Sherwood number, which is analogous equation in heat transfer is Nusselt number and Schmidt number; analogous equation in heat transfer is Prandtl number; Stanton number, for mass transfer; and similarly, in heat transfer, it is Stanton number for heat transfer; and then Peclet number for mass transfer and analogous equation in heat transfer is also Peclet number.

We have also discussed how to obtain the correlations for mass transfer coefficients, the correlations for convective mass transfer coefficients. Here, we have particularly

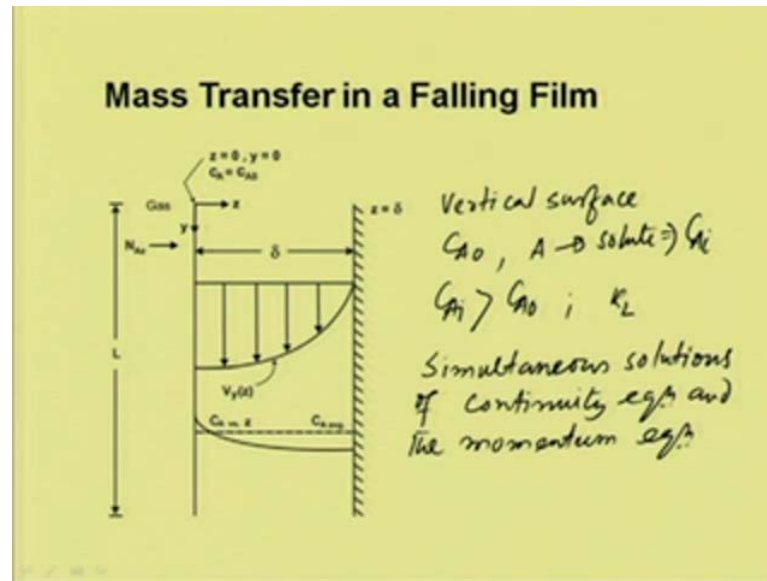
discussed Buckingham method to obtain correlations for different systems. So, in this lecture, we will consider mass transfer coefficient in laminar flow. Practically, we do not need to discuss the mass transfer coefficient for laminar flow, because whatever we have discussed in module 1, which is the molecular diffusion mass transfer, that will be prevailed in case of laminar flow conditions.

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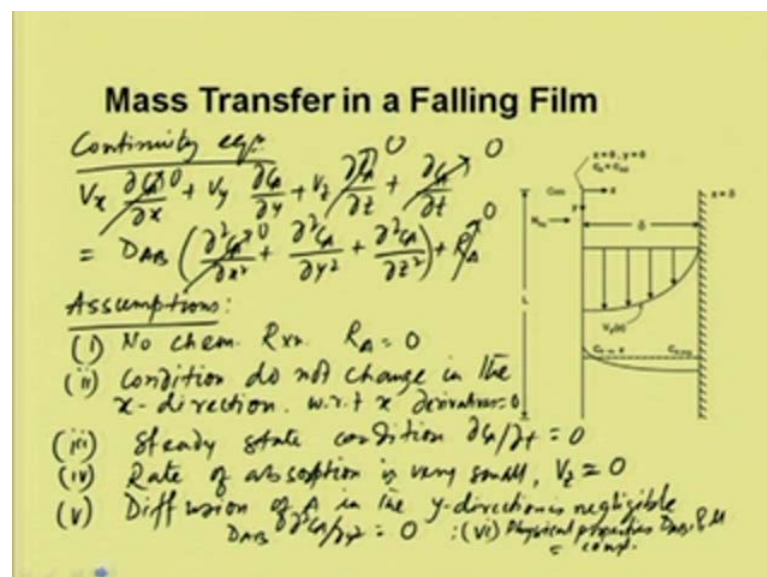
In module 1, we have discussed molecular diffusion and the relation developed in module 1 is applicable to calculate mass transfer coefficient in laminar flow. But a uniform method is desirable for the laminar and turbulent flow conditions; for laminar and turbulent, a uniform method is desirable. So, we will consider a simple case, and then, this lecture will make the basis for turbulent flow.

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Now, let us consider a mass transfer in a thin falling film. Consider a vertical surface; this is vertical surface and where the liquid is falling from the top; and it is at initial concentration C_{A0} and at the interface, the concentration of the solute A is C_{Ai} , which is equilibrium with the partial pressure of A in the gas phase; and C_{Ai} is greater than C_{A0} , so that, the solute is dissolved in the liquid phase. And, the liquid phase mass transfer coefficient is K_L . So, this problem requires the simultaneous solutions of continuity equation and the momentum equations, that is, the Navier-Stokes equation, with some simplifying assumptions.

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Now, let us consider the equation of continuity, which we have derived in case of unsteady state mass transfer, that is, continuity equation $V_x \frac{\partial C_A}{\partial x} + V_y \frac{\partial C_A}{\partial y} + V_z \frac{\partial C_A}{\partial z} + \frac{\partial C_A}{\partial t}$ will be equal to $D_{AB} \frac{\partial^2 C_A}{\partial x^2} + \frac{\partial^2 C_A}{\partial y^2} + \frac{\partial^2 C_A}{\partial z^2} + R_A$, the rate of reaction. Now, for this case, following assumptions can be made. No chemical reaction; so, R_A will be 0. Conditions do not change in the x direction. So, all derivatives with respect to x, with respect to x, all derivatives will be 0. So, this will be 0; this will be 0. And, steady state conditions prevail; so, this will be 0; steady state condition; $\frac{\partial C_A}{\partial t}$ will be 0. The rate of absorption of gas is very small. So, V_z is essentially 0. So, in this case, this will be 0; and, diffusion of A in the y direction is negligible. So, in these conditions, $D_{AB} \frac{\partial^2 C_A}{\partial y^2}$ will be 0. So, using this, and all other physical properties and chemical properties, that is, D_{AB} , ρ and μ , these are constant. So, using these assumptions, this equation reduces to the following form.

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Mass Transfer in a Falling Film

$$V_y \frac{\partial C_A}{\partial y} = D_{AB} \frac{\partial^2 C_A}{\partial z^2}$$

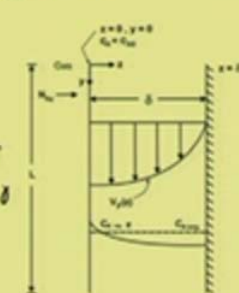
Egt. of motion

$$\mu \frac{\partial^2 V_y}{\partial z^2} + \rho g = 0$$

Soln: BC's: $V_y = 0$ at $z = \delta$
 $dV_y/dz = 0$ at $z = 0$

$$V_y = \frac{\rho g \delta^2}{2\mu} \left[1 - \left(\frac{z}{\delta} \right)^2 \right]$$

Maximum velocity $V_{y,max}$ at $z = 0$

$$V_{y,max} = \frac{\rho g \delta^2}{2\mu}$$


$V_y \frac{\partial C_A}{\partial y}$ will be $D_{AB} \frac{\partial^2 C_A}{\partial z^2}$. So, the equations of motion under these conditions will be reduced to $\mu \frac{\partial^2 V_y}{\partial z^2} + \rho g = 0$. And, the solutions of this equation with the conditions, boundary conditions, we can use V_y is equal to 0, at z equal to δ ; that is, at the surface; and dV_y/dz is 0, at z is equal to 0. So, the solution is well known and we can write V_y is equal to $\frac{\rho g \delta^2}{2\mu} \left[1 - \left(\frac{z}{\delta} \right)^2 \right]$. Now, the maximum velocity occurs at the surface. So, the maximum velocity $V_{y,max}$ will be at z

equal to 0; that is, at the gas liquid interface. So, V_y max, putting these conditions in this equation, V_y max will be $\rho g \delta^2$ by twice μ .

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Mass Transfer in a Falling Film

Bulk-average velocity

$$V_{y, avg} = \frac{1}{\delta} \int_0^\delta V_y dz$$

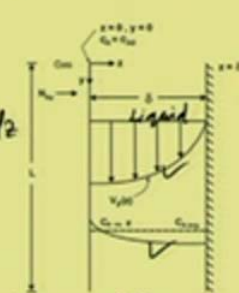
$$= \frac{1}{\delta} \int_0^\delta \frac{\rho g \delta^2}{2\mu} \left[1 - \left(\frac{z}{\delta}\right)^2\right] dz$$

$$= \frac{\rho g \delta^2}{3\mu}$$

$$\frac{3}{2} V_y \left[1 - \left(\frac{z}{\delta}\right)^2\right] \frac{\partial C_A}{\partial y} = D_{AB} \frac{\partial^2 C_A}{\partial z^2}$$

BC'S

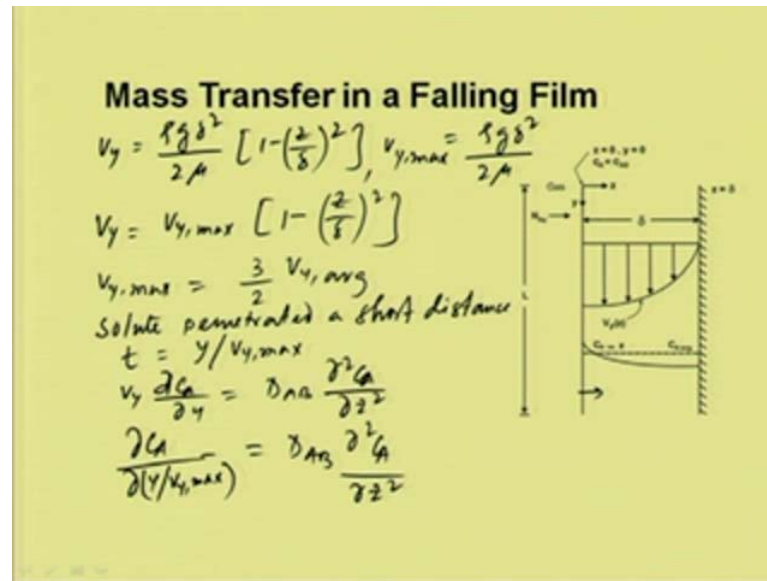
- (i) At $z=0$, $C_A = C_{Ai}$ for all y
- (ii) At $z=\delta$, $\partial C_A / \partial z = 0$ for all y
- (iii) At $y=0$, $C_A = C_{Ao}$ for all z



Now, the bulk average velocity can be obtained, V_y average will be equal to $\frac{1}{3} \rho g \delta^2 / \mu$, δ is the thickness of the liquid film, this is liquid film, integral 0 to δ $V_y dz$. So, V_y , we have obtained, V_y equations, $\rho g \delta^2$ by twice μ into $1 - z^2$ by δ^2 . So, if we include that, integral 1 by δ , integral 0 to δ , $\rho g \delta^2$ by twice μ into $1 - z^2$ by $\delta^2 dz$. So, integrating, this will be equal to $\rho g \delta^2$ by thrice μ . And, the film thickness, we can write, δ will be $3 \mu V_y$ average divided by ρg to the power half.

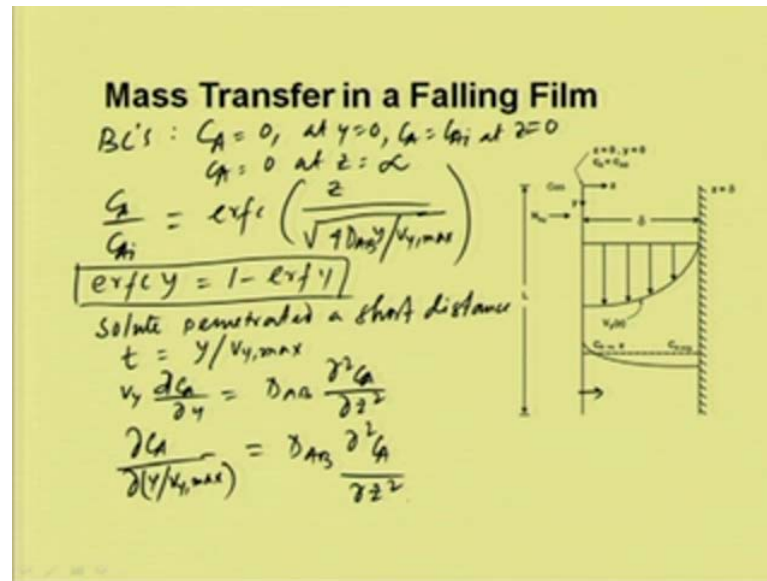
Now, if we substitute the velocity profile in the continuity equations, we have $3/2 V_y$ into $1 - z^2$ by δ^2 into $\partial C_A / \partial y$, will be $D_{AB} \partial^2 C_A / \partial z^2$. So, these equations can be solved with the following boundary conditions. One is, at z equal to 0 , C_A will be C_{Ai} , for all y . Secondly, at z is equal to δ , $\partial C_A / \partial z$ will be 0 , for all y ; that is, no diffusion takes place in the solid wall; and third, at y equal to 0 , C_A will be C_{Ao} , for all z . So, with this boundary conditions, if you solve these equations, the concentration profile $C_A(z)$, function of z , is shown with respect to length, this can be, look like this. This is concentration profile and this is the velocity profile. It is a result of infinite series solutions.

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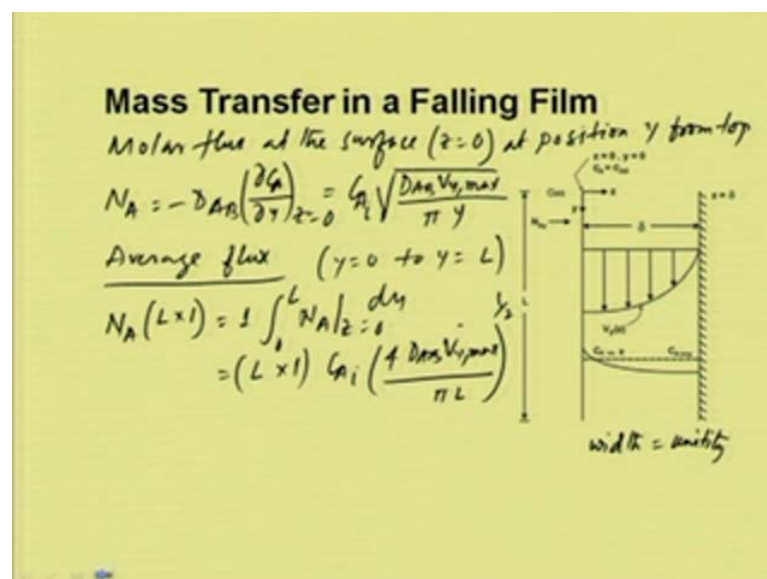
We know that, v_y is equal to $\rho g \delta^2$ by twice μ into $1 - (z/\delta)^2$. And, $v_{y,max}$, when z equal to 0, that is, at the surface, is $\rho g \delta^2$ by twice μ . So, if you divide the ratio v_y by $v_{y,max}$, so, we can write, v_y will be $v_{y,max}$ into $1 - (z/\delta)^2$. So, $v_{y,max}$, again, will be $3/2$ $v_{y,avg}$. We have calculated $v_{y,avg}$. So, the relations between $v_{y,max}$ and $v_{y,avg}$ will be this. Now, if we considered, the solute penetrated a short distance. So, if it is short distance from the surface, so, then, we can assume, the velocity will remain nearly maximum velocity. So, the time t will be y by $v_{y,max}$. So, the continuity equations we have derived, $v_y \frac{\partial C_A}{\partial y}$ will be $D_{AB} \frac{\partial^2 C_A}{\partial z^2}$; we can write $\frac{\partial C_A}{\partial y}$ by $v_{y,max}$; it will be maximum velocity, nearly will be equal to $D_{AB} \frac{\partial^2 C_A}{\partial z^2}$.

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Now, if you use the boundary conditions C_A is equal to 0, at y equal to 0 and C_A will be C_{Ai} , at interfacial concentration, at z equal to 0, and C_A will be 0, at z equal to infinity. So, if you integrate this equation, we can obtain C_A by C_{Ai} will be equal to complimentary error function $\text{erfc}(z \text{ by } \sqrt{4 D_{AB} y / v_{y, \max}})$. So, the definition of complimentary error function is $\text{erfc}(y)$ is equal to 1 minus error function of y . So, this is a tabulated function; from here, we can obtain the values.

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Now, the molar flux at the surface, that is, at z is equal to 0 and at position y from top, we can obtain N_A is equal to minus $D_{AB} \frac{dC_A}{dy}$, at z equal to 0, which is equal to $C_{A,i} \sqrt{\frac{D_{AB} V_{y,\max}}{5y}}$. And, the average flux over the entire surface can be written as N_A entire surface at y equal to 0 to y equal to L ; this is length of the film. So, N_A into L into unit width, width of the film is unity, is equal to $\frac{1}{L} \int_0^L N_A \text{ at } z \text{ equal to } 0 \, dy$. So, this will be equal to $\frac{1}{L} \int_0^L C_{A,i} \sqrt{\frac{D_{AB} V_{y,\max}}{5y}} \, dy$. So, if we integrate these equations, it will be equal to $\frac{1}{L} \times C_{A,i} \times \frac{4}{3} \sqrt{\frac{D_{AB} V_{y,\max}}{5}} \times L$ to the power half. So, this is the flux equation.

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Example

The absorption of pure CO_2 is carried out at 1 atm pressure and at 25°C by using water film flowing down a vertical wall of 1m long. The water is essentially CO_2 -free initially. The average velocity of the liquid is 0.2 m/s. The solubility of CO_2 in water at 25°C and at 1 atm is $c_{A,i} = 0.0336 \text{ kmol/m}^3$.

Calculate film thickness and the rate of absorption of CO_2 .

Use the following properties $D_{AB} = 2 \times 10^{-9} \text{ m}^2/\text{s}$, solution density $\rho = 997 \text{ kg/m}^3$; and viscosity $\mu = 8.95 \times 10^{-4} \text{ kg/m.s}$

Now, let us consider a simple example, where, in a vertical surface, the absorption of pure CO_2 is carried out at 1 atmosphere pressure, and at 25 degree centigrade by using water film falling down the surface; and the length of the surface is 1 meter. The water, initially CO_2 free; and the average velocity of the liquid is given, 0.2 meter per second. The solubility of CO_2 in water at temperature and pressure conditions is given, 0.0336 kilo mole per meter cube. Calculate the film thickness and the rate of the absorption of CO_2 ; and, the physical, chemical properties, diffusivity, density and viscosity are given.

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Example : Solution

$$V_{y,avg} = 0.2 \text{ m/s}, \rho = 997 \text{ kg/m}^3, \mu = 8.95 \frac{\text{kg}}{\text{m}\cdot\text{s}}$$

$$g = 9.81 \text{ m/s}^2, V_{y,max} = \frac{3}{2} V_{y,avg} = \frac{3}{2} \times 0.2 = 0.3 \text{ m/s}$$

$$\delta = \left(\frac{3 V_{y,avg} \mu}{\rho g} \right)^{1/2} = \left(\frac{3 \times 0.2 \times 8.95 \times 10^{-4}}{997 \times 9.81} \right)^{1/2} \text{ m} = 2.34 \times 10^{-4} \text{ m}$$

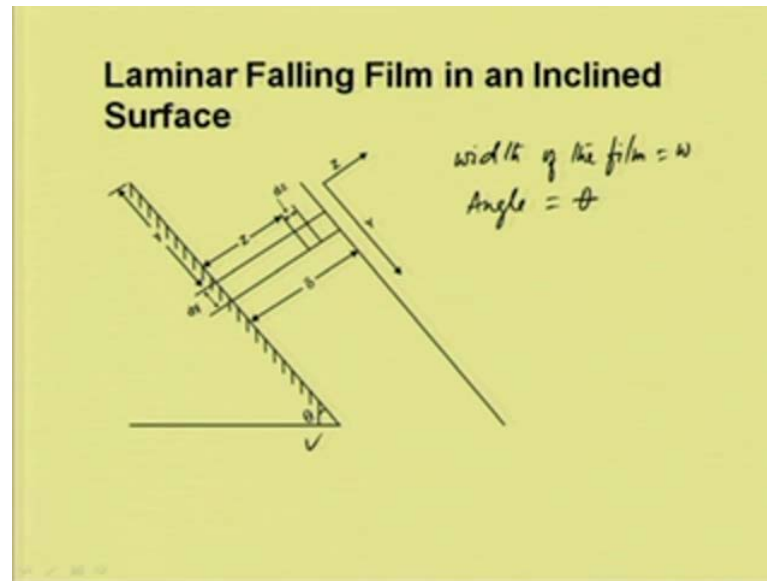
$$C_{Ai} = 0.0336 \text{ kmol/m}^3, D_{AB} = 2 \times 10^{-9} \text{ m}^2/\text{s}, L = 1 \text{ m}$$

$$N_A = C_{Ai} \left(\frac{4 D_{AB} V_{y,max}}{\pi L} \right)^{1/2} = 0.0336 \times \left(\frac{4 \times 2 \times 10^{-9} \times 0.3}{\pi \times 1} \right)^{1/2}$$

$$= 9.29 \times 10^{-7} \frac{\text{kmol}}{\text{m}^2 \cdot \text{s}}$$

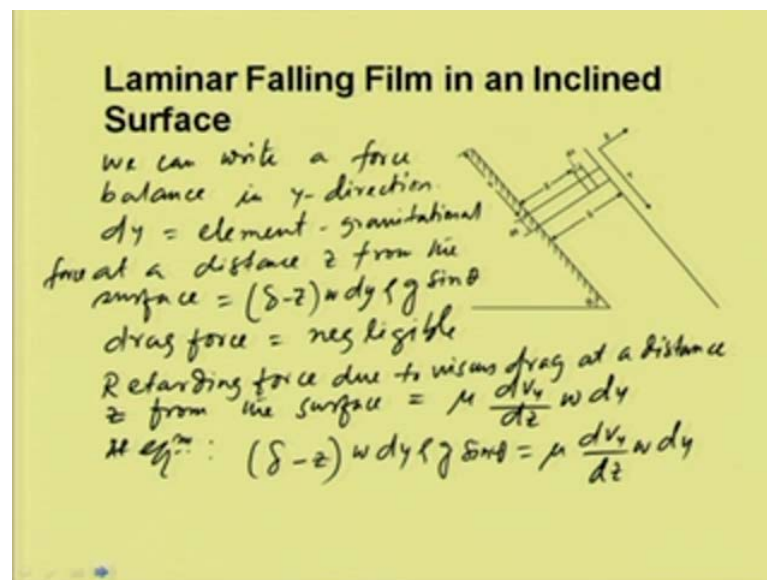
Given that, $V_{y,avg}$ is 0.2 meter per second, ρ is given, 997 kg per meter cubed; viscosity is given, 8.95 kg per meter second; and, g is 9.81 meter per second square; and $V_{y,max}$, we can calculate, 3 by 2 $V_{y,avg}$, which is equal to 3 by 2 into 0.2, is equal to 0.3 meter per second. Now, we can calculate the film thickness δ , is equal to 3 $V_{y,avg} \mu$ by ρg to the power half; so, this will be equal to 3 into 0.2 into 8.95 into 10 to the power minus 4, divided by 997 into 9.81 to the power half meter, which is equal to 2.34 into 10 to the power minus 4 meter. Now, C_{Ai} , the interfacial concentration is given, 0.0336 k mole per meter cube; D_{AB} is equal to 2 into 10 to the power minus 9 meter square per second; L is 1 meter. So, if you substitute, flux N_A will be $C_{Ai} \times 4 D_{AB} V_{y,max}$ by 5 into L to the power half, is equal to 0.0336 into 4 into 2 into 10 to the power minus 9 into 0.3, divided by 5 into 1 to the power half, which is equal to 9.29 into 10 to the power minus 7 k mole per meter square second.

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Now, let us consider another situation, where the laminar film is falling down a inclined surface. So, the depth of the liquid film is delta and width of the film is, is w , and the angle is theta.

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In this case, we can write a force balance in the y direction. Consider a element of fluid $d y$ and the gravitational force acting on this element at distance z from the surface. So, we can write, this is, $(\delta - z) w d y \rho g \sin \theta$; and, the drag force, if we considered negligible, then the retarding force due to viscous drag at z , at a distance z

from the surface, we can write $\mu \frac{dV_y}{dz} = \rho g \sin \theta$; V_y is the fluid velocity. Now, at equilibrium, we can write, $\Delta z \rho g \sin \theta$ will be equal to $\mu \frac{dV_y}{dz}$.

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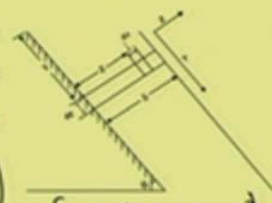
Laminar Falling Film in an Inclined Surface

Assume no slip betⁿ liq & surface, $V_y = 0$ at $z = 0$

$$\int_0^{V_y} dV_y = \frac{\rho g \sin \theta}{\mu} \int_0^{\delta} (\delta - z) dz$$

$$\Rightarrow V_y = \frac{\rho g \sin \theta}{\mu} \left(\delta z - \frac{1}{2} z^2 \right)$$

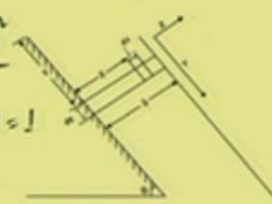
$$\dot{M} = \text{mass rate of flow} = \int_0^{\delta} \frac{\rho g \sin \theta}{\mu} \left(\delta z - \frac{1}{2} z^2 \right) \rho dz$$

$$= \frac{\rho^2 g \sin \theta}{\mu} \left(\frac{\delta^3}{2} - \frac{\delta^3}{6} \right) = \frac{\rho^2 g \sin \theta \delta^3}{3\mu}$$


Now, if we assume no slip condition, no slip between liquid and surface, then, V_y would be 0, at z equal to 0. So, we can integrate the earlier equations, $\int_0^{V_y} dV_y$; this is equal to $\rho g \sin \theta$ by μ ; $\int_0^z \Delta z$. So, V_y will be equal to $\rho g \sin \theta$ divided by μ into Δz minus half z square. So, the mass rate of flow of the liquid down the surface, if we define \dot{M} , which is equal to $\int_0^{\delta} \rho g \sin \theta$, divided by μ into $w \Delta z$ minus half z square ρdz . So, which is equal to $\rho^2 g \sin \theta$ by μ into $w \Delta z^3$ by 2 minus Δz^3 by 6, which is equal to $\rho^2 g \sin \theta w \Delta z^3$ by thrice μ .

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Laminar Falling Film in an Inclined Surface



$$V_{y,avg} = \frac{\dot{M}}{\rho w \delta} = \frac{\rho g \sin \theta \delta^2}{3\mu}$$

For vertical surface $\sin \theta = 1$

$$V_{y,avg} = \frac{\rho g \delta^2}{3\mu}$$

$$V_{y,max} = \frac{\rho g \sin \theta \delta^2}{2\mu} = \frac{3}{2} V_{y,avg}$$

Using the continuity eqn

$$\frac{C_A - C_{A0}}{C_{Ai} - C_{A0}} = \operatorname{erfc} \left(\frac{y}{2\sqrt{D_{AB}t}} \right)$$

The average velocity V_y average can be written as mass flow rate divided by $\rho w \delta$. So, if we substitute \dot{M} , it will be $\rho g \sin \theta \delta^2$ by thrice μ . Now, for vertical surface, $\sin \theta$ will be 1 and V_y average will be equal to $\rho g \delta^2$ square by thrice μ . The maximum velocity will be at the surface, that is, V_y max; we can obtain from the earlier equations, $\rho g \sin \theta \delta^2$ square by twice μ ; and, which is equal to 3 by $2 V_y$ average. The concentration profile can be obtained using the continuity equation. The concentration profile can be obtained as $C_A - C_{A0}$ by $C_{Ai} - C_{A0}$, will be complimentary error function y , divided by 2 into root over D_{AB} into t .

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Example

In an experimental column inclined at 45° with the horizontal, pure CO_2 is absorbed in water at 25°C . The concentration should not reach more than 1% of the saturation value at a depth below the surface at which the velocity is 90% of the surface velocity. What is the maximum length of the column to which the condition can be applied if the average velocity of water is 0.2 m/s? Given that: Viscosity of water = $8.95 \times 10^{-4} \text{ kg/m.s}$, solution density $\rho = 997 \text{ kg/m}^3$ and Diffusivity of CO_2 in water = $2 \times 10^{-9} \text{ m}^2/\text{s}$.

Now, let us consider a very simple example, where the surface is inclined at 45 degree with the horizontal surface and pure CO_2 is absorbed in water at 25 degree centigrade. The concentration should not reach more than 1 percent of the saturation value at a depth below the surface, at which the velocity is 90 percent of the surface velocity. What is the maximum length of the column to which the condition can be applied, if the average velocity of water is 0.2 meter per second; and, given that, all the physio-chemical properties like viscosity, density and diffusivity, which we have used for the earlier problem, is also given.

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Example : Solution

We know that

$$v_{y,avg} = \frac{\rho g \sin \theta \delta^2}{3\mu}$$

$$\delta = \left(\frac{3\mu v_{y,avg}}{\rho g \sin \theta} \right)^{1/2}$$

$$\mu = 8.95 \times 10^{-4} \text{ kg/m.s}; g = 9.81 \text{ m/s}^2$$

$$v_{y,avg} = 0.2 \text{ m/s}; \rho = 997 \text{ kg/m}^3, \theta = 45^\circ$$

$$\delta = \left(\frac{3 \times 8.95 \times 10^{-4} \times 0.2}{997 \times 9.81 \times \sin 45} \right)^{1/2} = 2.54 \times 10^{-4} \text{ m}$$

We know that, V_y average will be $\rho g \sin \theta \Delta z$ divided by thrice μ . So, here, Δz , we can write thrice μ , V_y average divided by $\rho g \sin \theta$ to the power half. So, the values which are given is μ is 8.95×10^{-4} kg per meter second; V_y average, 0.2 meter per second; ρ is 997 kg per meter cube and θ is 45 degree; g is 9.81 meter per second square. Now, if we substitute this in this equation, so, Δz will be $3 \times 8.95 \times 10^{-4} \times 0.2$ divided by $997 \times 9.81 \times \sin 45$ to the power half, which will be equal to 2.54×10^{-4} meter.

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Example : Solution

$$V_y = \frac{\rho g \sin \theta}{\mu} \left[s z - \frac{1}{2} z^2 \right]$$

$$V_{y, \max} = \frac{\rho g \sin \theta s^2}{2\mu}$$

We know that, the velocity V_y is equal to $\rho g \sin \theta$ divided by μ into Δz minus half z square; and $V_{y, \max}$ is equal to $\rho g \sin \theta \Delta z$ square by twice μ .

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Example : Solution

$$\frac{V_y}{V_{y,max}} = \frac{2}{\delta^2} \left(\delta z - \frac{z^2}{2} \right) = \frac{2z}{\delta} - \left(\frac{z}{\delta} \right)^2$$

$$\frac{V_y}{V_{y,max}} = 1 - \left(1 - \frac{z}{\delta} \right)^2$$

Given:

$$\frac{V_y}{V_{y,max}} = 0.9$$

$$1 - \frac{z}{\delta} = 0.316 \Rightarrow \delta - z = 0.316 \delta$$

$$= 8.03 \times 10^{-5} \text{ m}$$

The distance below the surface
 $= 8.03 \times 10^{-5}$

So, if we divide V_y by $V_{y,max}$, we can write, $2z$ by δ^2 into δz minus z^2 by δ^2 , which is equal to $2z$ by δ minus z by δ square, which is equal to 1 minus 1 minus z by δ whole square. And, given that, V_y by $V_{y,max}$ is 0.9 , 90 percent of the surface velocity. So, if you substitute in this equation, 1 minus z by δ would be equal to 0.316 ; and from this, we can write δ minus z is equal to 0.316δ , which is equal to 8.03 into 10 to the power minus 5 meter. So, the distance below the surface is equal to 8.03 into 10 to the power minus 5 meter.

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Example : Solution

$$\frac{C_A - C_{A0}}{C_{A1} - C_{A0}} = \text{erfc} \left(\frac{z}{2 \sqrt{D_{AB} t}} \right)$$

$$\frac{1\%}{0.01} = \text{erfc} \left(\frac{8.03 \times 10^{-5}}{2 \times \sqrt{2 \times 10^{-9} t}} \right)$$

$$= \text{erfc} \left(\frac{0.898}{\sqrt{t}} \right)$$

Now, the concentration profile $C_A - C_{A,0}$ divided by $C_{A,i} - C_{A,0}$ is equal to complimentary error function into z divided by $2\sqrt{D_{AB}t}$. It is given that, it should not reach more than 1 percent of the saturation value. So, this value is given, 1 percent, which is 0.01. So, 0.01 is equal to complimentary error function into z ; this thickness is 8.03×10^{-5} divided by $2\sqrt{D_{AB}t}$. Diffusivity value is given, 2×10^{-9} into t ; so, which is equal to complimentary error function of 0.898 divided by \sqrt{t} .

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Example : Solution

$$\text{erfc } y = 1 - \text{erf } y$$

$$1 - 0.01 = 1 - \text{erfc} \left(\frac{0.898}{\sqrt{t}} \right)$$

$$\Rightarrow \text{erf} \left(\frac{0.898}{\sqrt{t}} \right) = 1 - \text{erfc} \left(\frac{0.898}{\sqrt{t}} \right) = 0.99$$

using tables of error function

$$\frac{0.898}{\sqrt{t}} = 1.822$$

$$\Rightarrow t = 0.243 \text{ s}$$

So, we can write the definition, 1 minus 0.01 is equal to, since we know $\text{erfc } y$ is equal to 1 minus $\text{erf } y$, complimentary error function of y . So, we can write, 1 minus 0.01 is equal to 1 minus error function, complimentary error function of 0.898 divided by root over t . Now, we can write from this, $\text{erf } 0.898$ divided by root over t , is equal to 1 minus $\text{erfc } 0.898$ root over t , is equal to 0.99. So, if we use the table, error function, we can obtain, 0.898 divided by root over t is equal to 1.822. So, t is equal to 0.243 second.

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Example : Solution

The surface velocity

$$v_{y,max} = \frac{\rho g \sin \theta \delta^2}{2\mu}$$
$$= \frac{997 \times 9.81 \times \sin 45 \times (2.54 \times 10^{-4})^2}{2 \times 8.95 \times 10^{-4}}$$
$$= 0.3 \text{ m/s}$$
$$v_{y,max} = \frac{3}{2} v_{y,avg} = \frac{3}{2} \times 0.2 = 0.3 \text{ m/s}$$
$$\text{Maximum length of column} = 0.243 \text{ sec} \times 0.3 \frac{\text{m}}{\text{s}}$$
$$= 0.0723 \text{ m}$$

So, the surface velocity, that is $V_{y \text{ max}}$, is equal to $\rho g \sin \theta \delta^2$ by twice μ ; so, which is equal to 997, into 9.81, into $\sin 45$ into 2.54×10^{-4} to the power minus 4 square, divided by 2 into 8.95×10^{-4} , which is equal to 0.3 meter per second; or, in other words, we can calculate $V_{y \text{ max}}$ is equal to 3 by 2 $V_{y \text{ average}}$, which will be the same value, 3 by 2 into 0.2, which is equal to 0.3 meter per second. So, the maximum length of the column would be equal to time, 0.243 second into 0.3 meter per second. So, it will be 0.0723 meter. So, this way, we can calculate the maximum length for the column to be required.